

PHOSPHORENE: THE P-ANALOGUE OF GRAPHENE

Maria Caporali,^a Manuel Serrano-Ruiz,^a Andrea Ienco,^a Stefan Heun,^b Maurizio Peruzzini^a

^a National Research Council - Institute of Chemistry of Organometallic Compounds, Via Madonna del Piano 10, 50019-Sesto Fiorentino, Florence, Italy

^b National Enterprise for Nanoscience and Nanotechnology (NEST), Istituto Nanoscienze-CNR and Scuola Normale Superiore, Piazza San Silvestro 12, 56127, Pisa, Italy

maria.caporali@iccom.cnr.it

Recently it was discovered that black phosphorus (BP), a bulk layered material, can be exfoliated either by Scotch tape method¹ or by ultrasound activation in a medium with high dielectric constant². The monolayer BP nanosheet, nicknamed phosphorene, which results from the exfoliation procedure, is a new 2D material endowed with peculiar characteristics, due to its puckered honeycomb structure, but unlike graphene, possesses intrinsic semiconductor properties. This opens the way to many important applications for high-performance electronic and optoelectronic devices.³ In this lecture, we'll focus on the preparation of phosphorene by liquid exfoliation in dimethylsulfoxide especially highlighting the unexpected role of water in the process.

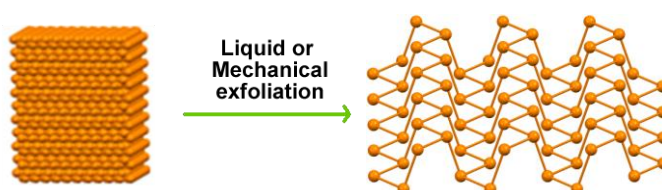


Figure 1. Top-down approach to prepare phosphorene from black phosphorus.

Moreover, our preliminary studies on the functionalization of phosphorene will be presented and discussed, highlighting, for instance, the immobilization of transition metal nanoparticles on the polyphosphorus network aimed at preparing new advanced nanostructured catalysts.

Acknowledgement: Thanks are expressed to EC for funding the project PHOSFUN “Phosphorene functionalization: a new platform for advanced multifunctional materials” (ERC ADVANCED GRANT 2015 – 2019; grant agreement n°670173) under the European Union’s Horizon 2020 research and innovation programme.

References

¹ a) Reich, E.S.; *Nature* **2014**, *506*, 19; b) Li, K.; Yu, Y.J.; Ye, G.J.; Ge, Q.Q.; Ou, X.D.; Wu, H.; Feng, D.L.; Chen, X.H.; Zhang, Y.B.; *Nat. Nanotechnol.* **2014**, *9*, 372-377. ² a) Brent, R.; Savjani, N.; Lewis, E.A.; Haigh, S.J.; Lewis, D. J.; O’Brien, P.; *Chem. Commun.* **2014**, *50*, 13338-13341; b) Yasaei, P.; Kumar, B.; Foroozan, T.; Wang, C.; Asadi, M.; Tuschel, D.; Indacochea, J.E.; Klie, R. F.; Salehi-Khojin, A.; *Adv. Mater.* **2015**, *27*, 1887-1892. ³ Zhang, X.; Tan, C.; Lin, J.; Li, H.; Sun, L.; Chen, W.; Huang, W.; Zhang, H.; *Angew. Chem. Int. Ed.* **2015**, *54*, 3653-3657.