

Multi-Scale Simulations of Graphene for Energy Applications



Valentina Tozzini

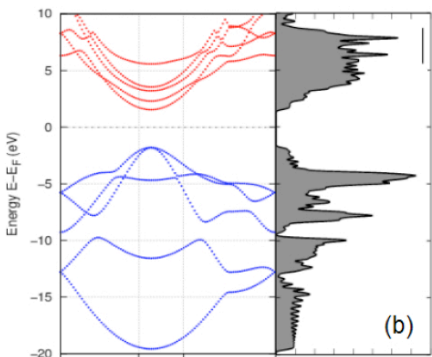
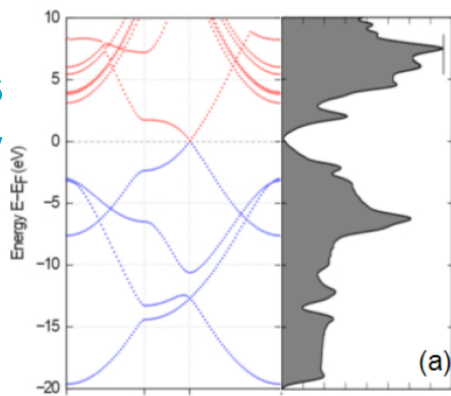
Istituto Nanoscienze del Cnr, NEST-SNS

Layout

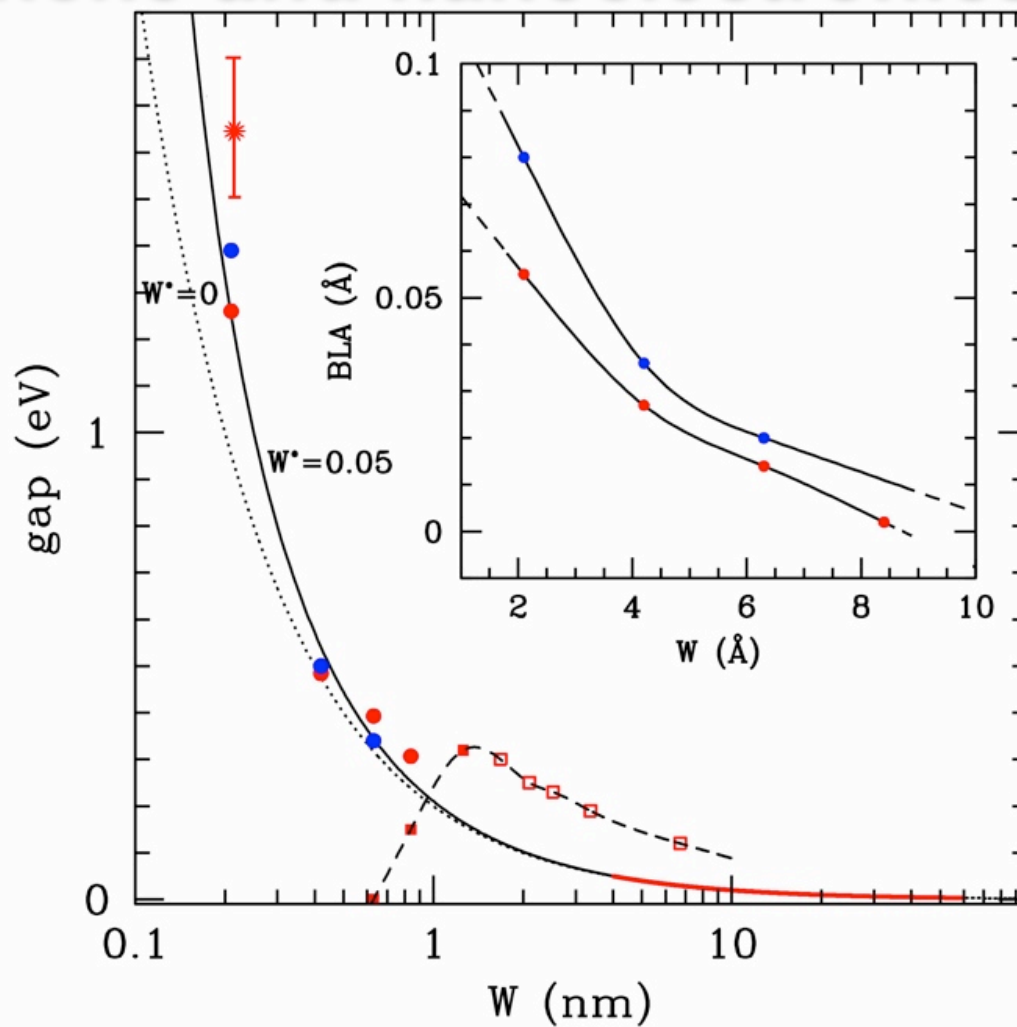
- ❖ Hydrogen and graphene: applications
- ❖ The method: Multi-scale simulations
- ❖ Results and perspectives

Hydrogenated graphene and nanoelectronics

Graphene is a conductor

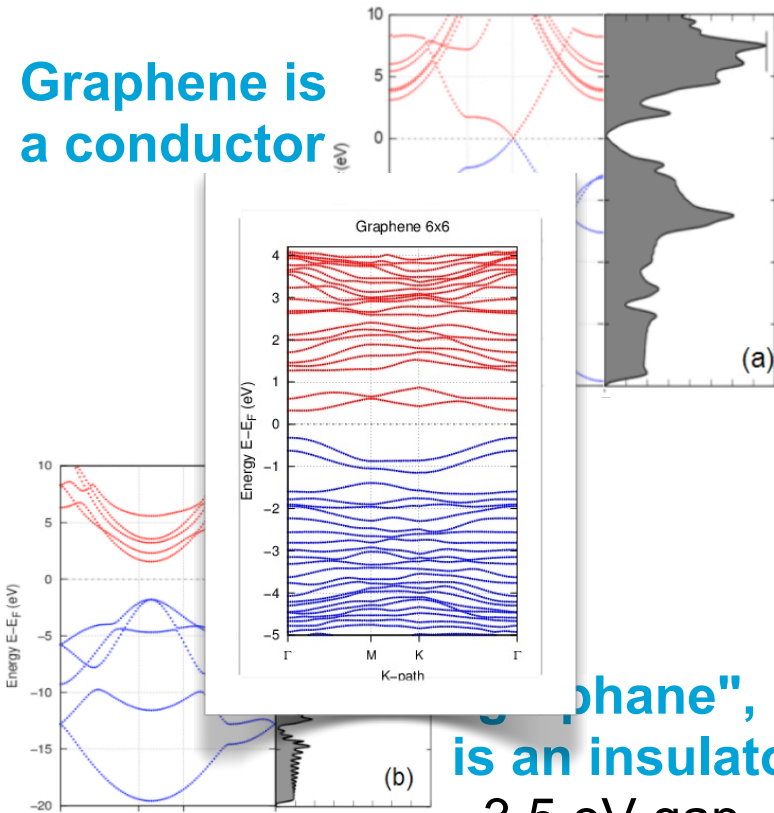


"graphene", is an insulator
~3.5 eV gap



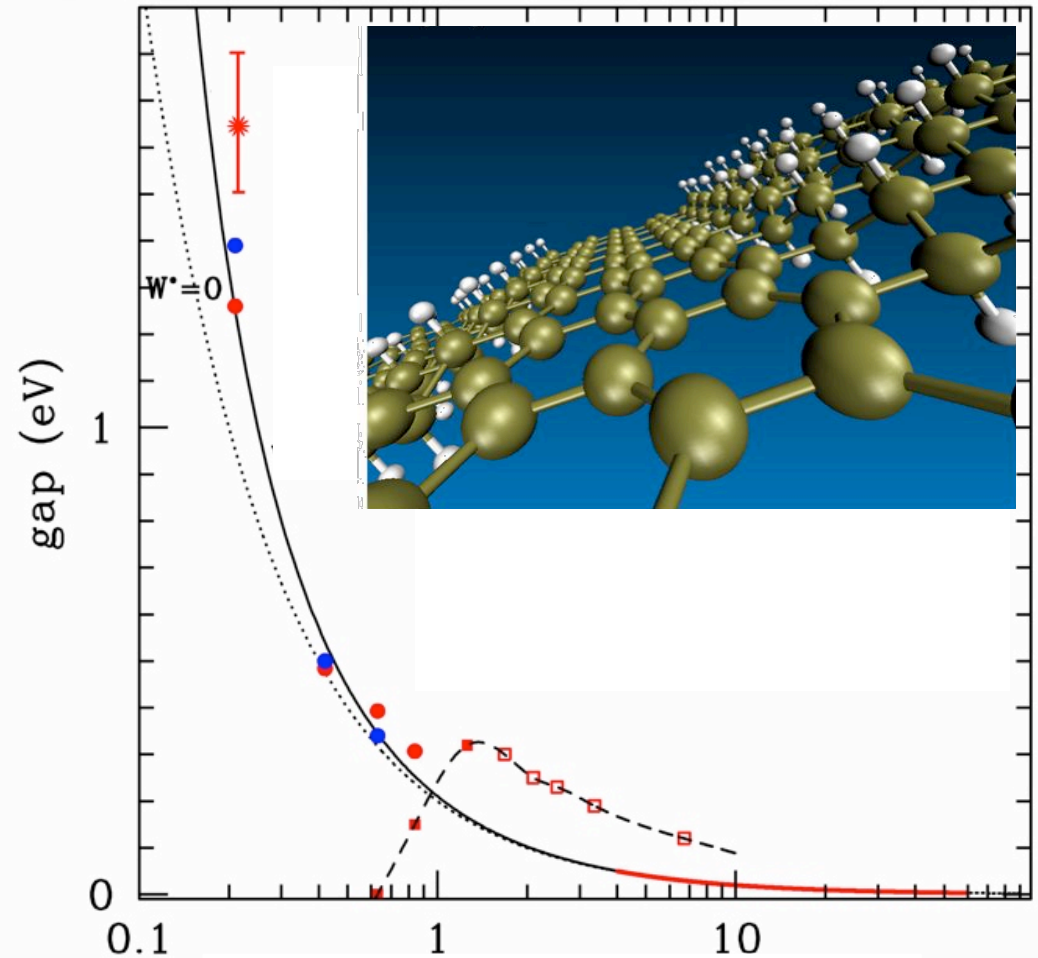
Hydrogenated graphene and nanoelectronics

Graphene is a conductor



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partially hydrogenated graphene is a tunable semiconductor, provided hydrogenation amount and decoration is controllable



Decreasing H content, W (nm)

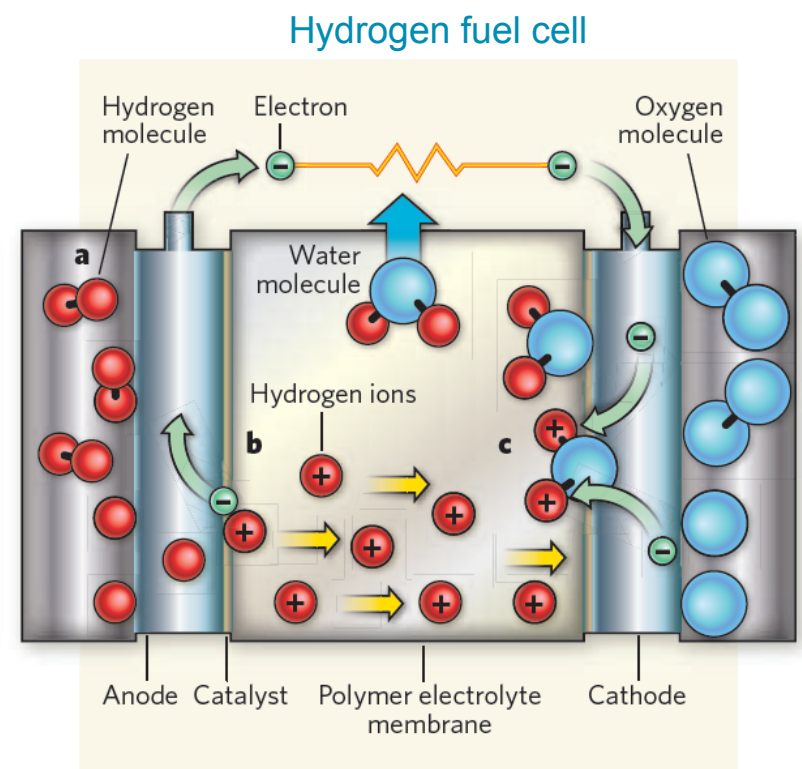
Hydrogen & energy

As a **fuel**, hydrogen has advantages

- ❖ high energy-to-mass ratio



- ❖ Non-toxic, “clean” (product = water) and renewable



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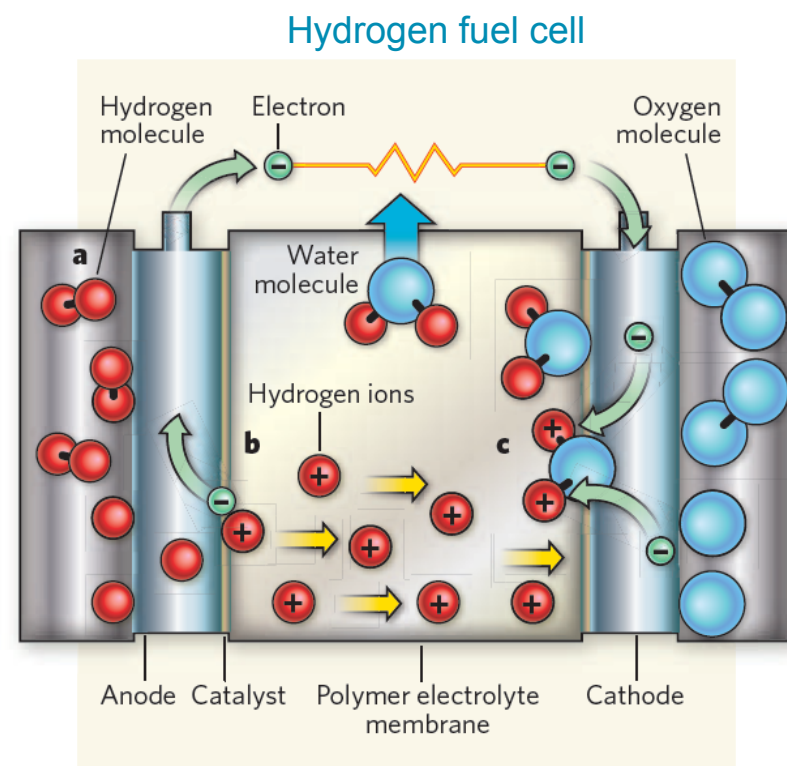


- ❖ Non-toxic, “clean” (product = water) and renewable

However, hydrogen is NOT an **energy source**: it must be produced e.g. by electrolysis, needing +2.96 eV, with zero balance with respect to energy production

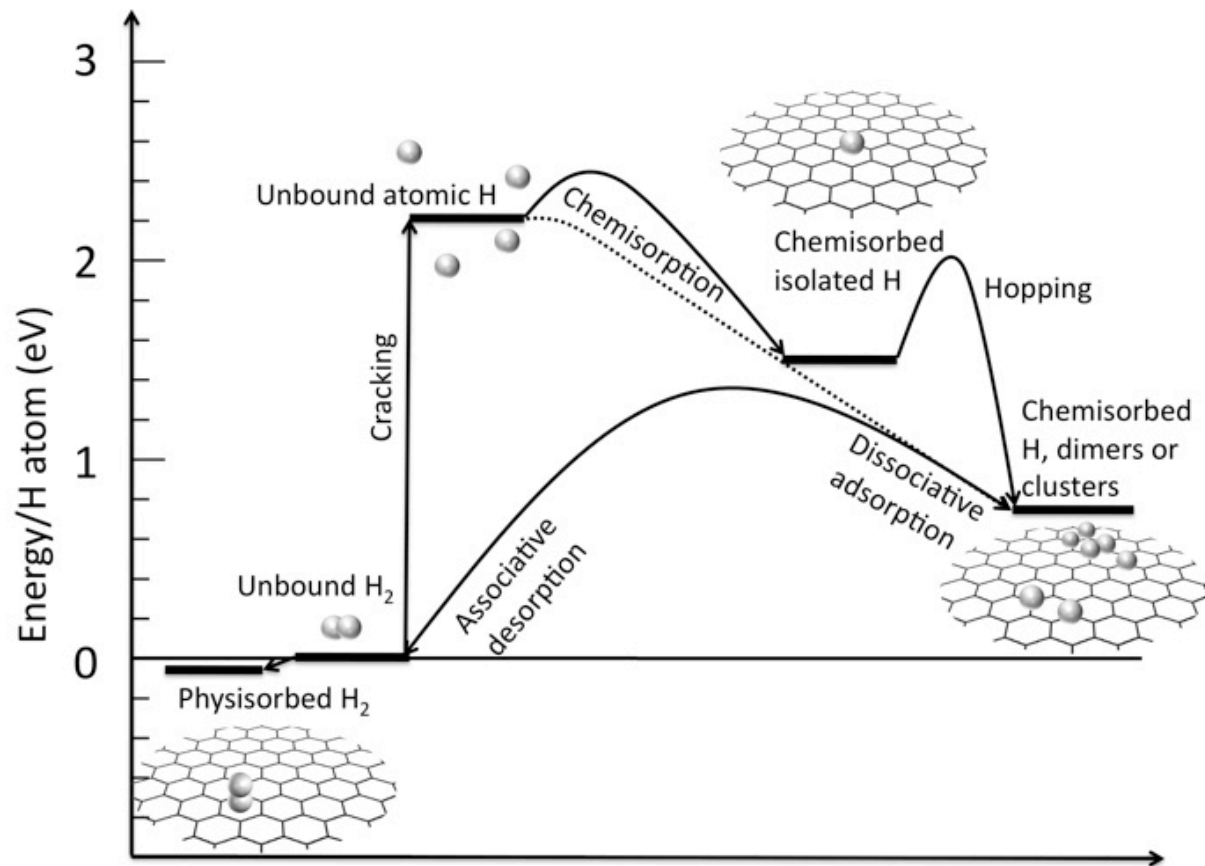
Hydrogen is an **energy carrier** (as electricity) and its advantages must be considered with respect to storage and transportation devices

- ❖ High energy storage capacity ✓
- ❖ Low dispersion (✓)
- ❖ Easy and practical use in standard conditions (✓)
- ❖ Safety (✓)



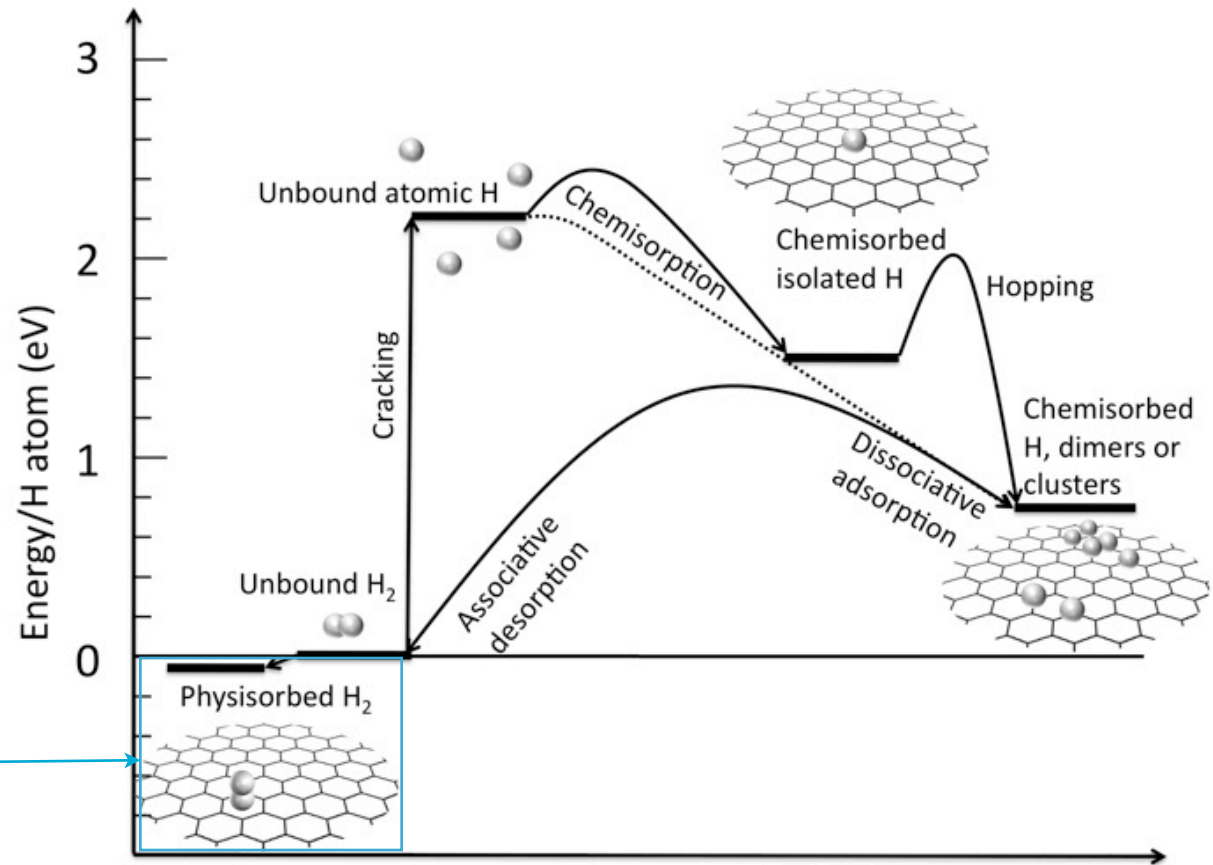
graphene has potentially all of these properties

H storage in graphene



H storage in graphene

❖ Physisorption weakly binds hydrogen \Rightarrow acceptable storage densities only at low temperatures and/or high pressure

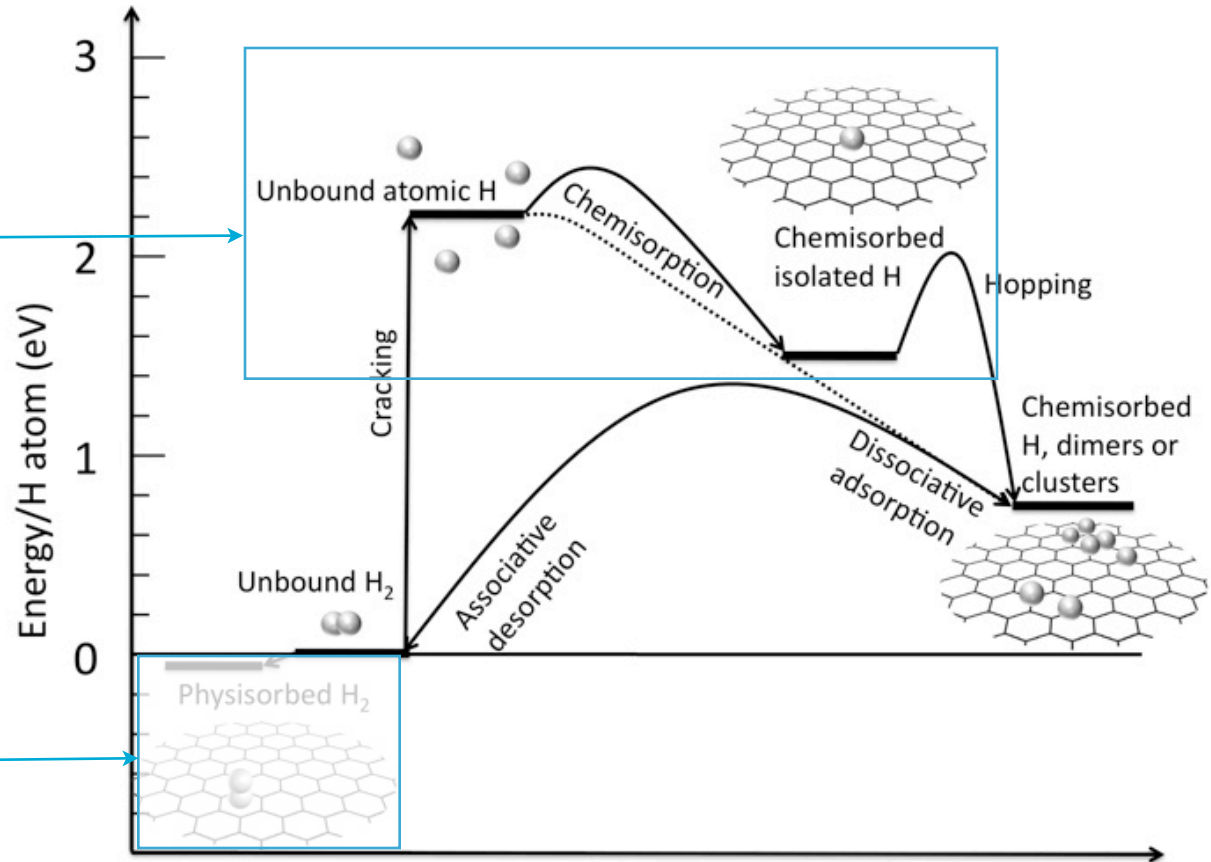


H storage in graphene

❖ Atomic hydrogen chemisorption has a small or negligible chemisorption barrier \Rightarrow

feasible, but H_2 must be cracked

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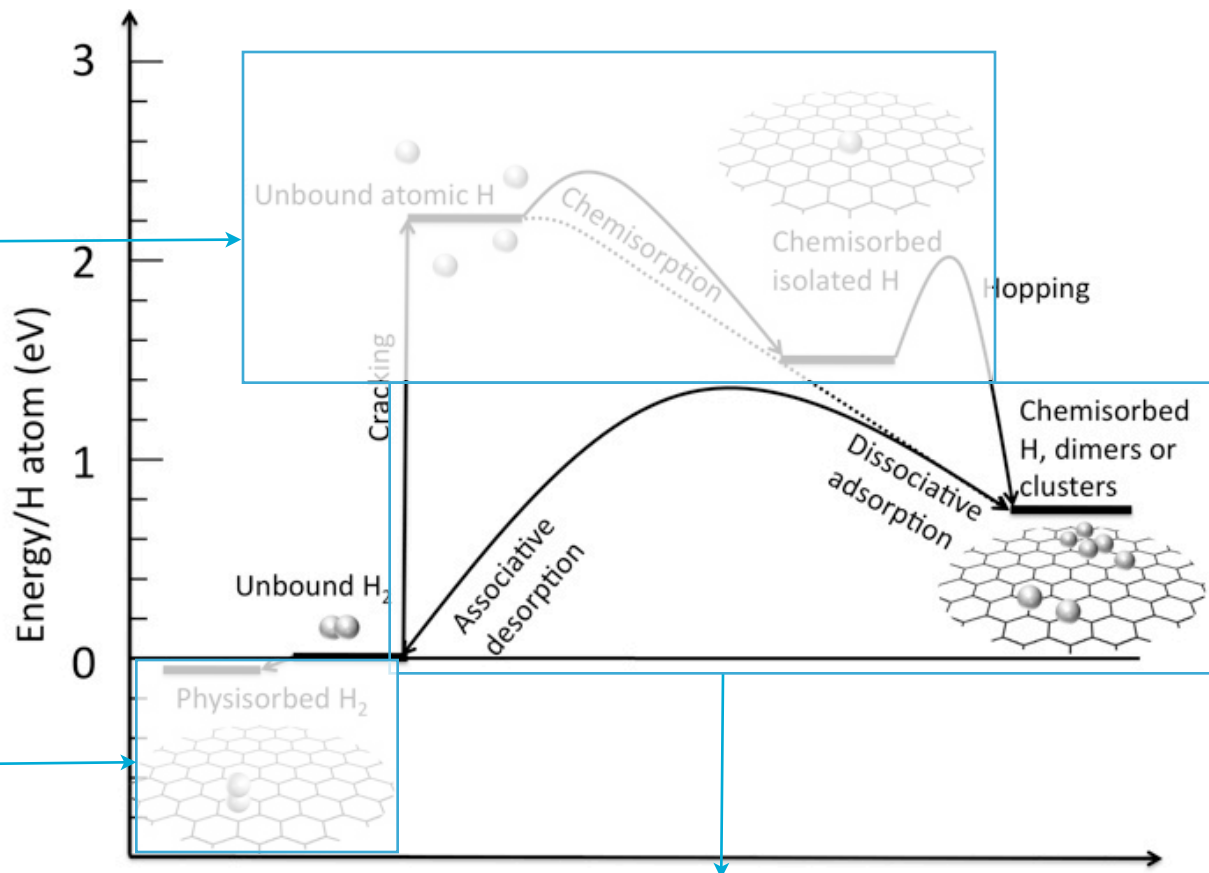


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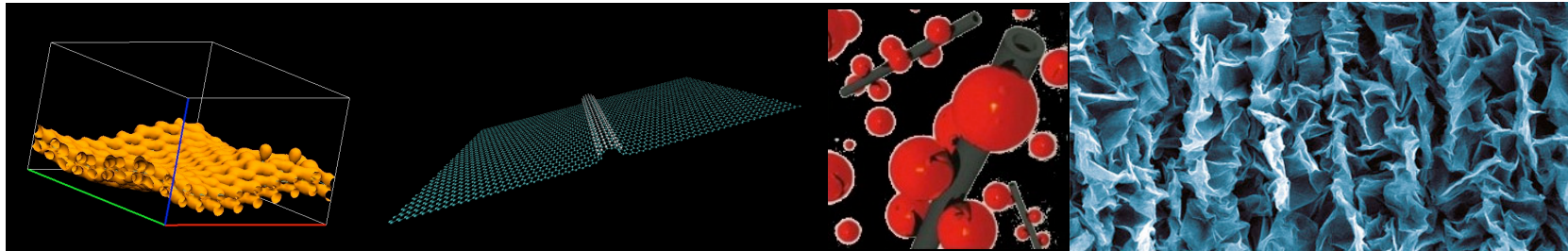
❖ Molecular hydrogen chemi(de)sorption has high barrier (theoretical estimate $\sim eV$) \Rightarrow chemisorbed H is stable for transportation etc, but catalytic mechanisms are necessary in the loading-release phases

\Rightarrow **Chemi(de)sorption is preferable to physisorption** but it is a process with a slow kinetics and difficult to control

(Partial) summary:

- ❖ **Controlling chemi(de)sorption** (or physisorption) of hydrogen is crucial for energetics and nanoelectronics applications, but it requires deep control on the material and innovative catalytic strategies
- ❖ **Theoretical and simulations studies** can help designing these strategies, but the processes involve different length and time scales

The multi-scale simulation approach



QM-DFT
calculations/simulations

electronic props
fine structure
hydrogenation chemistry

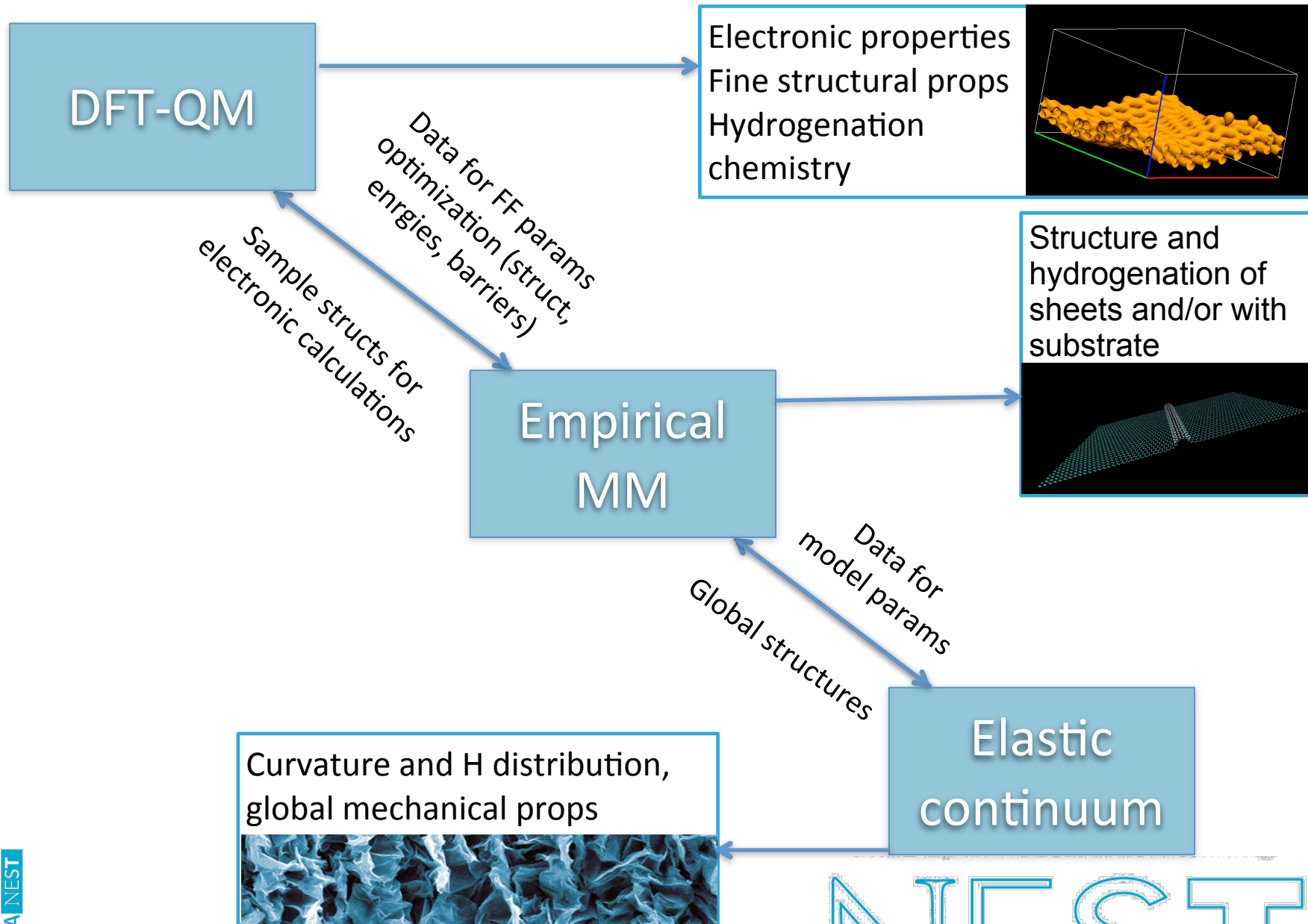
Classical MD
modeling/simulations

structure and
dynamics on the
nano-micro scale

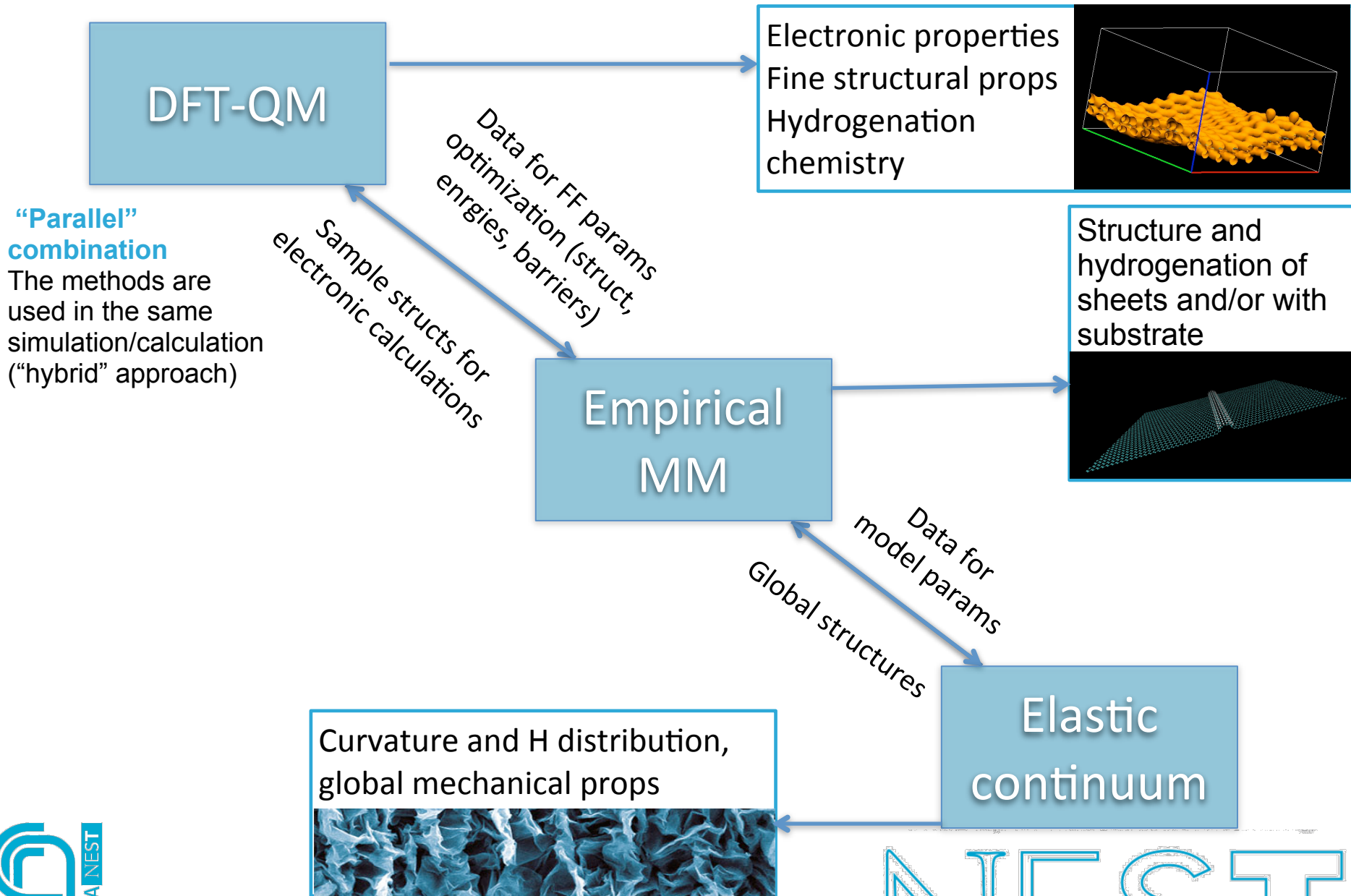
Continuum mechanistic
approaches

statistical behavior
thermodynamics
large scale mechanical/
chemical props

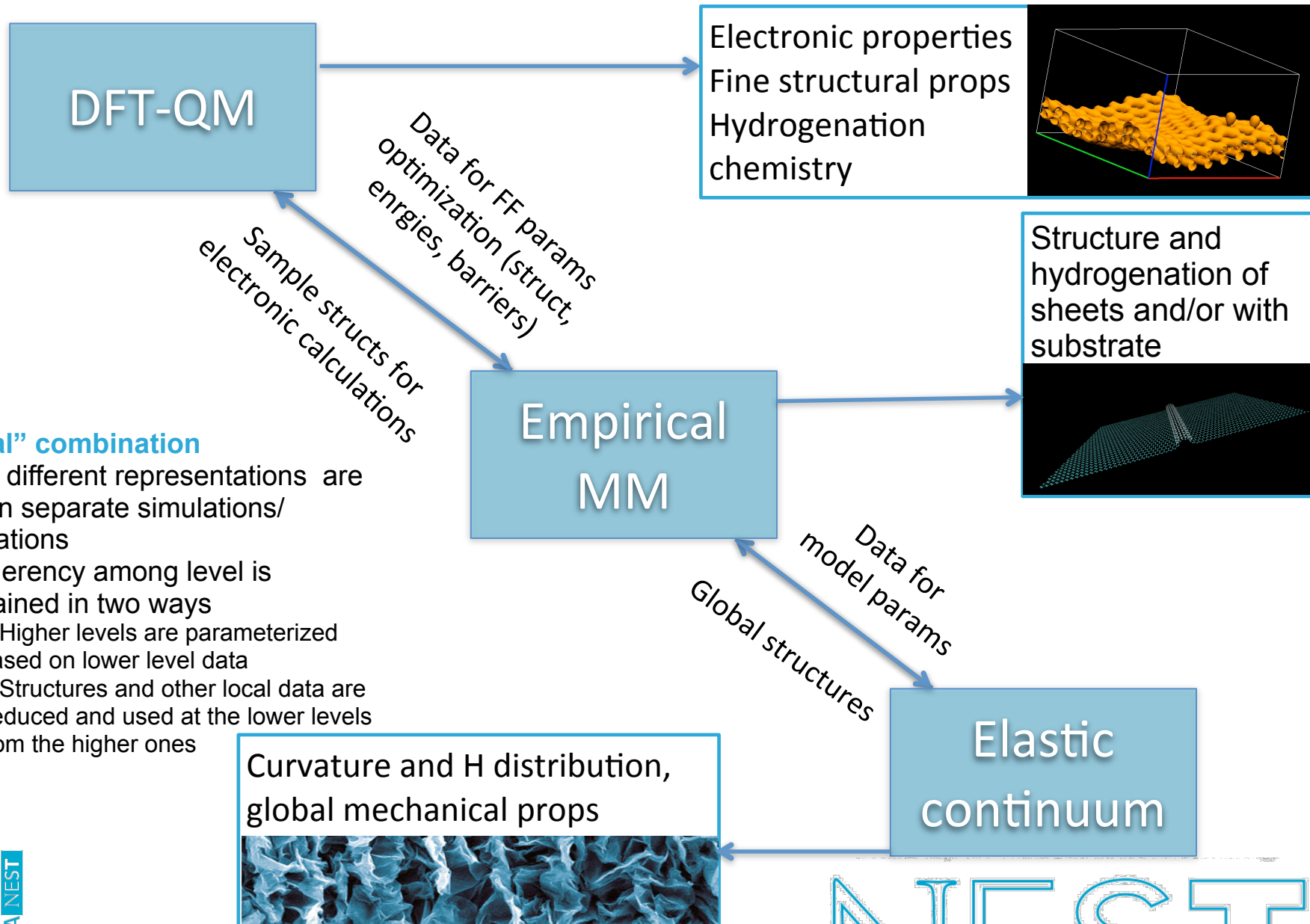
The multi-scale simulation approach



The multi-scale simulation approach



The multi-scale simulation approach



“Serial” combination

- ❖ The different representations are used in separate simulations/ calculations

- ❖ Coherency among level is maintained in two ways

- Higher levels are parameterized based on lower level data
- Structures and other local data are deduced and used at the lower levels from the higher ones

QM-DFT: study of reactivity vs rippling

Scheme 1 DFT-CPMD (CPMD 3.13)

❖ Systems: ~Squared cells

- Large cell: 180 C atoms, 22.13x21.30x15 Å (and contracted) Γ point
- Small cell: 24 C atoms, 7.28x8.41x14.6 Å, 40 K points MP

❖ DFT

- TM pseudo-potentials + VdW (Grimme, 2006)
- plane waves (60 Ryd cutoff), Davidson diag
- PBE functional (checks with BLYP)

❖ System Relaxation and dynamics

- CP dyn, electron mass preconditioning, timestep = ~ 0.1 fs
- Simulated annealing + local optimization
- Nosé Thermostat, restrained/damped MD ([metadyn in progress](#))

Scheme 2 DFT-QE (QE 5.0.2)

❖ Systems: orthorhombic cells

- 6x6(30deg): 54 C atoms, 12.78 Å (and contracted/expanded), K points 7x7x1
- 13x13: 338 C atoms, 31.97 Å, Γ point, checks with K points
- Same on SiC substrate, 1300-1500 atoms and with electric field ([in progress](#))

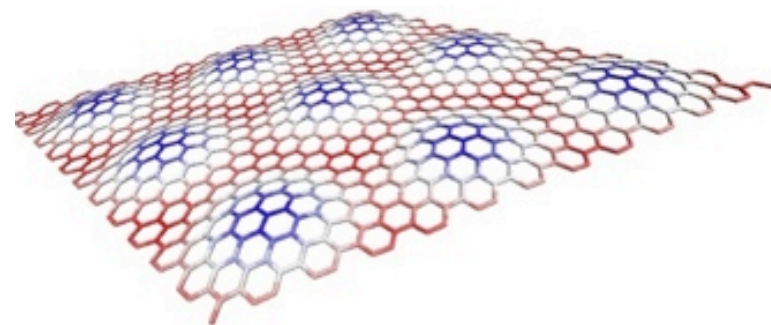
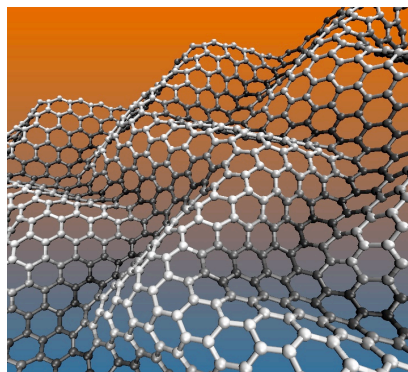
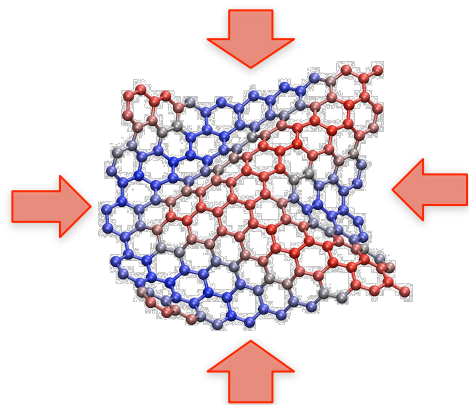
❖ DFT

- Ultrasoft PPs(RRKJ)+VdW (Grimme, 2006)
- plane waves (25 ryd), Davidson diag
- PBE functional (checks with LDA)

❖ System Relaxation and dynamics

- CP and BO timestep = 0.1-0.5fs
- Simulated annealing + local optimization
- Nosé Thermostat, restrained/damped MD

Different patterns and local curvature levels are generated by different boundary or compression conditions

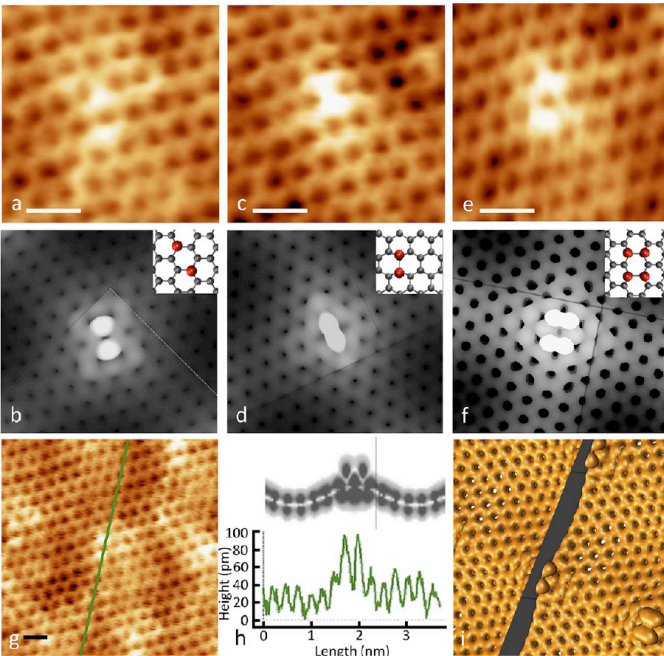
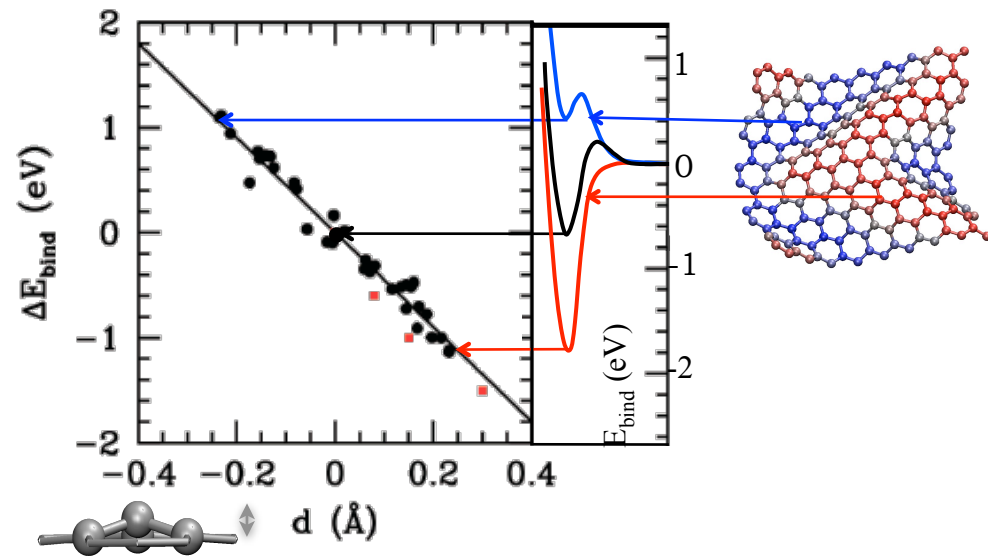


QM-DFT: study of reactivity vs rippling

The relationship between curvature and H binding energy is linear

The reactivity towards H is enhanced on convexities and decreased in concavities \Rightarrow

the curvature can be used to control H adsorption

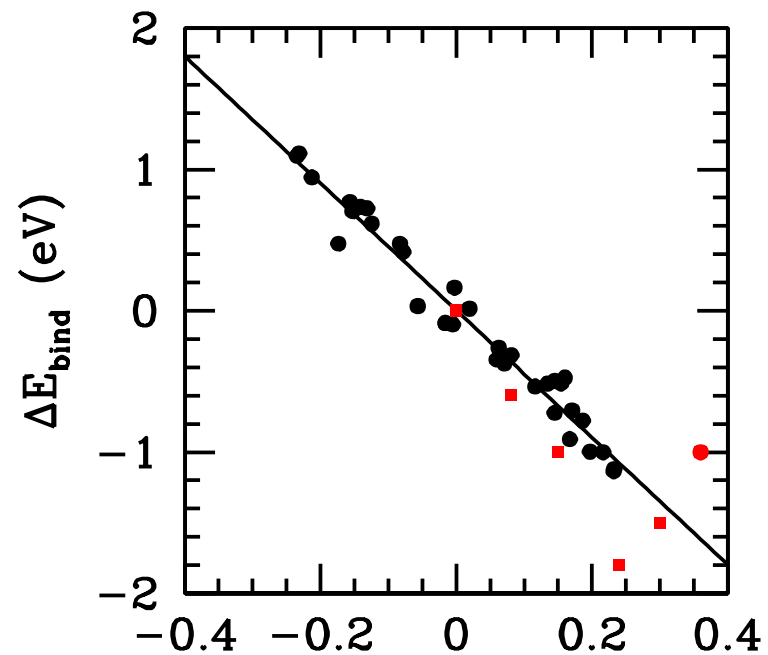


This was experimentally verified by hydrogenation of naturally curved graphene grown on SiC

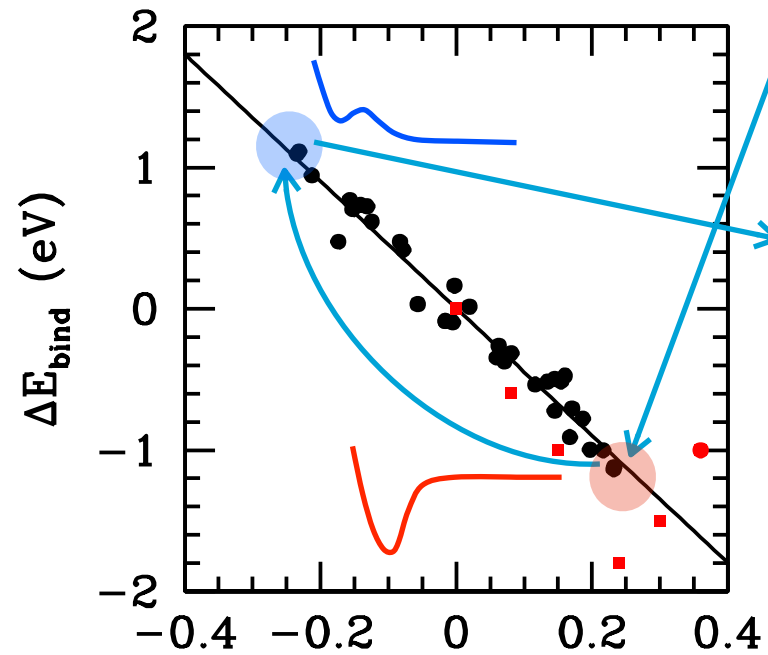
S. Goler et al. [The influence of graphene curvature on H adsorption](#) JPCC (2013)

V Tozzini, V Pellegrini [Reversible H storage by controlled buckling of graphene Layers](#) JPCC (2011)

QM-DFT: study of reactivity vs rippling



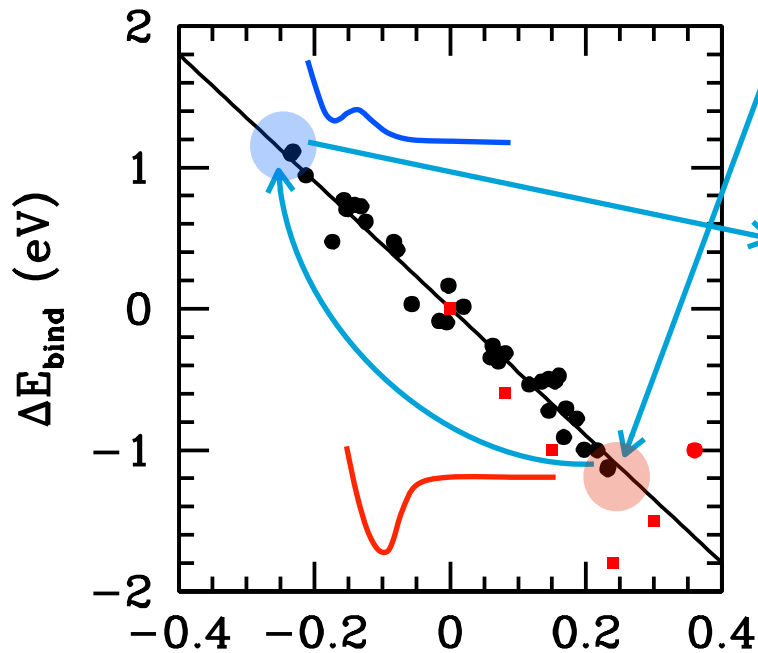
QM-DFT: study of reactivity vs rippling



1. H is adsorbed on convex sites
2. Inverting the curvature, H is found on concave sites: unstable adsorbate...
3. ... and H dissociation

⇒ Curvature inversion could be used as H release mechanism

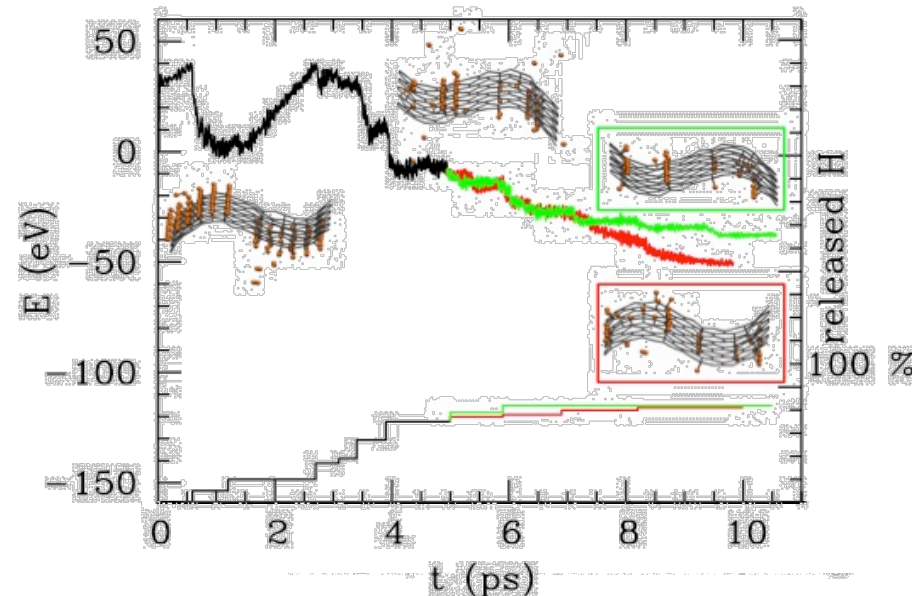
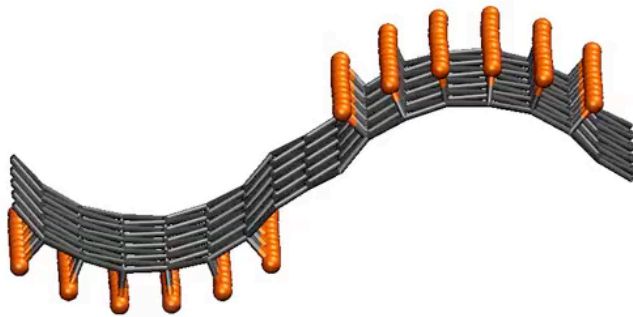
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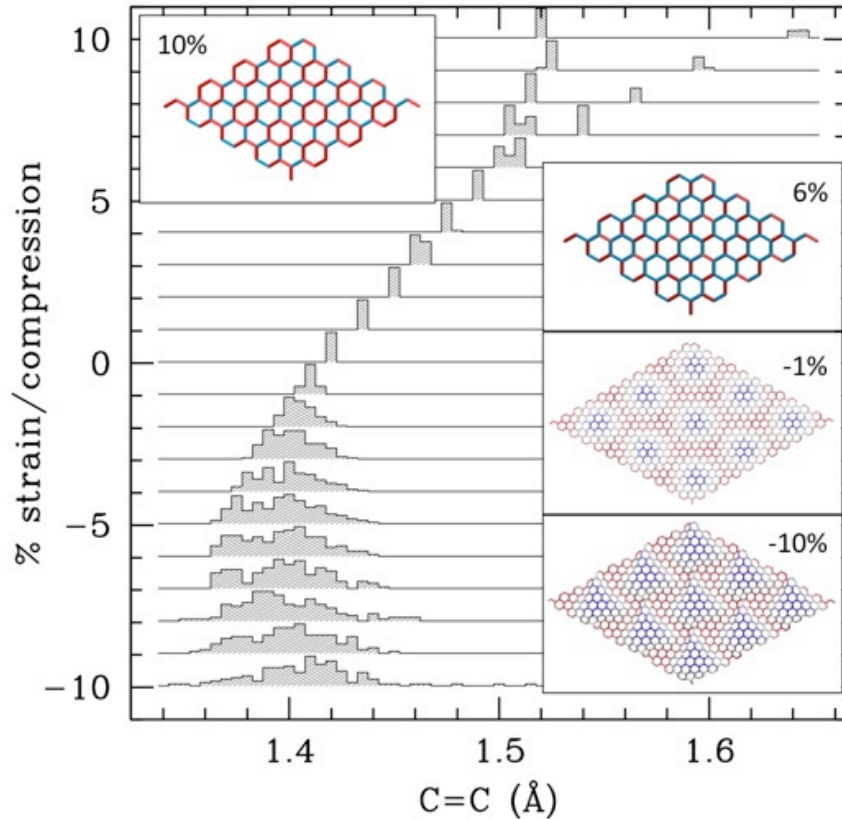
Curvature inversion by mechanical wave



⇒ Application to H storage

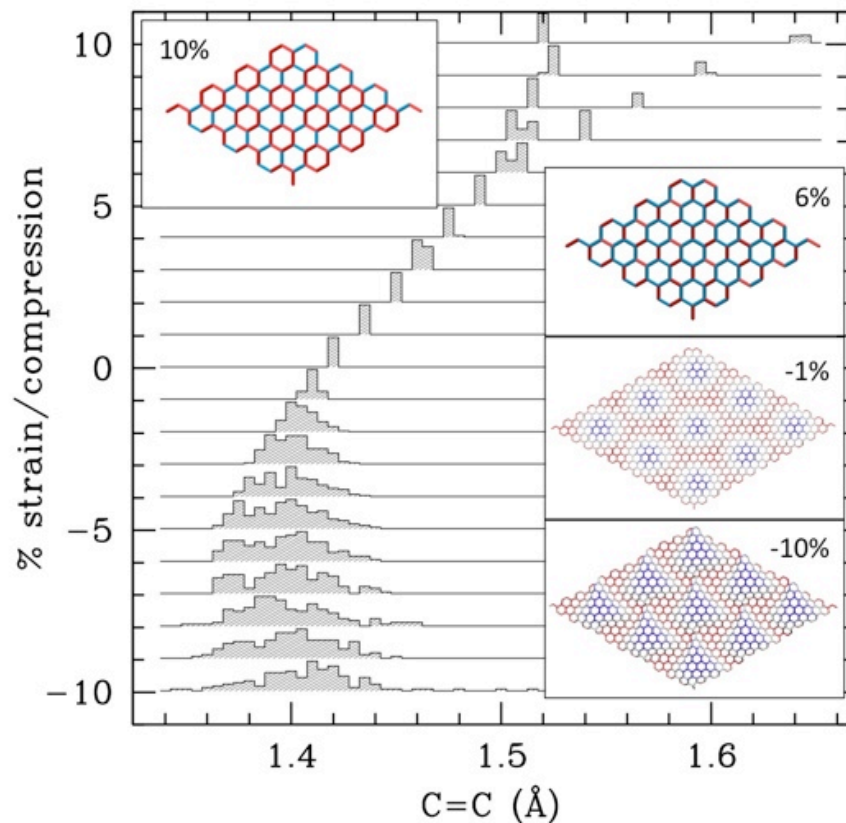
Toulouse, May 8th 2014

QM-DFT: study of reactivity vs rippling



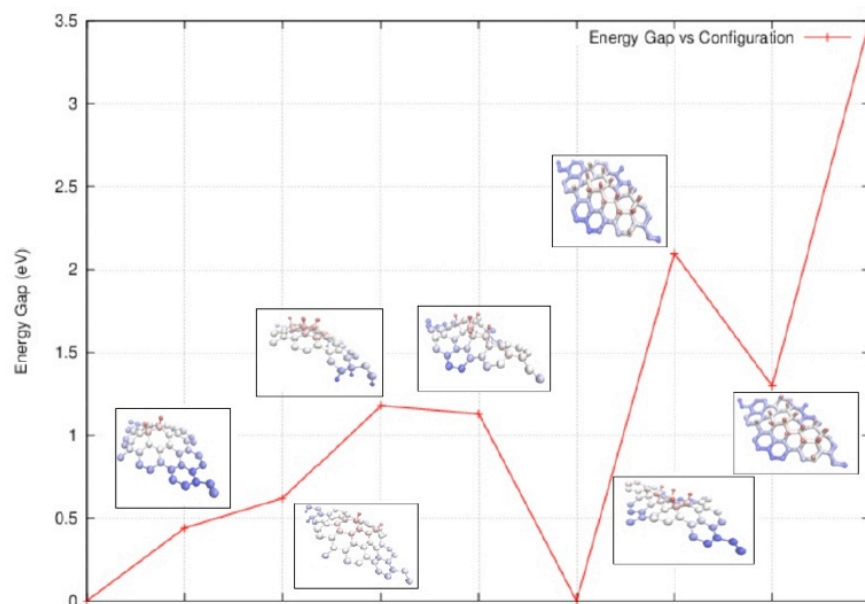
- Strain generate a bond distortion with a formation of isolated “benzenes” separated by single bonds
- Contraction induces pyramidalization and enhances reactivity

QM-DFT: study of reactivity vs rippling



- When hydrogen is added on convexities, different pattern and level of hydrogenation generate systems with different band structure and gap

- Strain generate a bond distortion with a formation of isolated “benzenes” separated by single bonds
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A Rossi, V tozzini [Structure, electronic properties and stability of nano-scopically corrugated/hydrogenated graphene: a Density Functional Theory study in preparation](#)

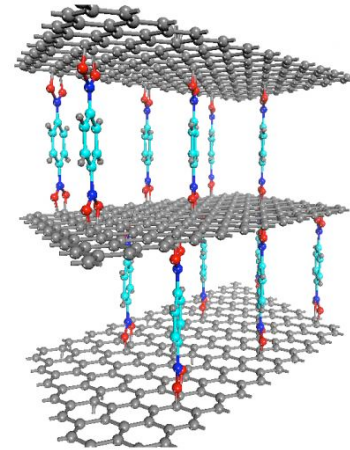
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Towards real devices

- ❖ Create and maintain an extended multilayer structure

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“Spacers” molecules are currently under consideration to create “pillared” multilayer structure



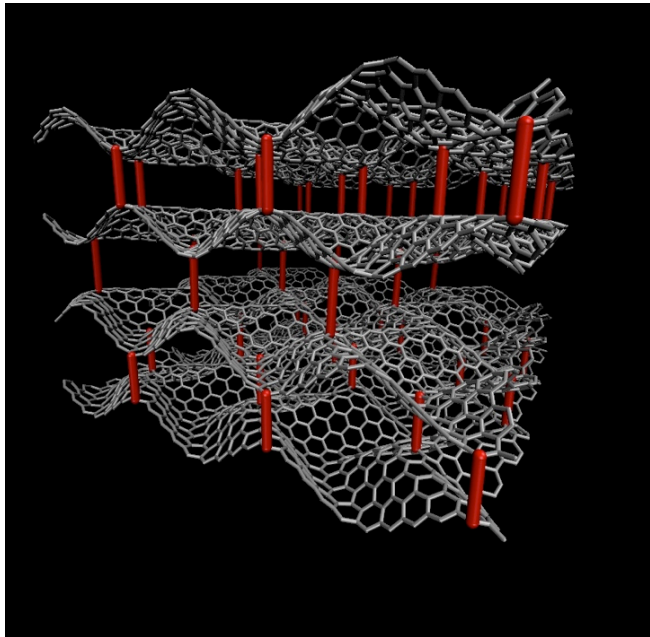
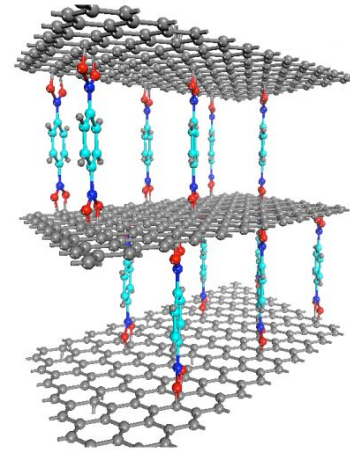
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❖ Control the curvature

if designed sensitive to external stimuli (e.g. light or E/M fields) pillars could also serve to create and control the curvature



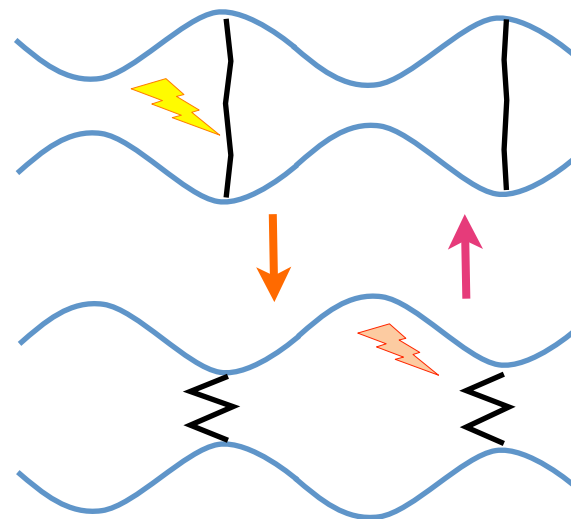
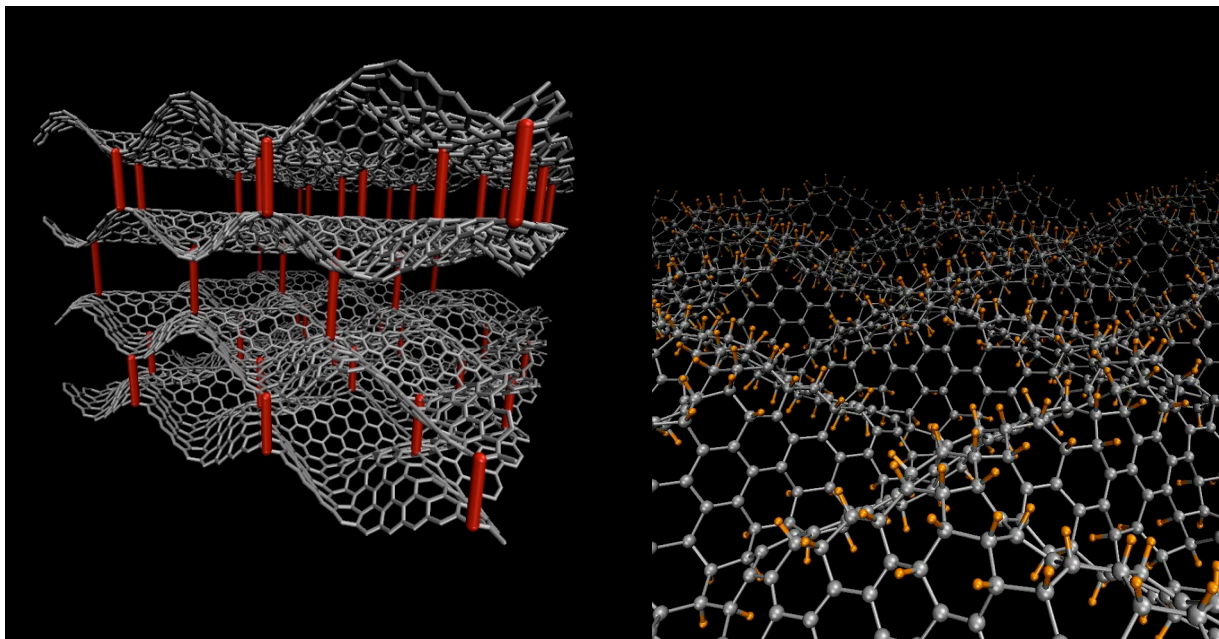
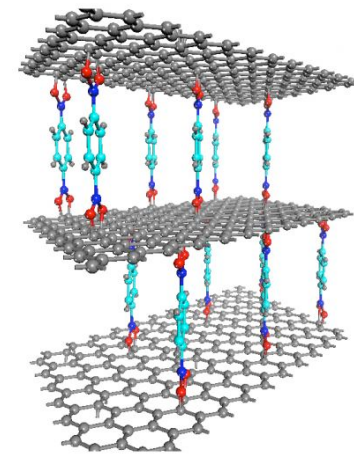
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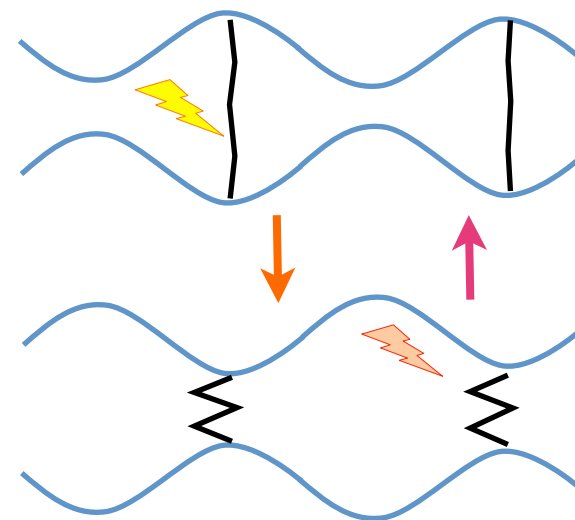
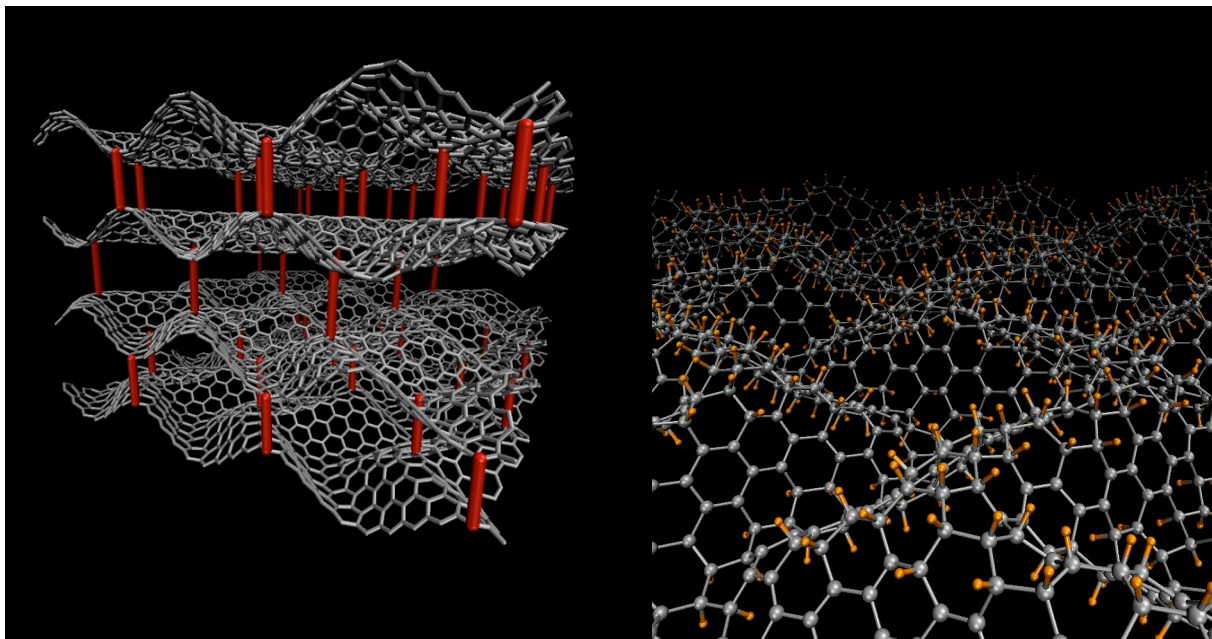
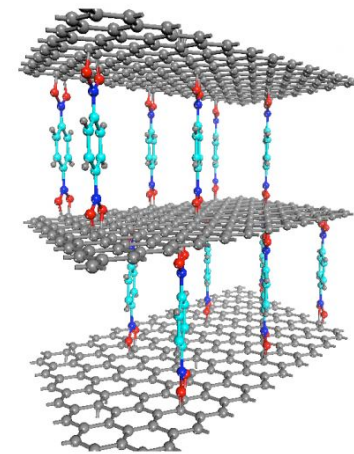
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- ❖ Control the curvature
 - if designed sensitive to external stimuli (e.g. light or E/M fields) pillars could also serve to create and control the curvature
- ❖ Evaluate the effects on the macroscopic scales



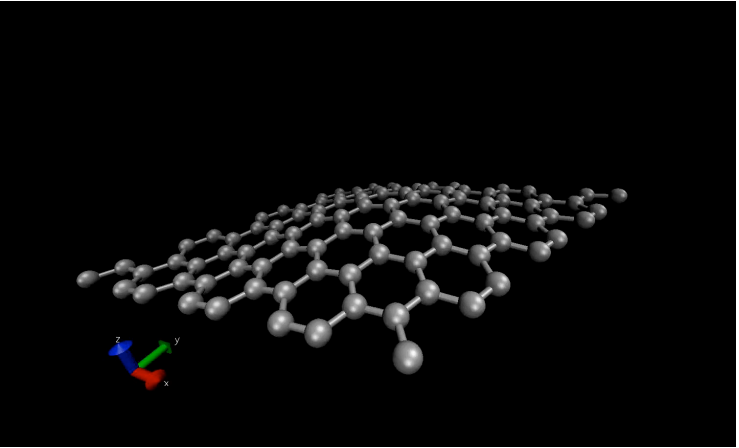
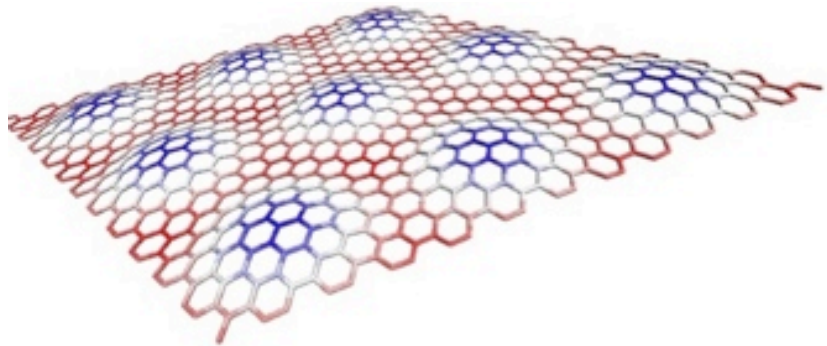
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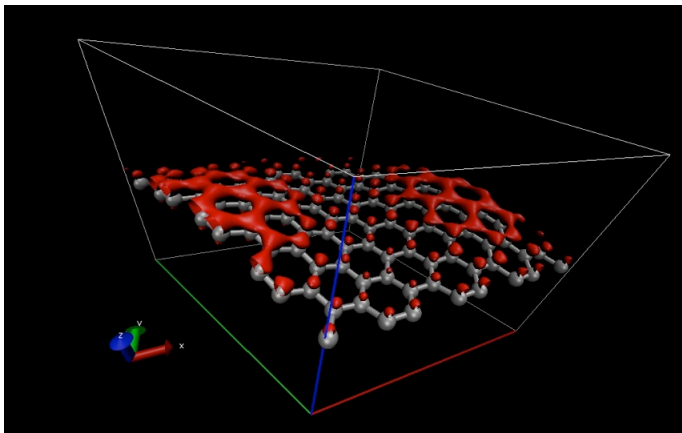
Towards real devices

❖ Control the curvature

Control by external **electric field system**: corrugated graphene sheet (like graphen on SiC)



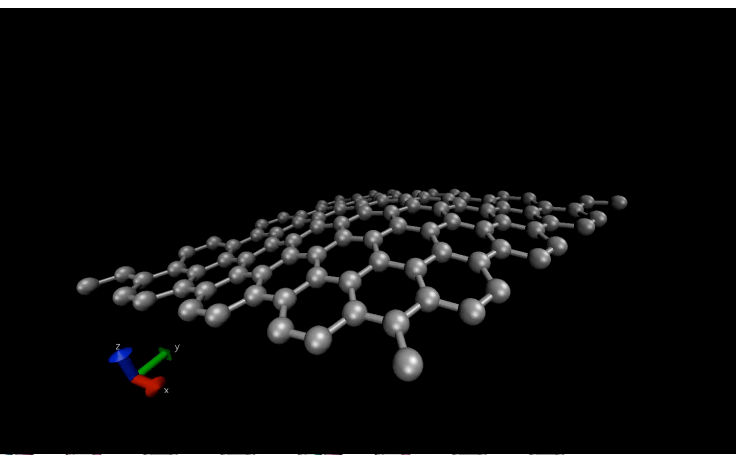
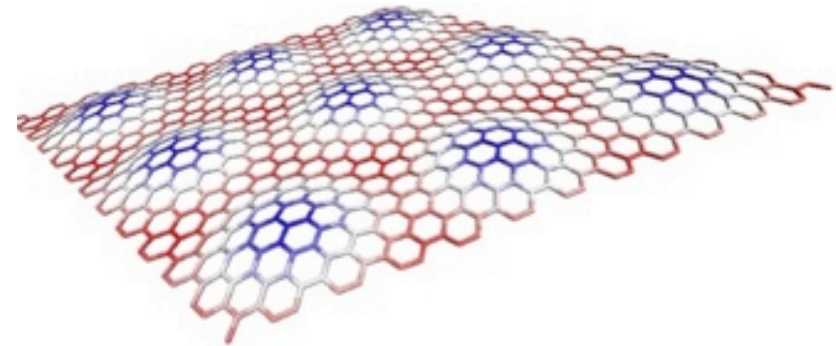
The local curvature is enhanced by an electric field orthogonal to the sheet, due to electronic charge rearrangement



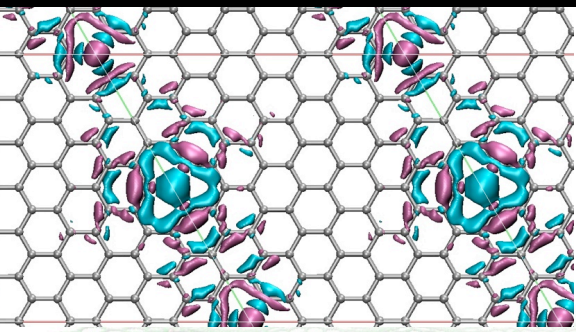
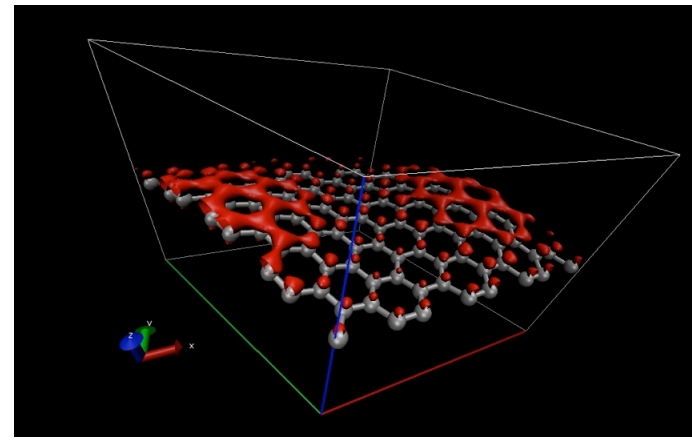
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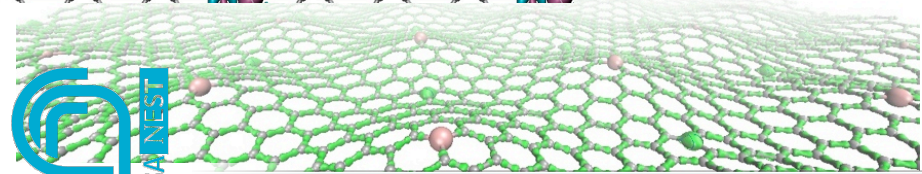
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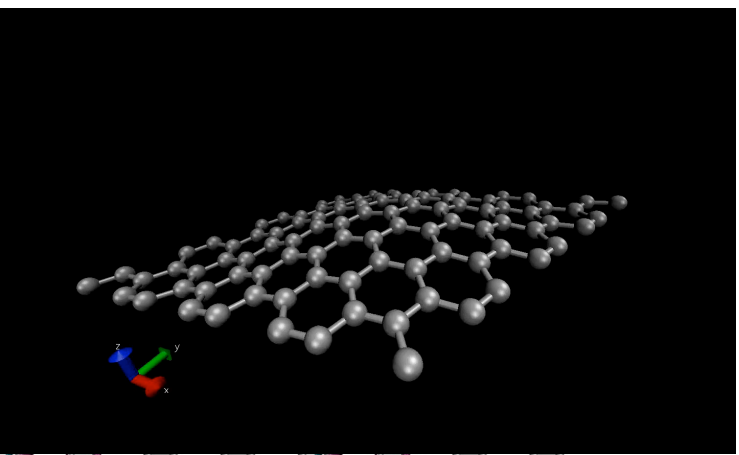
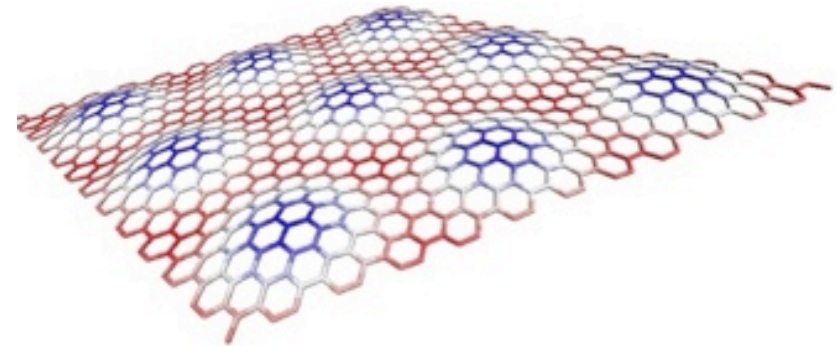
The effect is enhanced by B and N substitutions, creating local charges



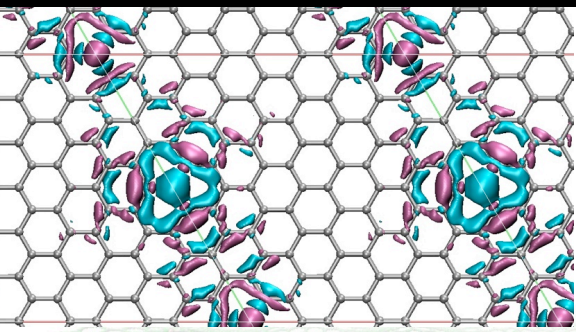
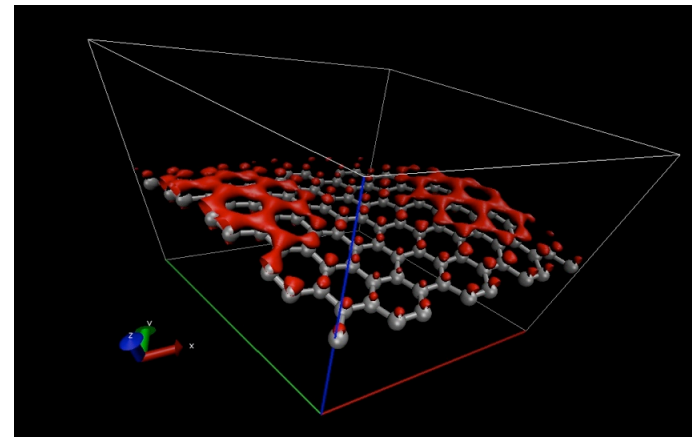
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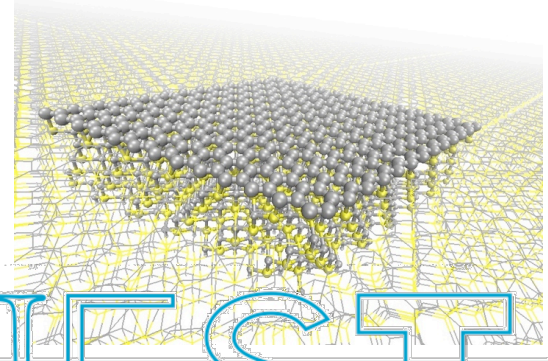


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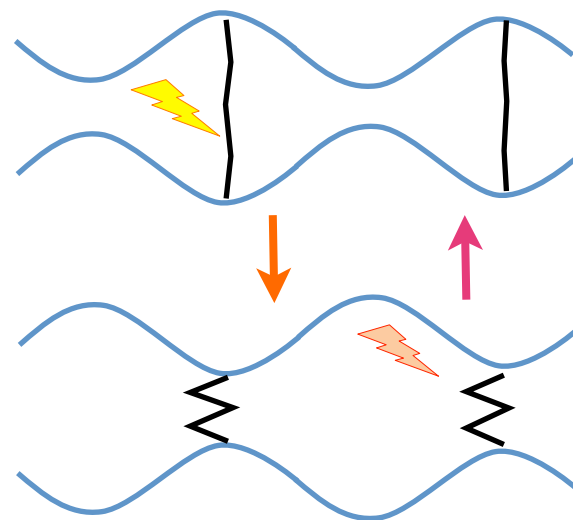
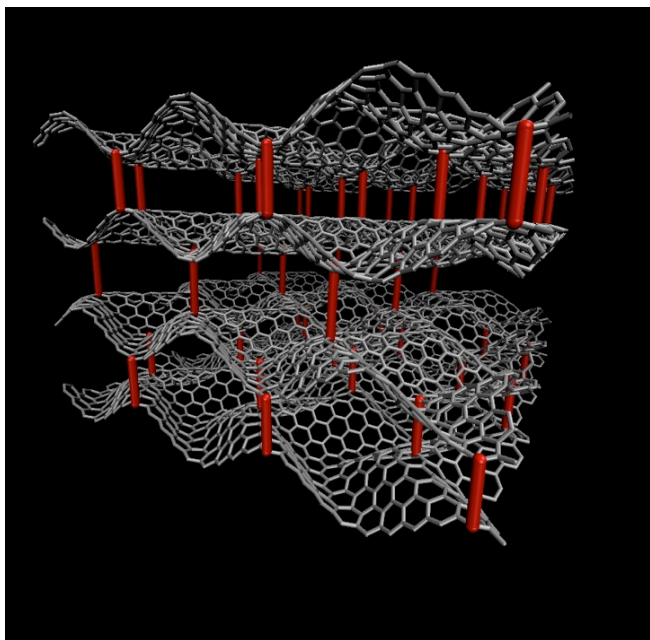
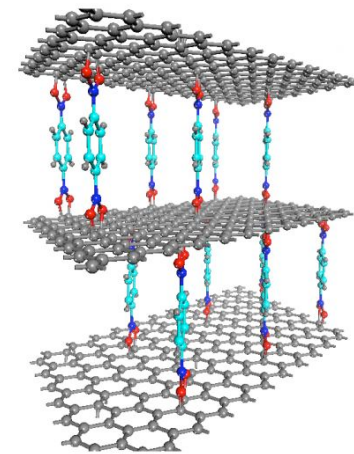
Next: Substrate adding, for direct comparison with experiment



T Cavallucci, V Tozzini *in progress*

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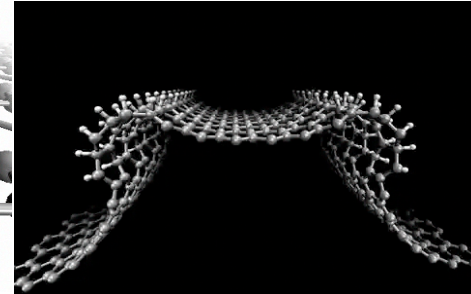
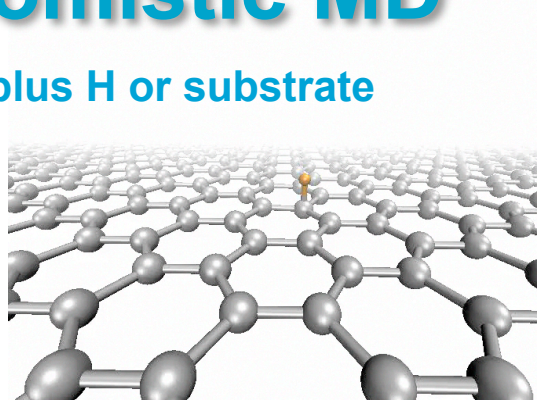
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Classical classical atomistic MD

❖ **System: 10000-100000 atoms (30-50 nm) plus H or substrate**

- Any periodic boundary conditions
 - Software: DL_POLY, LAMMPS
- ❖ System Relaxation and dynamics
- Classical dynamics, timestep $\sim 1-2$ fs
 - Simulated annealing + local optimization
 - NVE, NVT, NPT

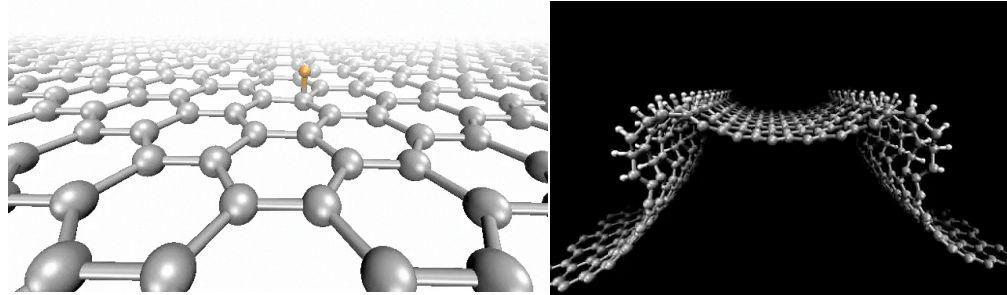


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❖ **Force Fields**

❖ **Scheme 1: “Tersoff-like” potentials**

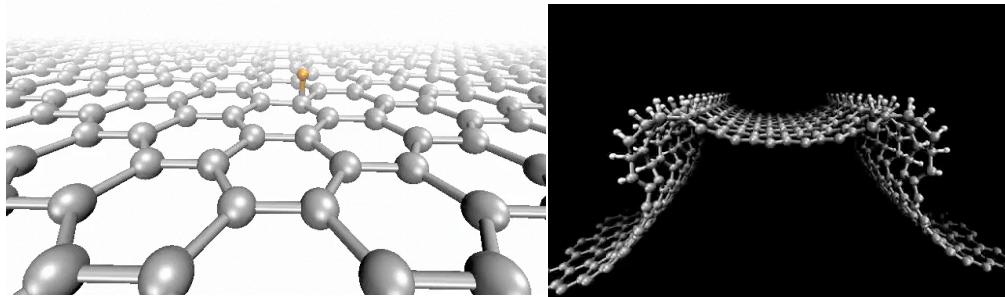
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BUT...
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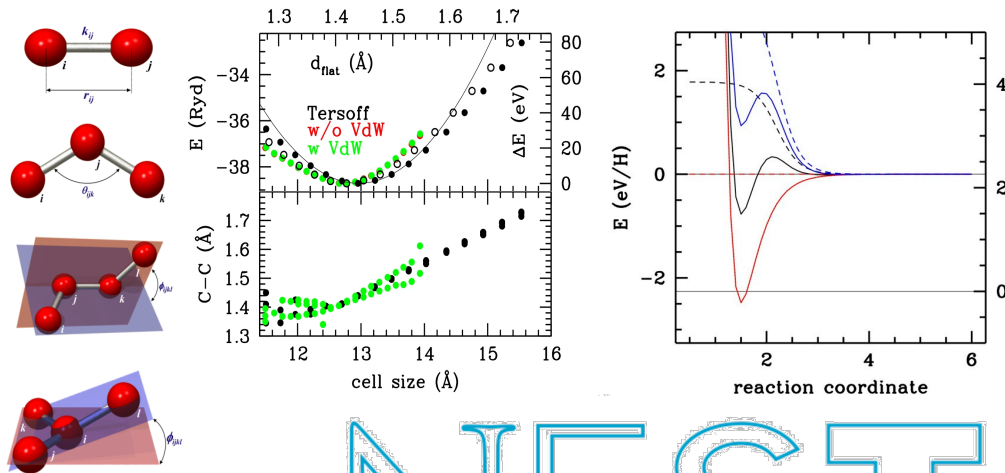
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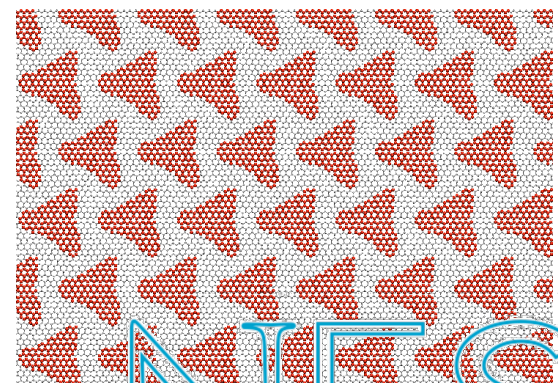
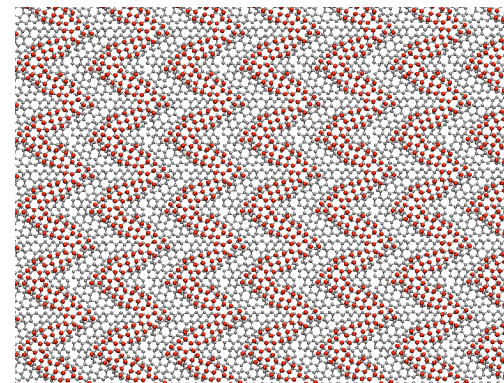
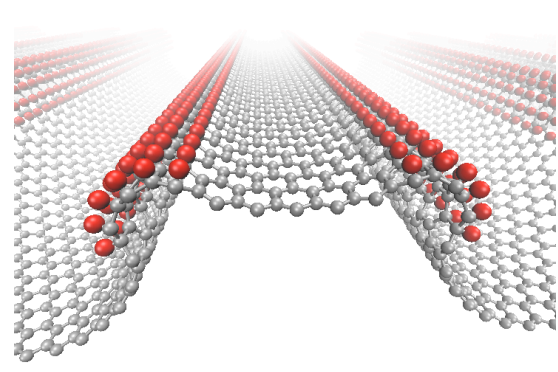
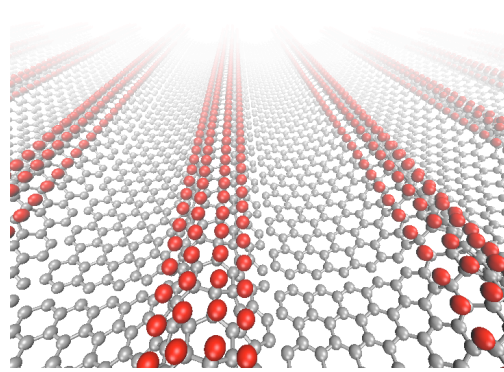
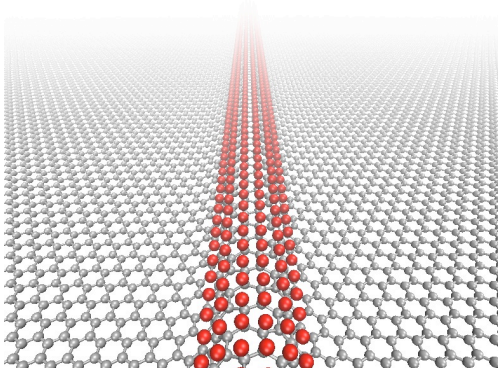
❖ Scheme 2: “connective” FF

including

- ⊙ energetics of compression/strain
- ⊙ sp^2 - sp^3 transition
- ⊙ Interplay between curvature and hydrogenation
- ⊙ interplay between BN doping, curvature and electric fields

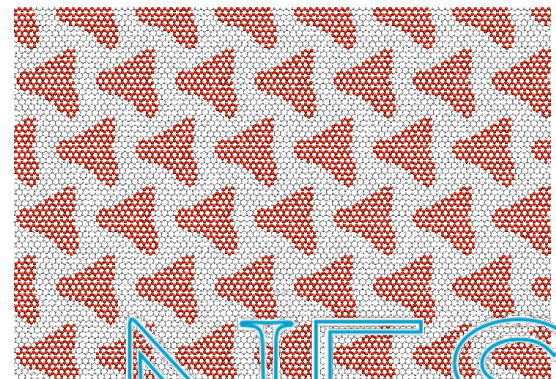
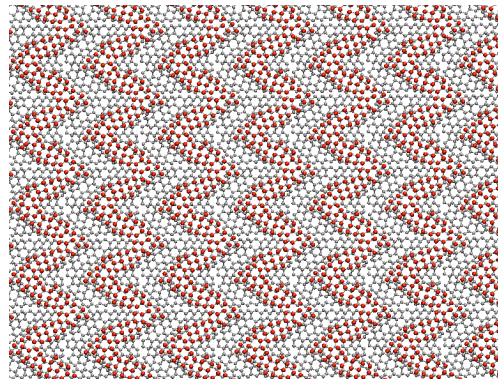
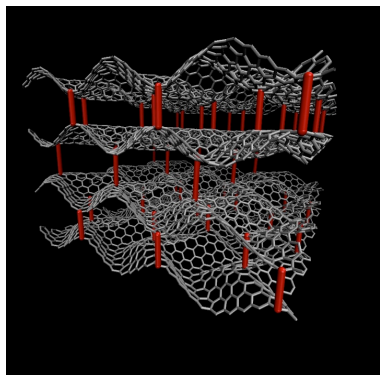
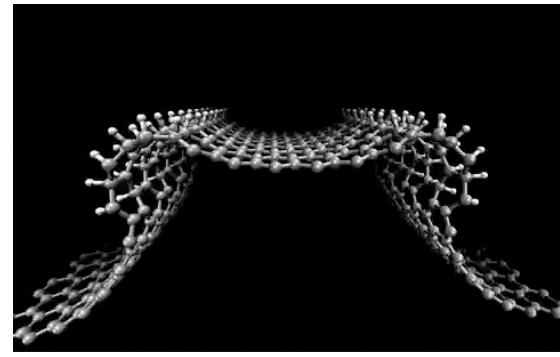
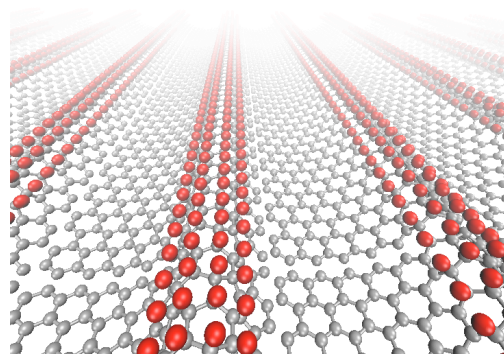
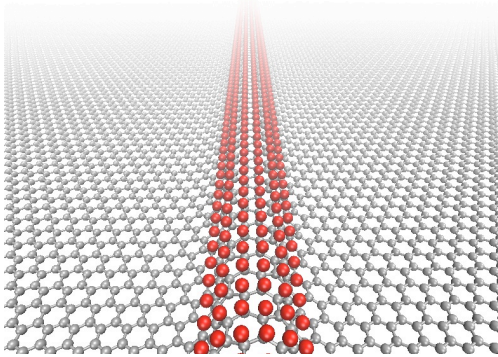


Towards real devices



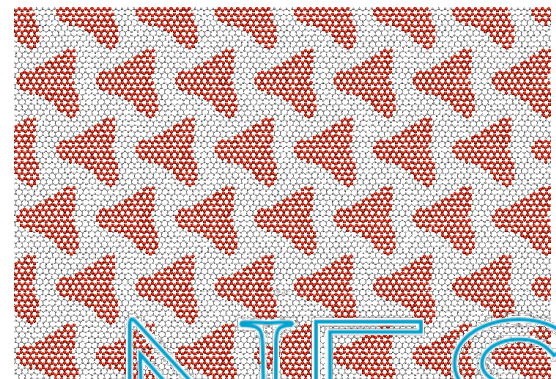
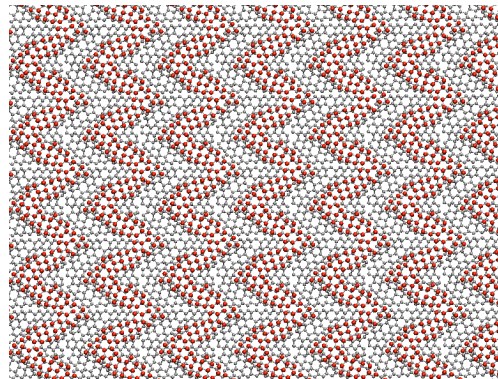
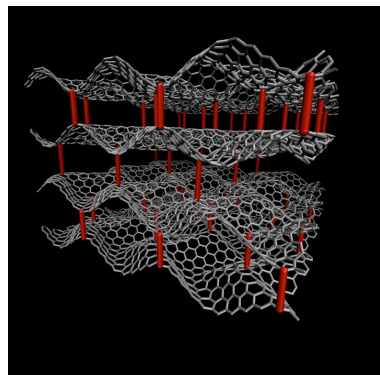
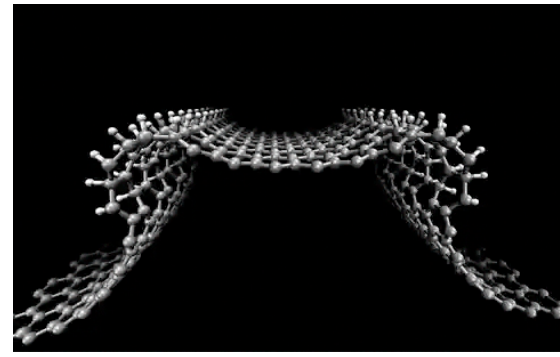
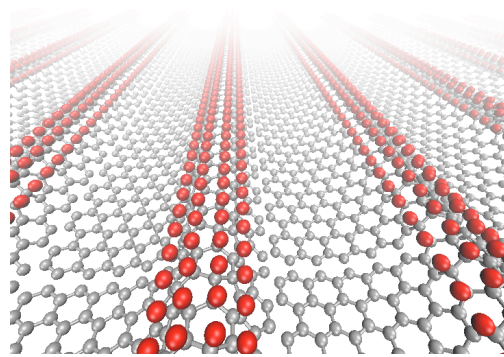
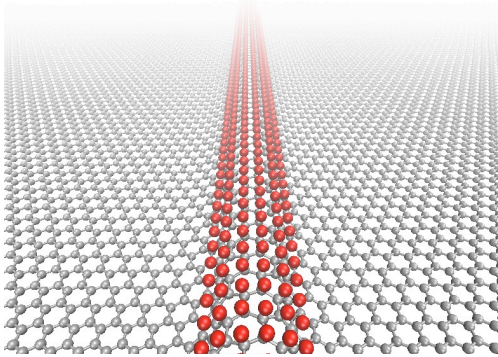
Towards real devices

- ❖ Create and maintain an extended (multilayer) structure
“Spacers” molecules are currently under consideration to create “pillared” multilayer structure



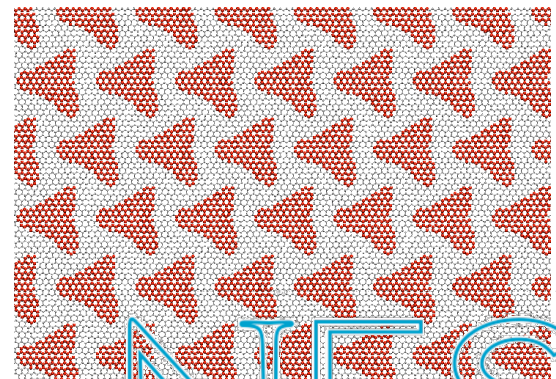
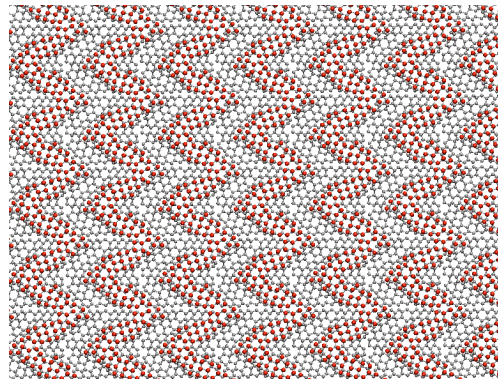
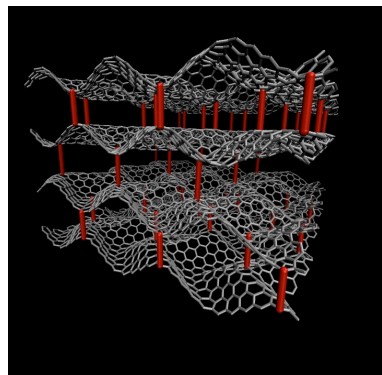
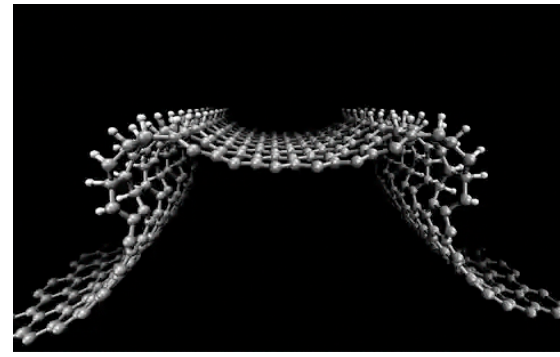
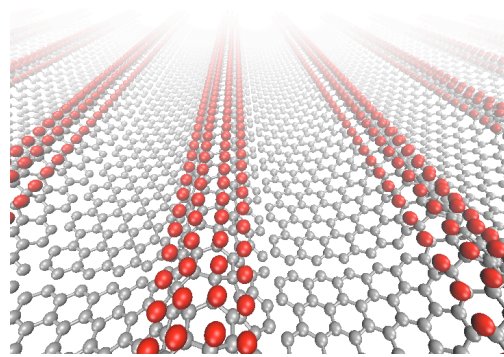
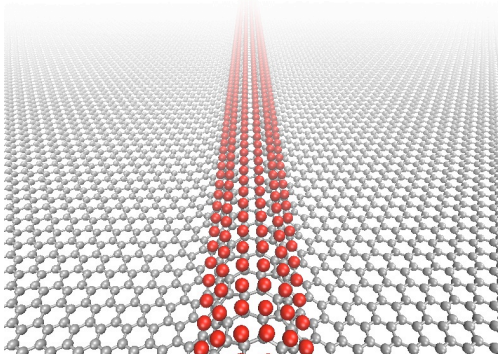
Towards real devices

- ❖ Create and maintain an extended (multilayer) structure
 - “Spacers” molecules are currently under consideration to create “pillared” multilayer structure
- ❖ Control the curvature
 - if designed sensitive to external stimuli (e.g. light or E/M fields)
 - pillars could also serve to create and control the curvature



Towards real devices

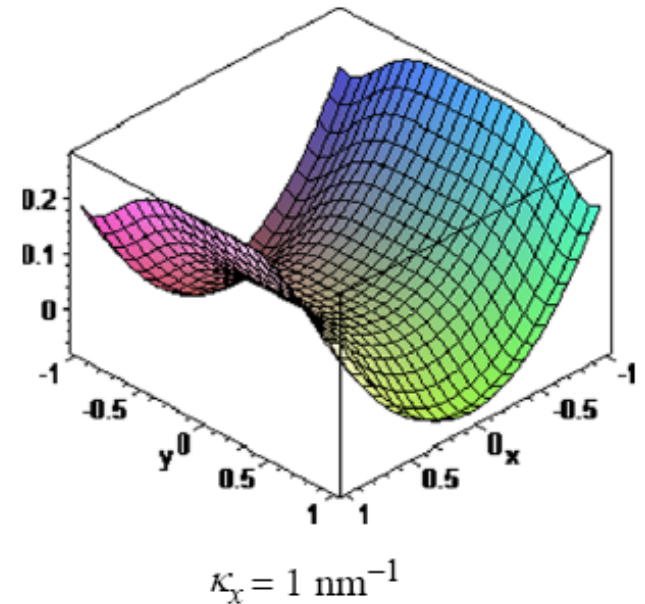
- ❖ Create and maintain an extended (multilayer) structure
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pillars could also serve to create and control the curvature
- ❖ Evaluate the effects on the macroscopic scales



Towards real devices

❖ Evaluate the effects on the macroscopic scales

- ❖ System: 2D surface, virtually infinite, any geometry
- ❖ Linear Elasticity
 - Bending energy
 - Theory of plates
- ❖ Interactions
 - Adhesion energy (theory of adhesion)
 - H binding
 - Self-consistent modification of the local curvatures
- ❖ Parameterization based on DFT and MM calculations. The following properties should be mapped onto the surface
 - Dependence between H binding probability/adhesion and curvature
 - elasticity and its dependence on H binding
 - Barriers and other energetics



This approach will allow to address the thermodynamics of the process and the macroscopic scales in time and space ⇒

- ❖ Direct comparison with experiment
- ❖ Devices design

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N. Pugno, A. Astron., 82, 221 (2013).
N.Pugno, J. Mech. Phys. Solid., 58, 1397 (2010)
X. Shi, Y. Cheng, N. Pugno, H. Gao, Small, 6, 739 (2010)

Conclusions

- ❖ Multi-scale approaches are useful to design new devices, since they couple the accuracy and microscopic physics to the macroscopic effects
- ❖ We addressed several problems related to the design graphene-based devices for energy and nano-electronics applications
 - ⦿ control of reactivity by means of control of curvature
 - ⦿ different means of curvature control (mechanical, electric field, chemical, in progress)
 - ⦿ how to create stable multi-layers (in progress)
 - ⦿ how to evaluate macroscopic energetics and thermodynamics (in progress)

Collaborators

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NANO-Cnr

NANO-Cnr PostDoc Flagship

Università di Pisa, undergraduate student

Università di Pisa, undergraduate student

NANO-Cnr (experiment)

iit Graphene Labs (experiment)

Univ Trento, FBK

Support

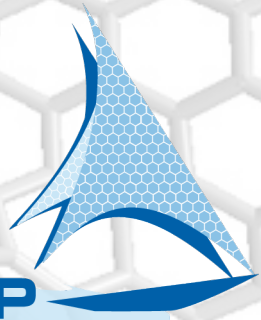


SuperComputing Applications and Innovation

SCAI



Platform
Computation



GRAPHENE FLAGSHIP



Toulouse, May 8th 2014

