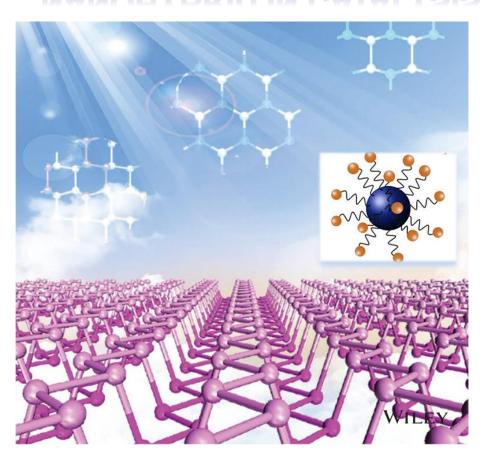
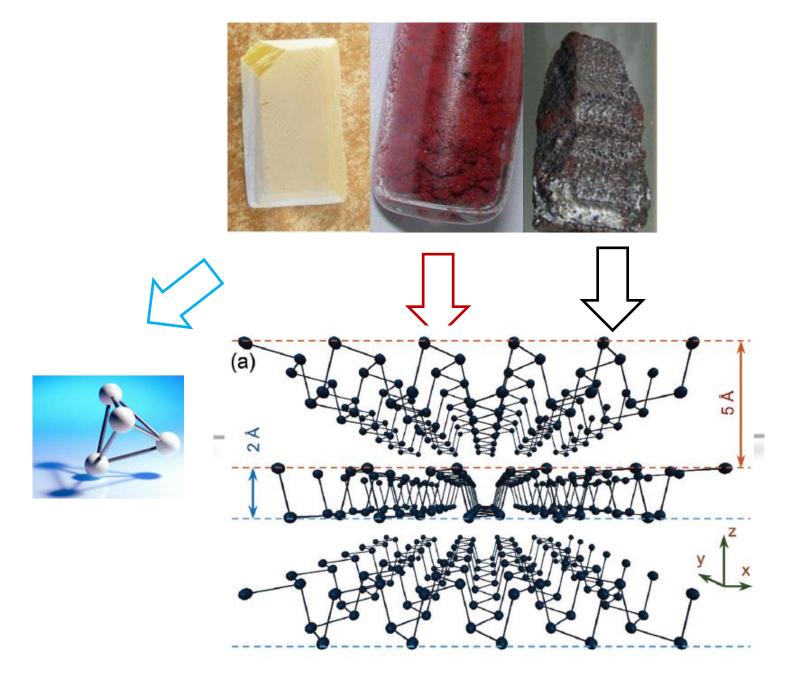
# DECORATING FEW-LAYER BLACK PHOSPHORUS WITH NICKEL NANOPARTICLES AND APPLICATION OF THE NANOHYBRID IN CATALYSIS



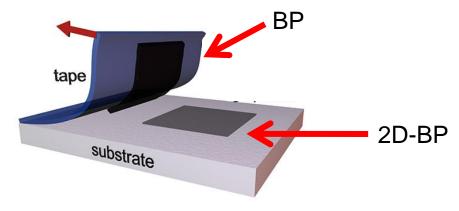
Maria Caporali
CNR ICCOM, Florence (ITALY)





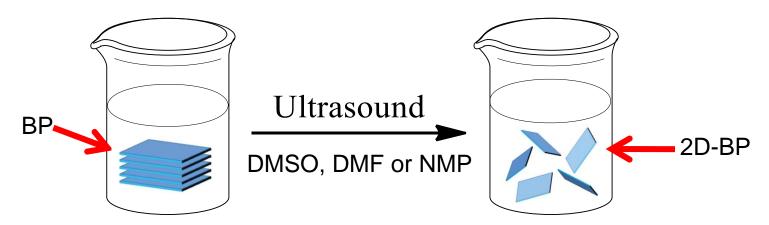
#### Preparation of few-layer BP

#### ✓ Mechanical exfoliation



Ye et al. ACS Nano 2014, 8, 4033; Zhang, Nat. Nanotechnol. 2014, 9, 372

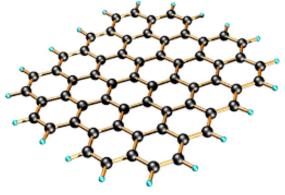
#### ✓ Liquid phase exfoliation



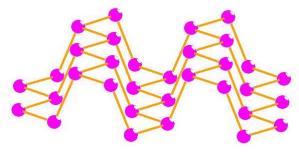
Chem. Commun. **2014**, 50, 13338; Nano Lett. **2014**, 14, 6964; ACS Nano **2015**, 9, 3596; Adv. Mat. **2015**, 27, 1887; 2D Materials, **2014**, 1,11002.

#### **2D Materials**

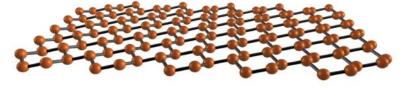
#### **Elemental 2D materials**



graphene

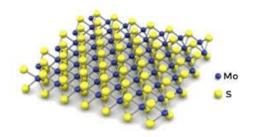


phosphorene

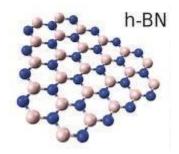


silicene, germanene, stanene

# 2D Materials composed by two (or more) elements

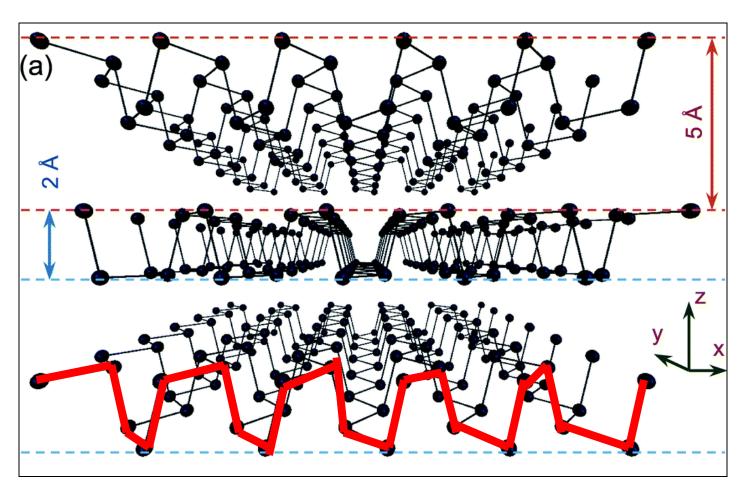


Molybdenum disulfide (MoS<sub>2</sub>)



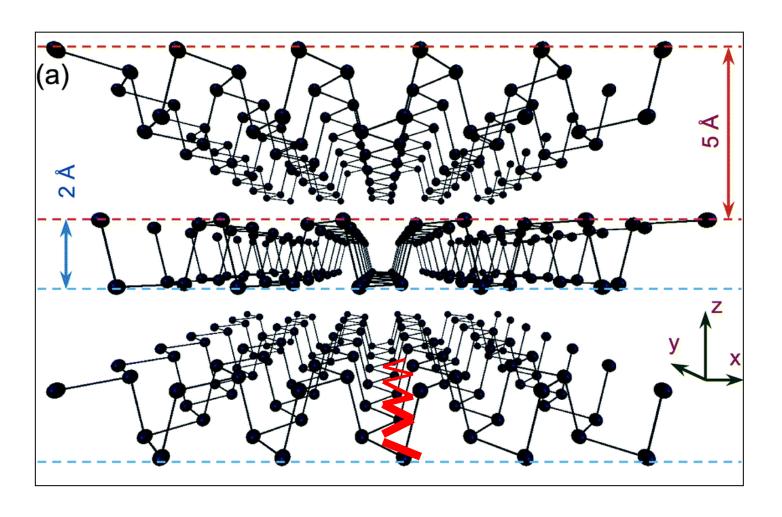
Hexagonal boron nitride (h-BN)

#### Anisotropic structure of black phosphorus



**Armchair (x-axis)** 

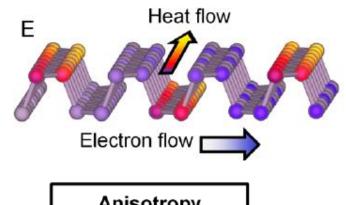
#### Anisotropic structure of black phosphorus



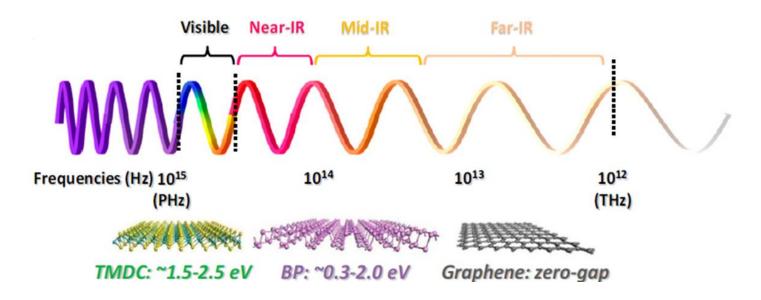
Zig-zag (y-axis)

- ✓ High carrier mobility: 1000 cm²/Vs
- ✓ On / off ratio:  $10^3 10^5$
- ✓ Thermal conductivity (300 K):

30 W/m K (zig-zag); 13.7 W/m K (armchair)

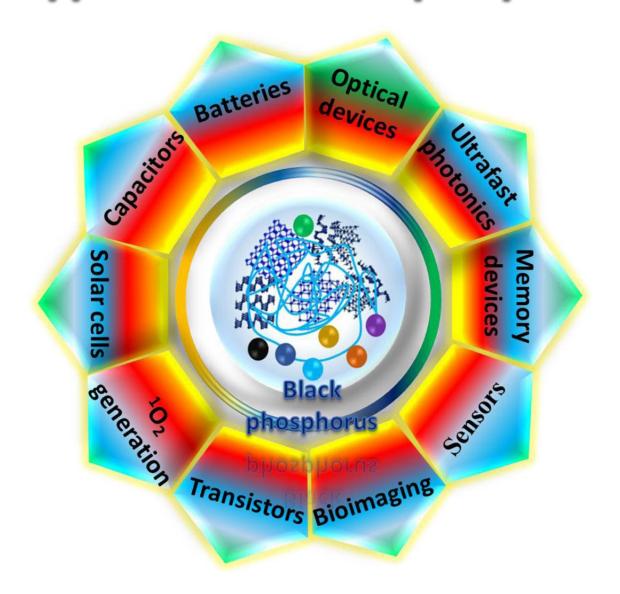


Anisotropy



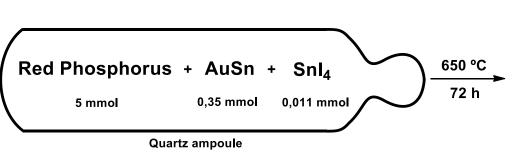
- ✓ p-type semiconductor, with a thickness-depending direct band gap (0.3-2.0 eV)
- ✓ The band gap can be modulated either applying an electrical field or by strain.

#### **Applications of black phosphorus**

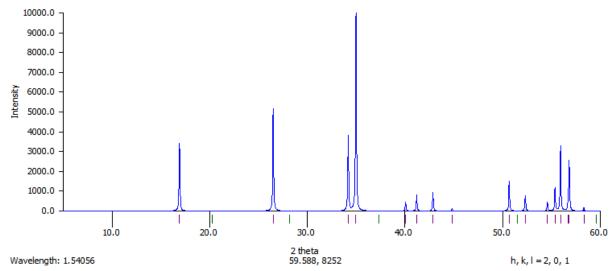


FlatChem. 2017, 2, 15-37

## Synthesis of Black Phosphorus

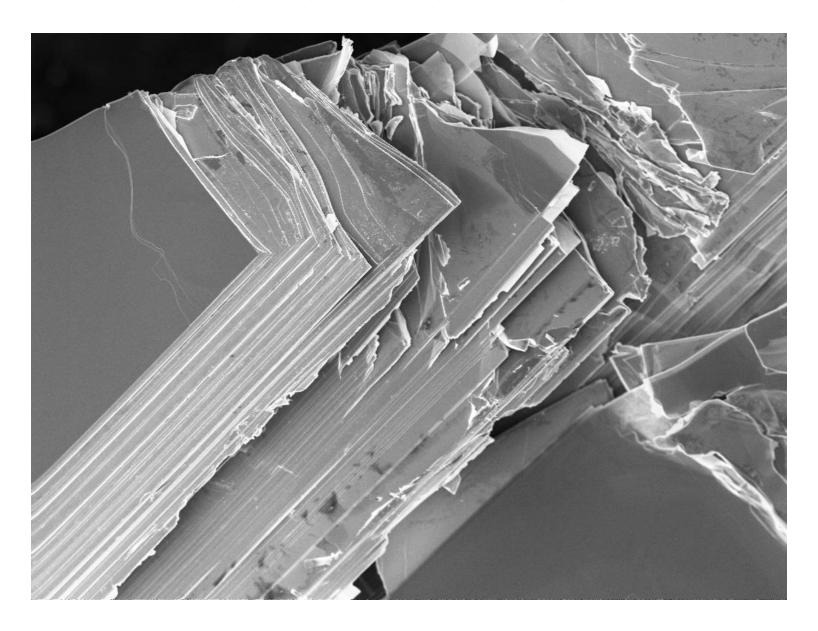




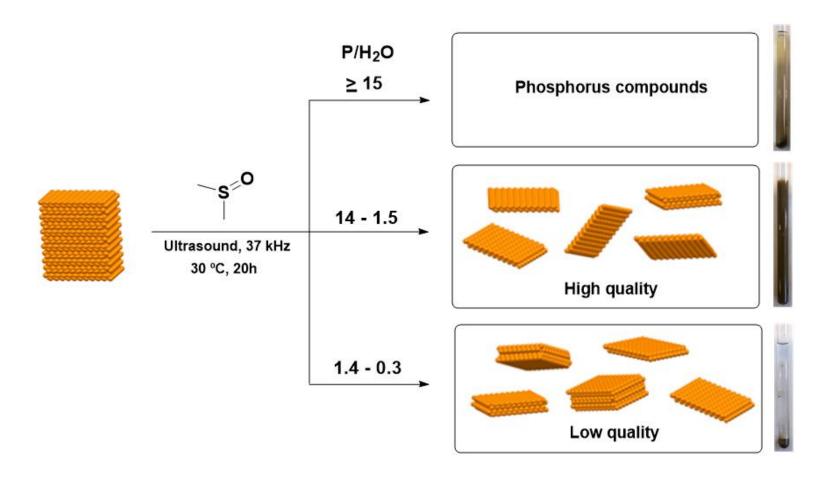


Inorg. Chem. **2007**, *4*6, 4028; J. Solid State Chem. **2008**, 181, 1707.

#### Micro-mechanical exfoliation

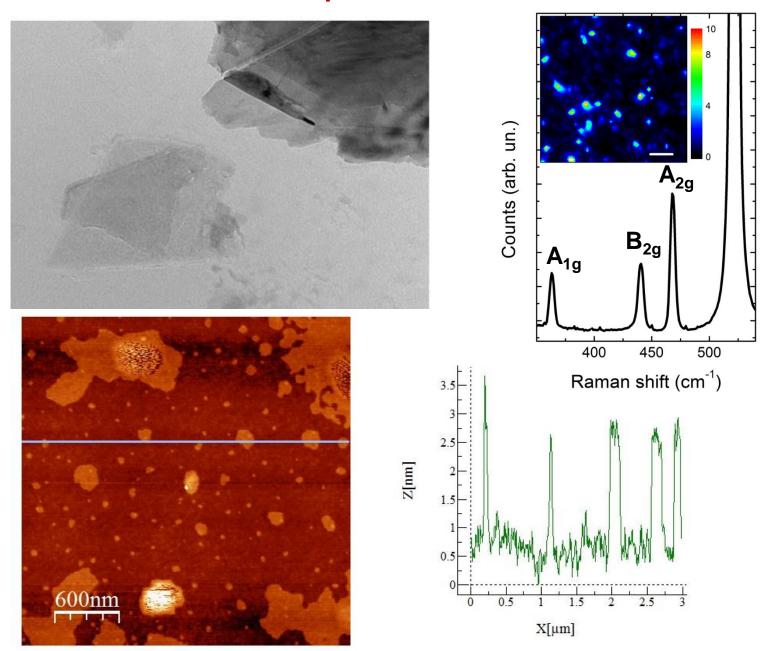


## Liquid-phase exfoliation

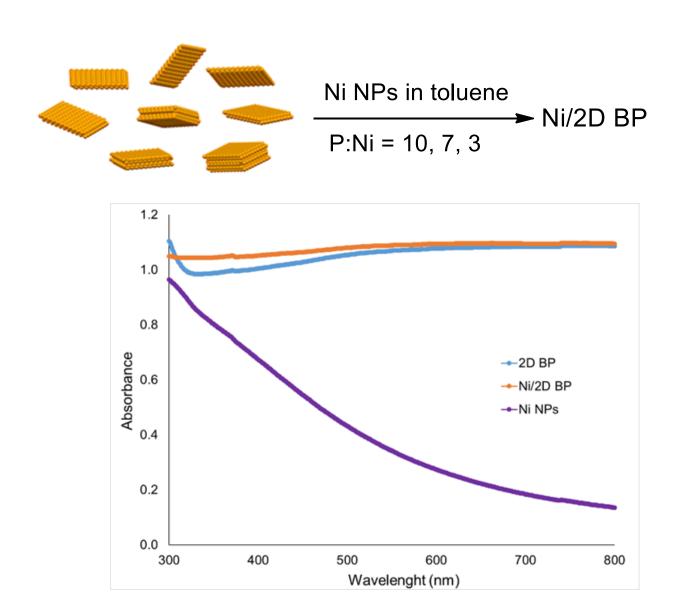


M. Serrano-Ruiz, M. Caporali, A. Ienco, V. Piazza, S. Heun, M. Peruzzini, *Adv. Mat. Interfaces* **2016**, 3, 1500441.

## 2D Black Phosphorus: characterization

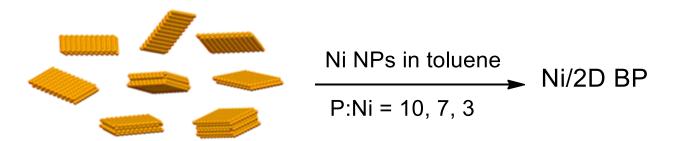


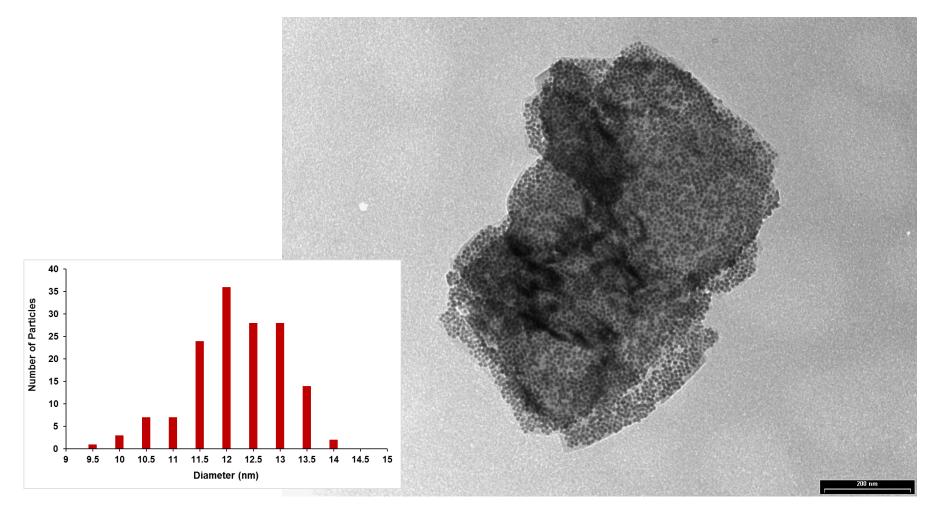
## Surface functionalization of 2D black P



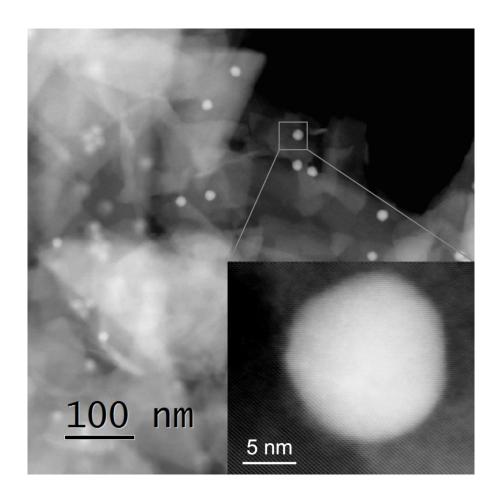
Mezailles et al. Chem. Mater. 2010, 22, 1340.

## Surface functionalization of 2D black P





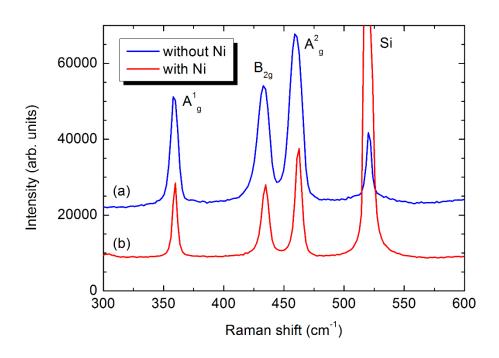
# HAADF STEM on Ni/2D BP



STEM-EELS gave chemical information of the surface of the nanohybrid.

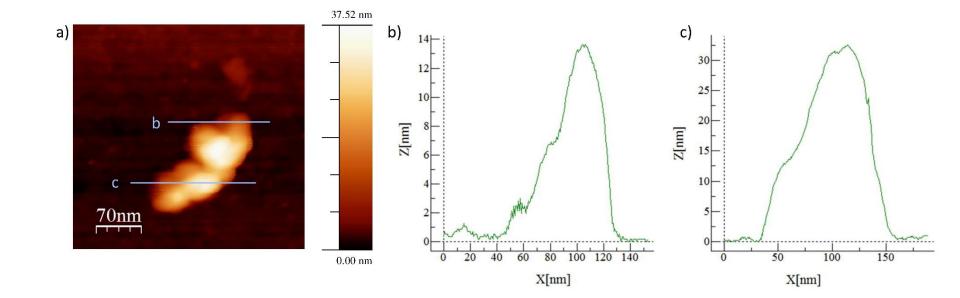


# Raman: comparison between pristine 2D BP and Ni/2D BP



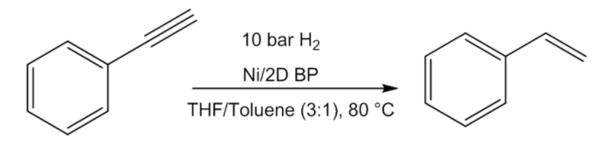


## **Atomic Force Microscopy**





#### Semihydrogenation of phenylacetylene



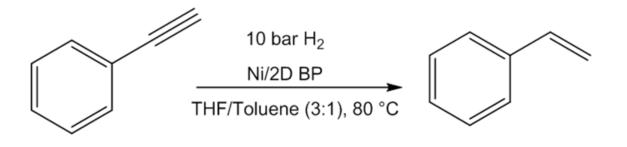
Entry	Conversion (%)	Selectivity to styrene (%)	S/cat	T (°C)
Ni NPs	100.0	78.6	56.0	80
2D BP	0.0	-	-	80
Ni/2D BP	93.2	92.8	56.0	80
$Ni/AI_2O_3$	99.6	0. <b>7</b> <sup>a</sup>	16.5	100
Ni/MgO	98.5	36.0 <sup>b</sup>	15.0	50
Ni@C	99.8	59.6 <sup>c</sup>	-	100-150

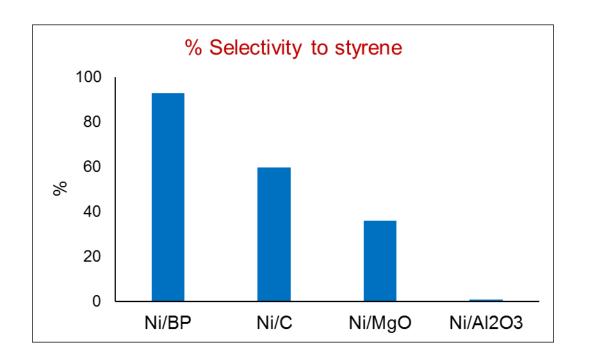
<sup>&</sup>lt;sup>a</sup>ACS Catal. **2015**, *5*, 5756: 2 hours, 3 bar H<sub>2</sub>

<sup>&</sup>lt;sup>b</sup> Chem. Cat. Chem. **2014**, 6, 824: 5 bar H<sub>2</sub>, 2 h

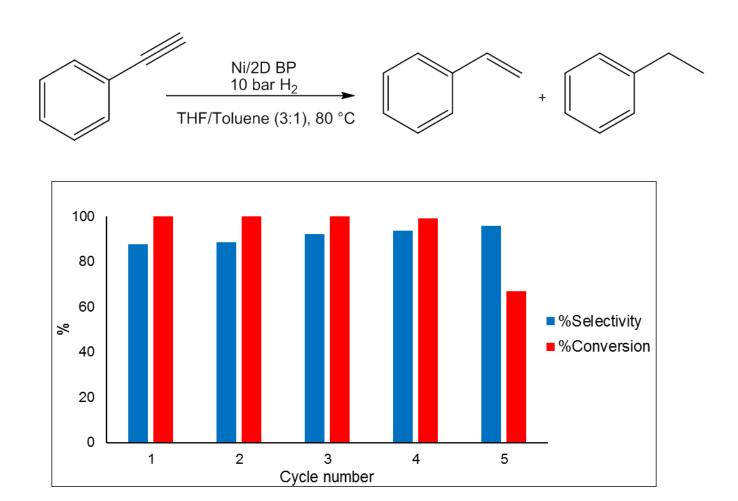
<sup>&</sup>lt;sup>c</sup> Carbon **2014**, 74, 291: flow bed reactor.

## Semihydrogenation of phenylacetylene





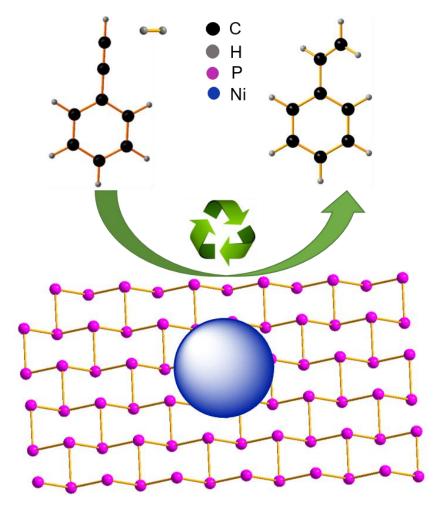
## Recycling Ni/2D BP



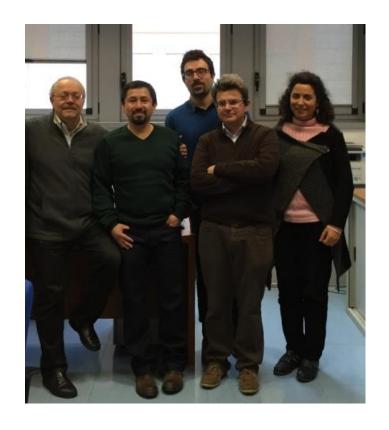
ICP-AES: no leaching of nickel

## Summary

- ➤ Nickel nanoparticles were dispersed on the surface of few-layer black phosphorus achieving a new nanohybrid Ni/2D BP.
- Ni/2D BP catalyzed the semihydrogenation of phenylacetylene and showed good catalytic activity and much higher selectivity than similar systems bearing Ni NPs supported on MgO, Al₂O₃ or graphene.
- The catalytic activity and selectivity remained unaltered after recycling tests.



#### Acknowledgements



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Giuseppe Nicotra Corrado Spinella



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