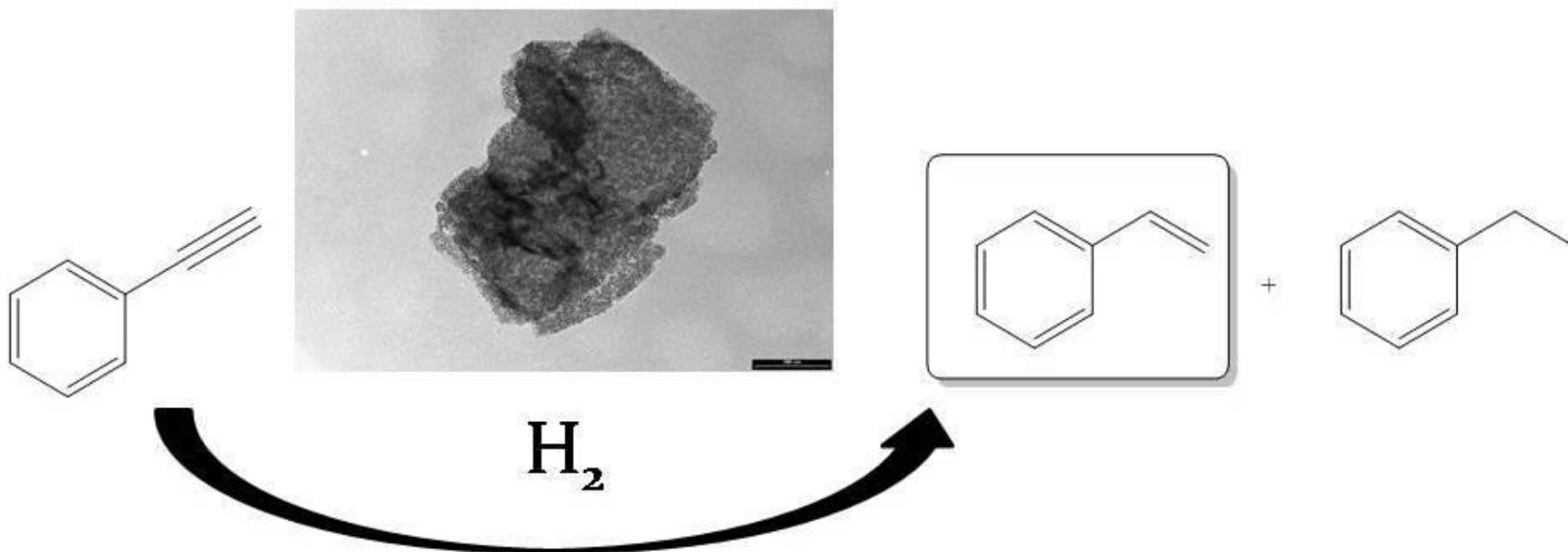


2D-BLACK PHOSPHORUS FUNCTIONALIZED WITH METAL NANOPARTICLES: FULL CHARACTERIZATION AND CATALYTIC ACTIVITY ON HYDROGENATION REACTIONS



Outline

Intro

- 2D-black phosphorus (BP): a new wonder material
- 2D-BP as novel support for metal nanoparticles

Experimental section

- Preparation and characterization of 2D-BP functionalized with TM nanoparticles
- Preliminary study on phenylacetylene hydrogenation catalyzed by NiNPs/2D-BP

Outline

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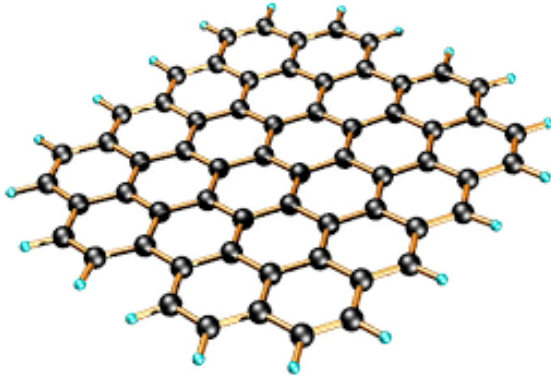
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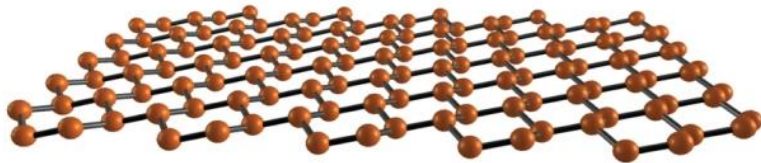
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2D-Materials

Elemental 2D-materials



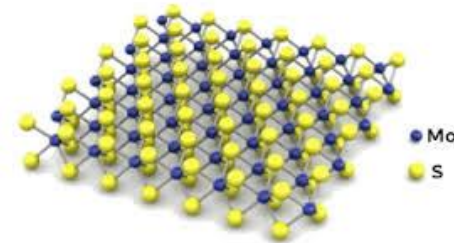
Graphene



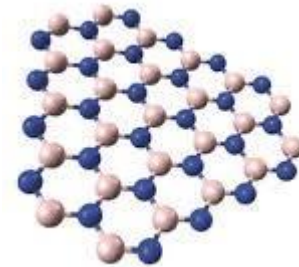
Silicene

.....

2D-Materials composed by two (or more) elements



Molybdenum disulfide (MoS₂)



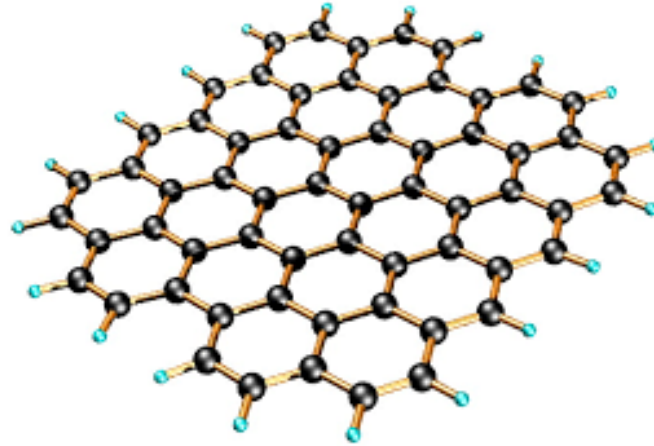
Hexagonal boron nitride (*h*-BN)

.....

2D-Materials

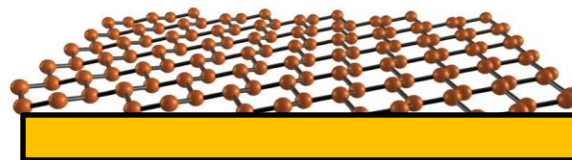
- Microelectronics**
- Optoelectronics**
- Catalysis**
- Chemical and biological sensors**
- Supercapacitors**
- Solar cells**
- Lithium ion batteries**

Elemental 2D-Materials



Graphene

Isolated through exfoliation



Other elemental 2D-materials

Epitaxially grown onto specific substrates

2D-black phosphorus (BP): a new wonder material?



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APS March Meeting 2013

Volume 58, Number 1

Monday–Friday, March 18–22, 2013; Baltimore, Maryland

Session J6: Growth, Structure, Properties, and Defects

2:30 PM–5:30 PM, Tuesday, March 19, 2013

Room: 302

Sponsoring Unit: DMP

Chair: Paola Barbara, Georgetown University

Abstract ID: BAPS.2013.MAR.J6.13

Abstract: J6.00013 : Electronic Properties of Few-layer Black Phosphorus

4:54 PM–5:06 PM

[Preview Abstract](#)

[Manual on J6](#) | [← Abstract →](#)

Abstract: J6.00013 : Electronic Properties of Few-layer Black Phosphorus 4:54 PM–5:06 PM

(Hefei National Laboratory for Physical Science at Microscale and Department of Physics, University of Science and Technology of China)

X.H. Chen

(Hefei National Laboratory for Physical Science at Microscale and Department of Physics, University of Science and Technology of China)

Yuanbo Zhang

(State Key Laboratory of Surface Physics and Department of Physics, Fudan University)

Black phosphorus is a layered allotropy of phosphorus that closely resembles graphite. In a single atomic layer, phosphorus atoms are covalently bonded into a puckered honey comb structure. All five valence electrons are localized, so unlike graphene monolayer black phosphorus is a semiconductor with a band gap of ~ 2 eV. In a bulk crystal the interlayer coupling reduces the band gap to ~ 0.3 eV. Using mechanical exfoliation method, we have successfully fabricated few layer black phosphorus field effect transistors. Our samples exhibit bipolar behavior with on-off ratio up to 10^6 , and a low off-state current. Electronic mobilities up to ~ 1000 $\text{cm}^2\text{V}^{-1}\text{s}^{-1}$ are currently achieved, with possibilities for further improvement. Such characteristics make black phosphorus a potential candidate for future nanoelectronic applications.

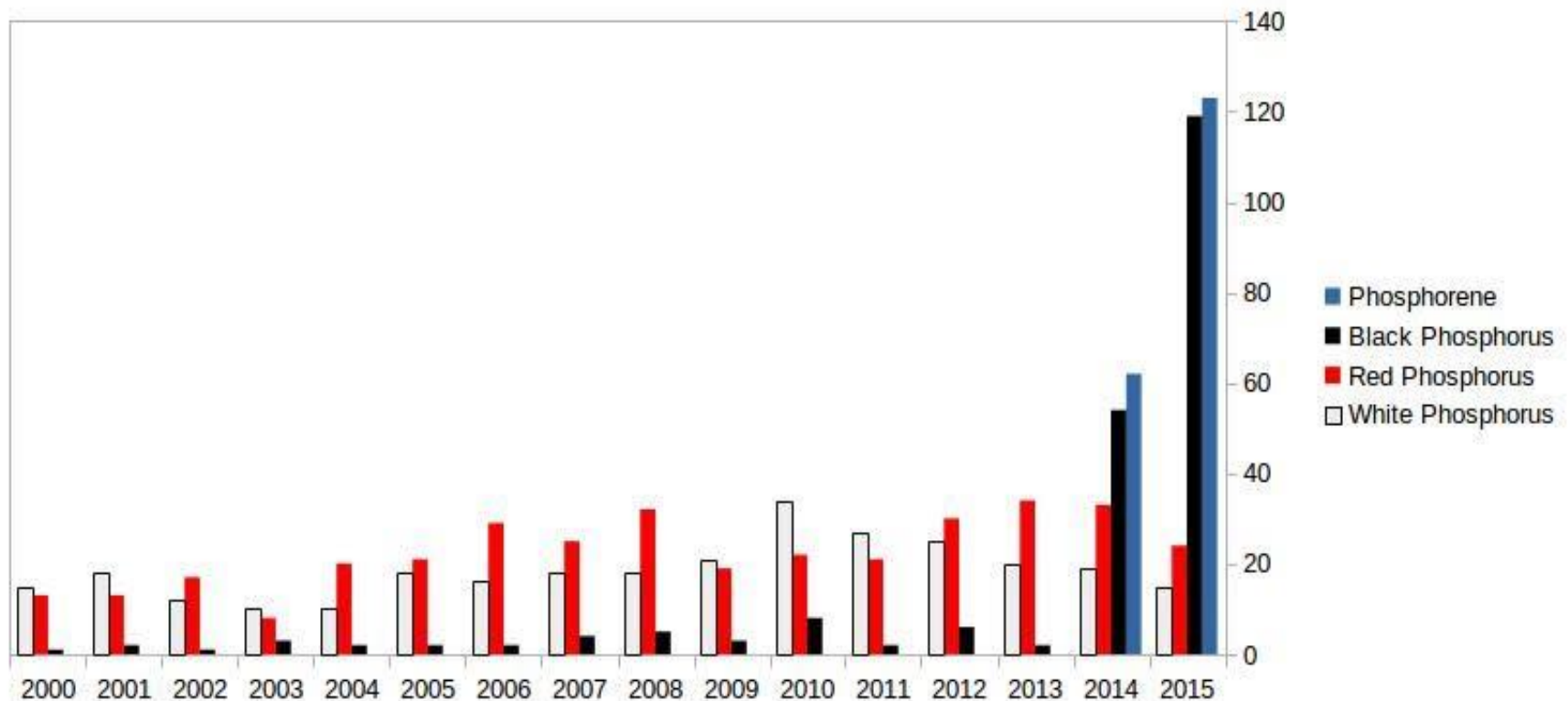
Li, L.; Yu, Y.; Ye, G. J.; Chen, X. H.; Zhang, Y. Electronic Properties of Few-Layer Black Phosphorus. In APS March Meeting; American Physical Society: College Park, MD, 2013; abstract ID Most of the breakthroughs in this field are still to come. The Journal of Physical Chemistry Letters Perspective DOI: 10.1021/acs.jpcclett.5b01686 J. Phys. Chem. Lett. 2015, 6, 4280–4291 4286 BAPS.2013.MAR.J6.13, <http://meetings.aps.org/Meeting/MAR13/Session/J6.13>.

2D-BP: a new wonder material?

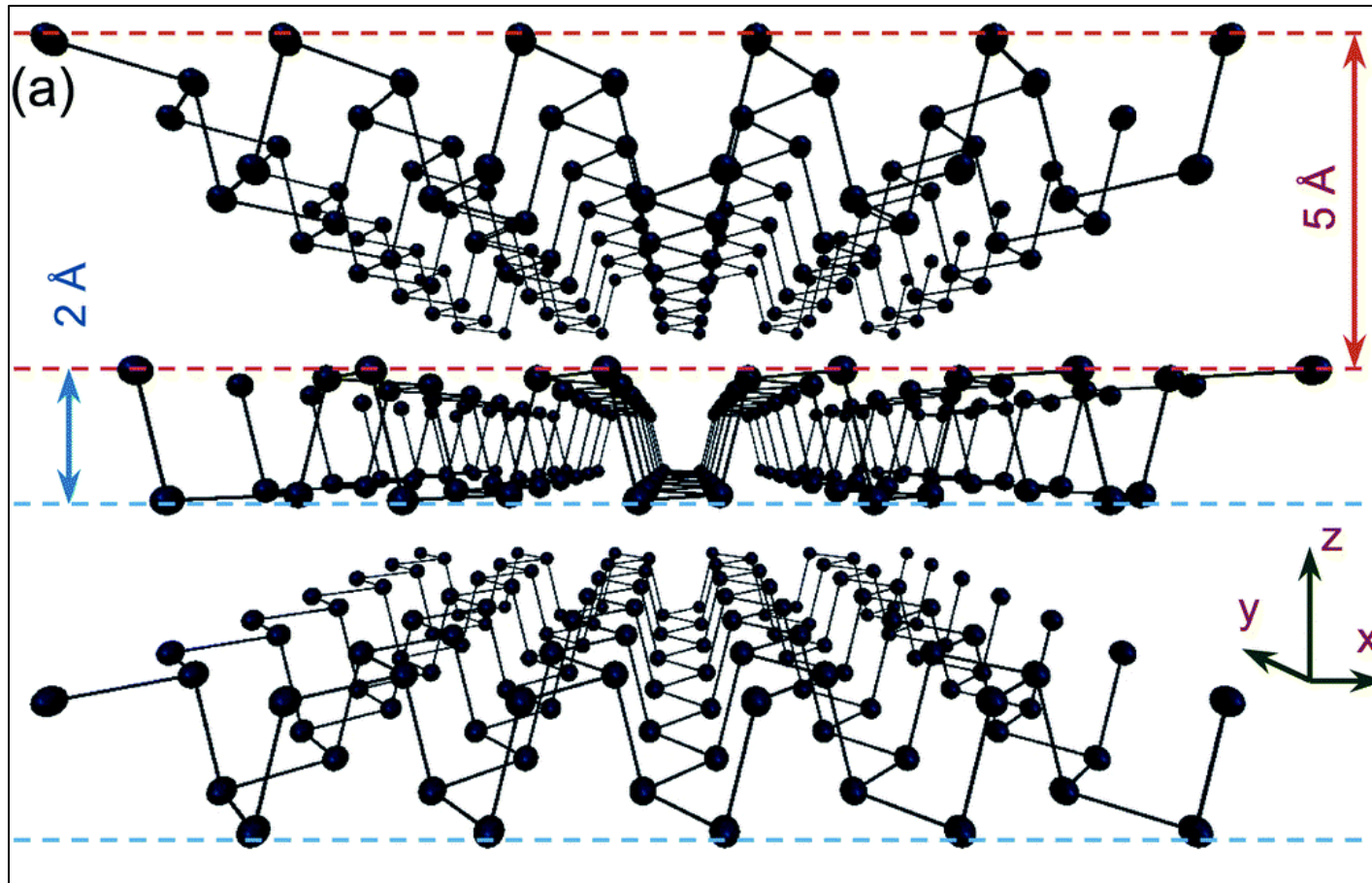
The renaissance of black phosphorus

Xi Ling^a, Han Wang^{b,1}, Shengxi Huang^a, Fengnian Xia^c, and Mildred S. Dresselhaus^{a,d,1}

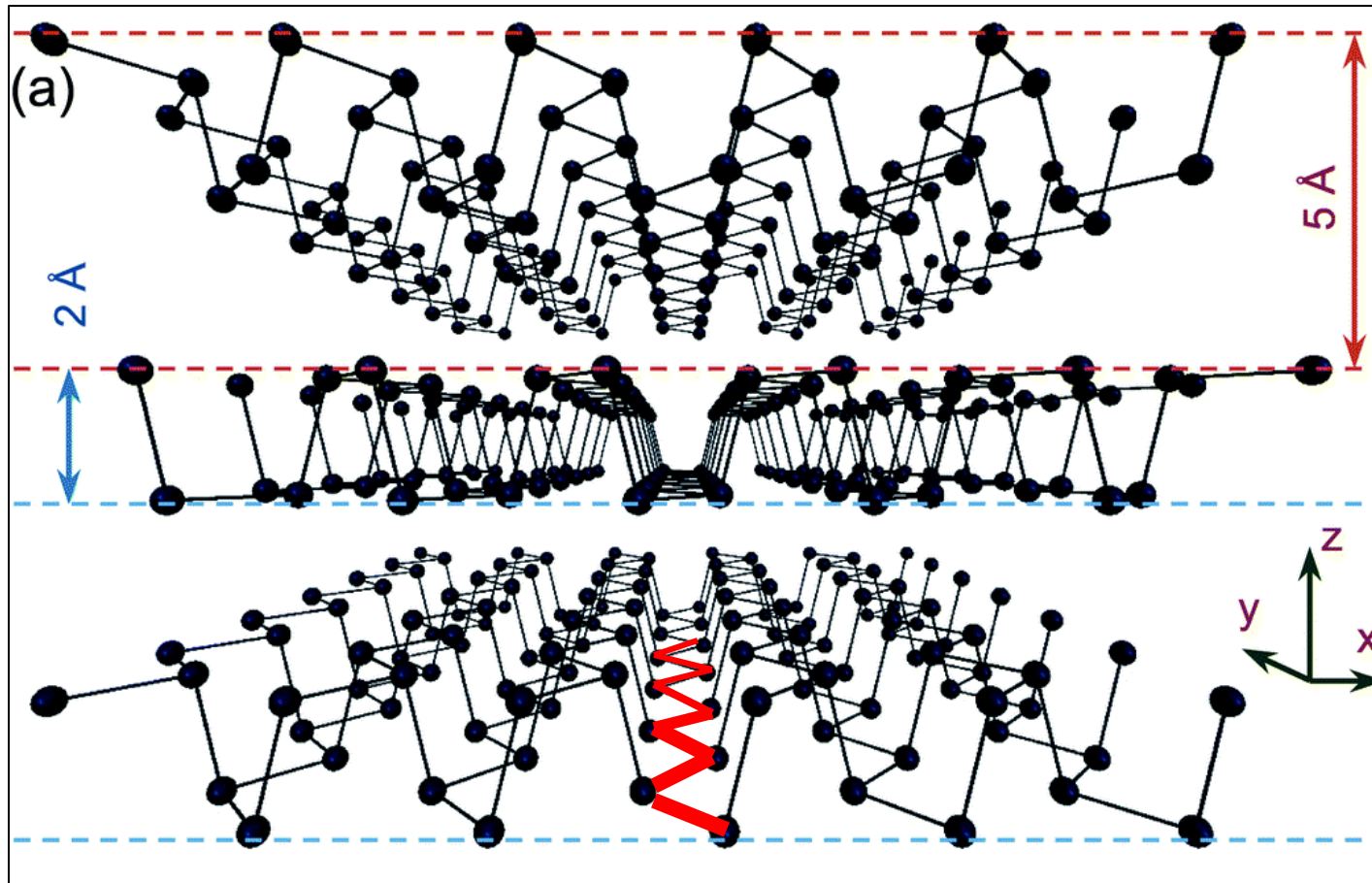
^aDepartment of Electrical Engineering and Computer Science and ^dDepartment of Physics, Massachusetts Institute of Technology, Cambridge, MA 02139; ^bMing Hsieh Department of Electrical Engineering, University of Southern California, Los Angeles, CA 90089; and ^cDepartment of Electrical Engineering, Yale University, New Haven, CT 06511



Crystal structure of BP

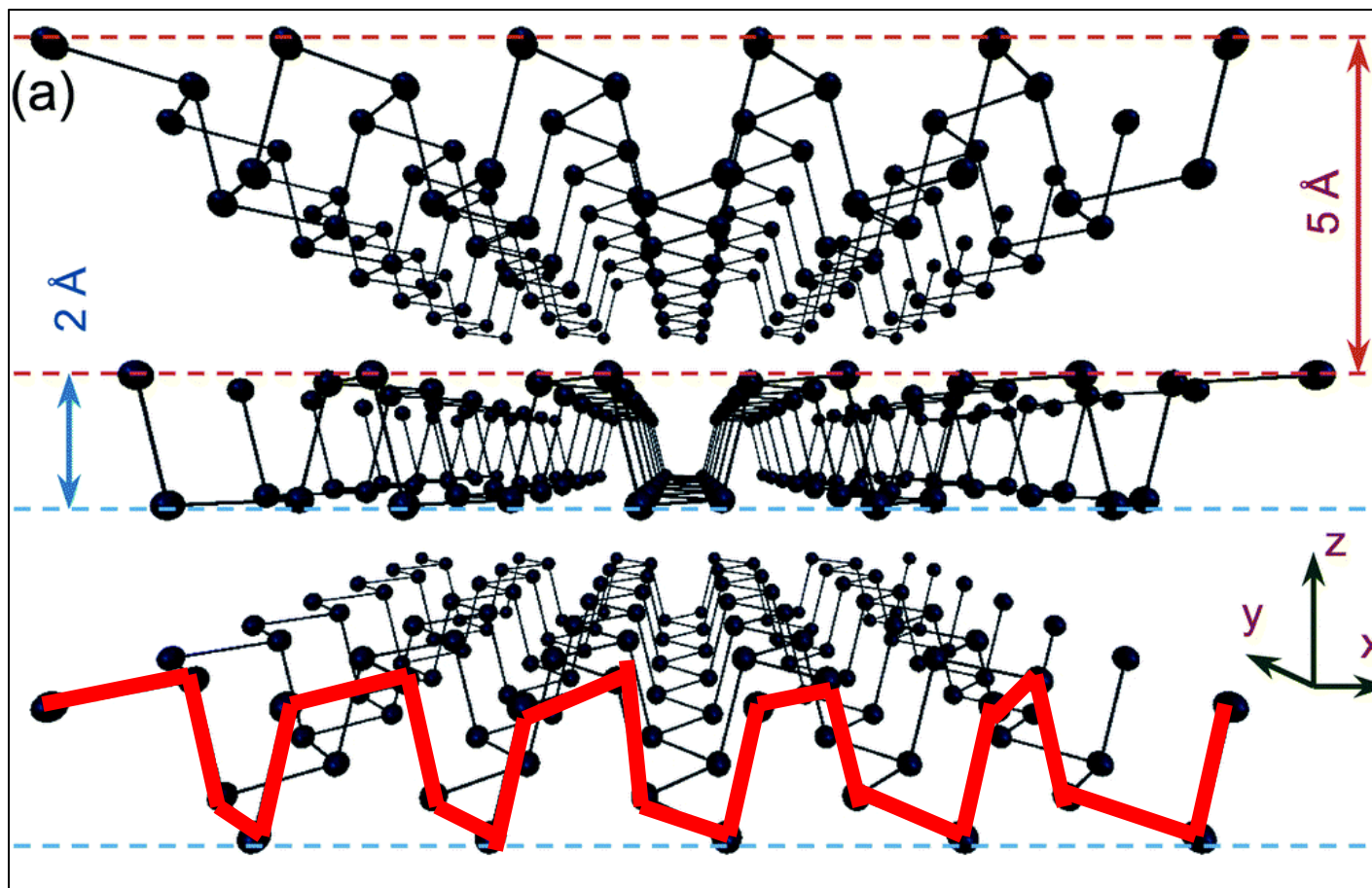


Anisotropic structure of BP



Zig-zag (y-axis)

Anisotropic structure of BP



Armchair (x-axis)

2D-BP: highlights

- ✓ Semiconductor, with a thickness-dependent direct band gap ($\sim 2 - 0.3$ eV)
- ✓ Good carrier mobility (~ 1000 cm² V⁻¹ s⁻¹ at room temperature)
- ✓ High surface area
- ✓ Higher chemical reactivity compared to other 2D-materials stable in free standing form
 - Careful handling in inert environments/encapsulation/passivation
 - Controllable/mild chemical functionalization (not possible for graphene)

2D-BP: highlights

✓ Semiconductor, with a thickness-dependent direct band gap ($\sim 2 - 0.3$ eV)

- Communication
- Light vision goggles
- Thermal infrared imaging
- Astronomy
- Medicine
- Document security systems
- DVD players
- Display
- LED, OLED

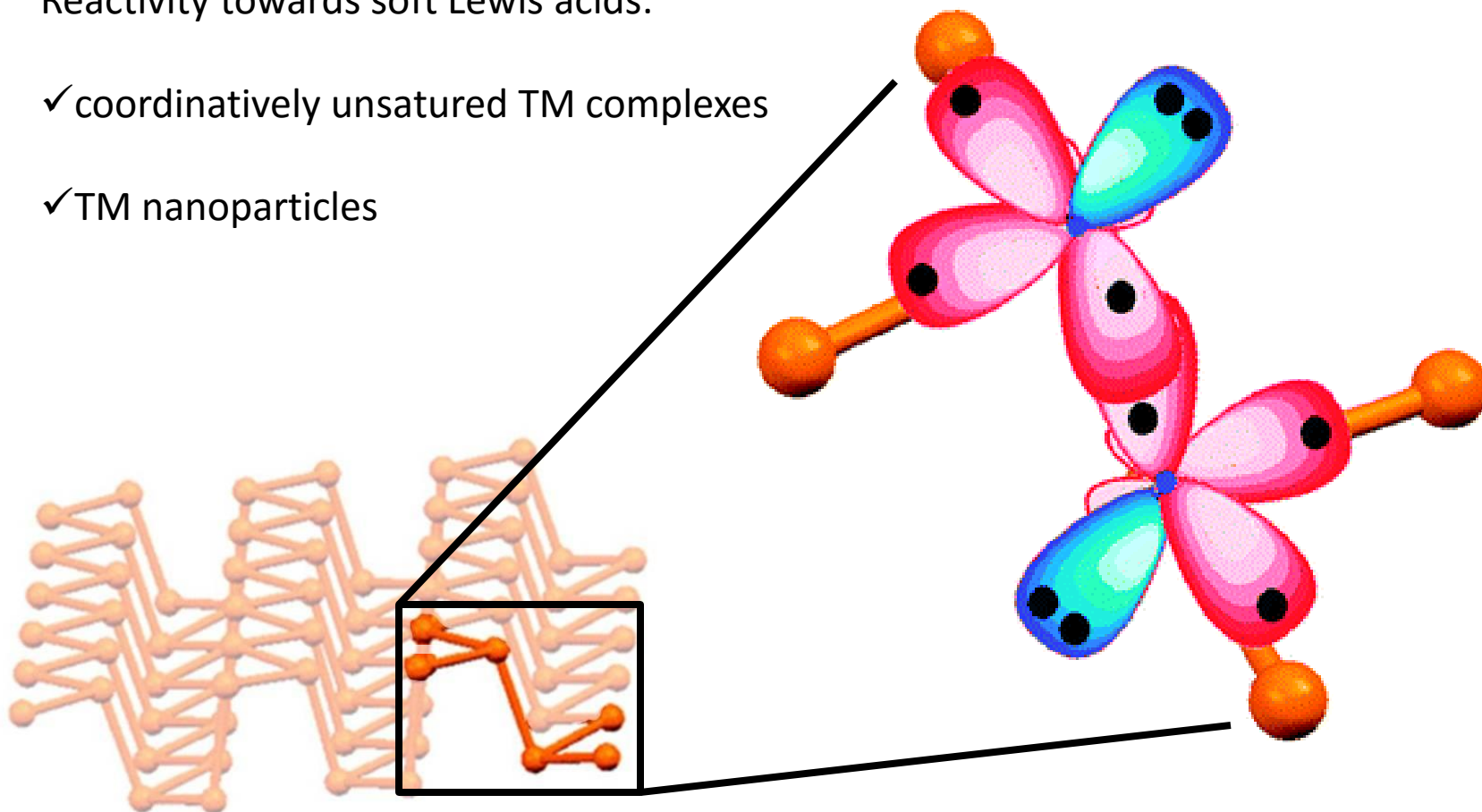
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2D-BP functionalization: a coordination approach?

Reactivity towards soft Lewis acids:

- ✓ coordinatively unsaturated TM complexes
- ✓ TM nanoparticles

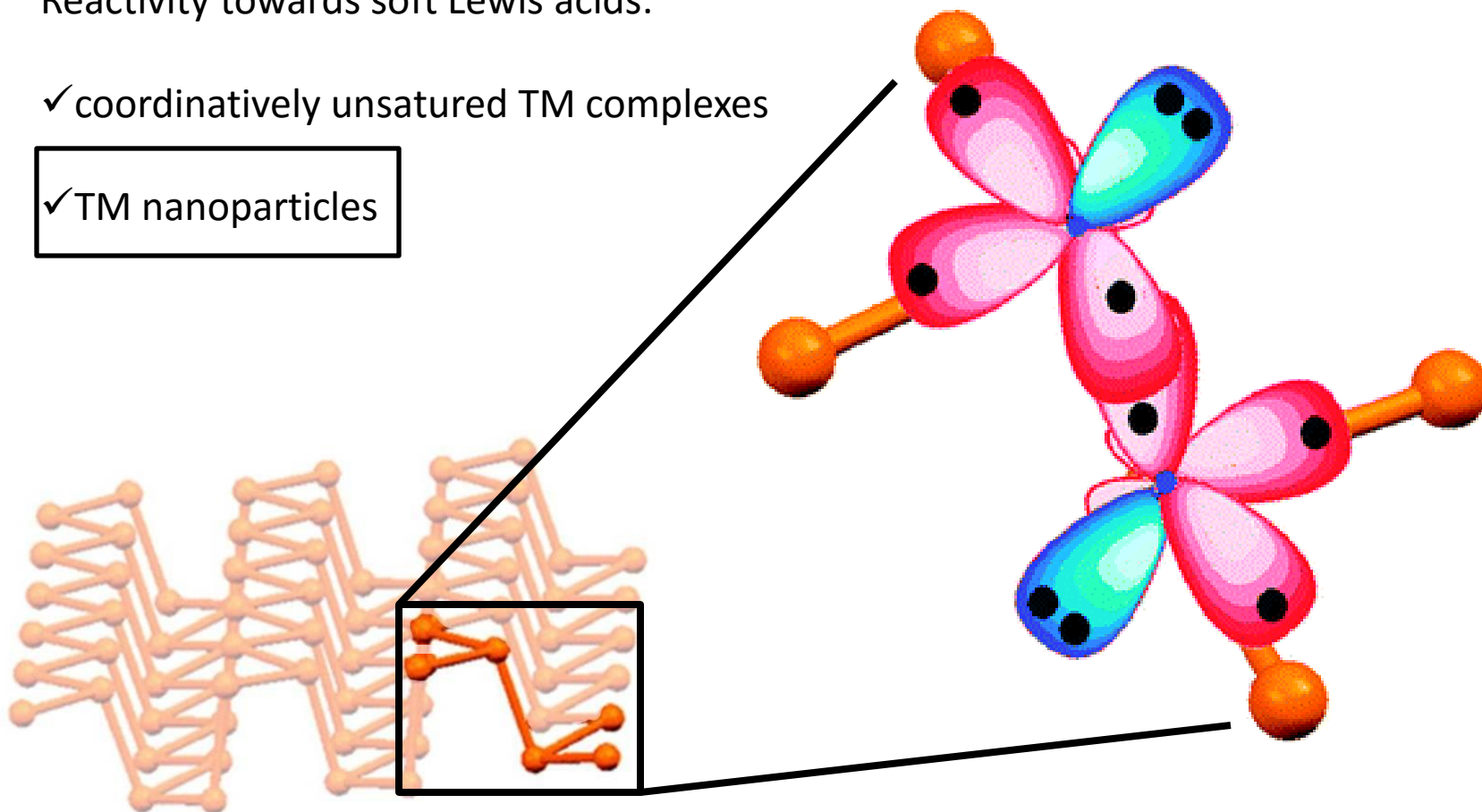


2D-BP functionalization: a coordination approach?

Reactivity towards soft Lewis acids:

✓ coordinatively unsaturated TM complexes

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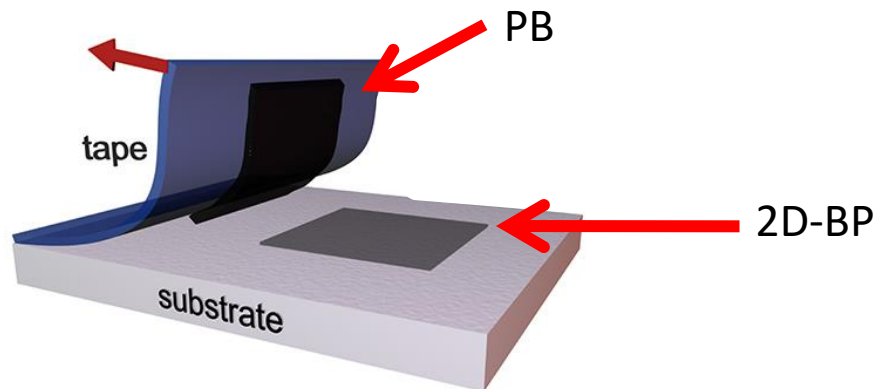
- 2D-black phosphorus (BP): a new wonder material
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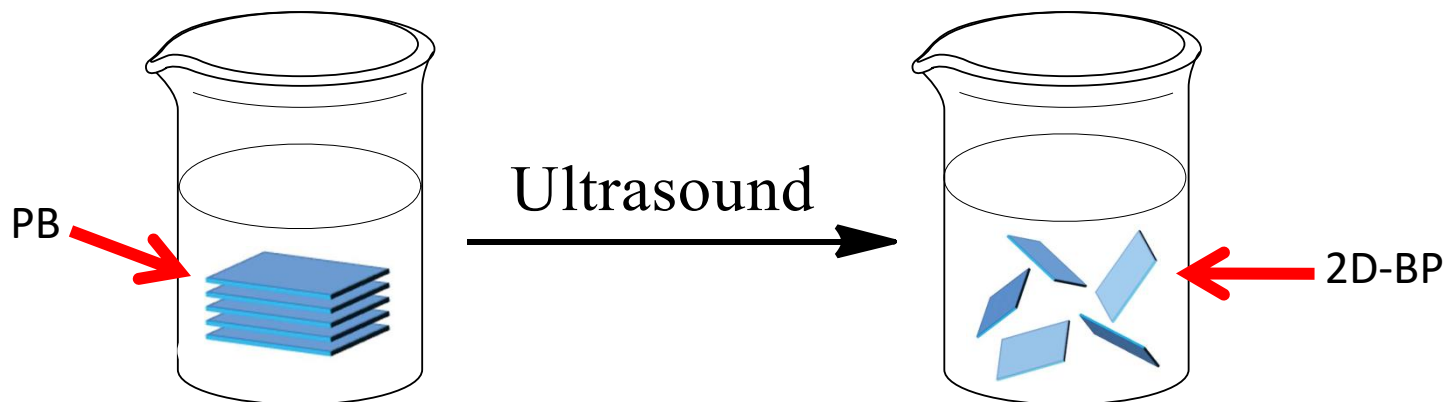
- Preparation and characterization of 2D-BP functionalized with TM nanoparticles
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Preparation of 2D-BP

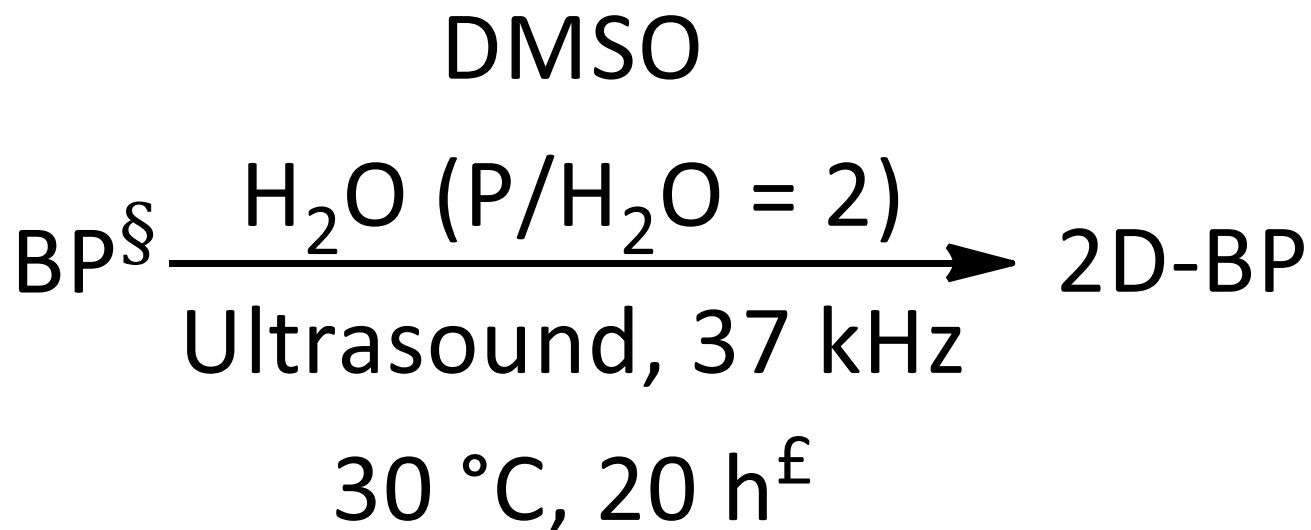
✓ Mechanical exfoliation



✓ Liquid phase exfoliation



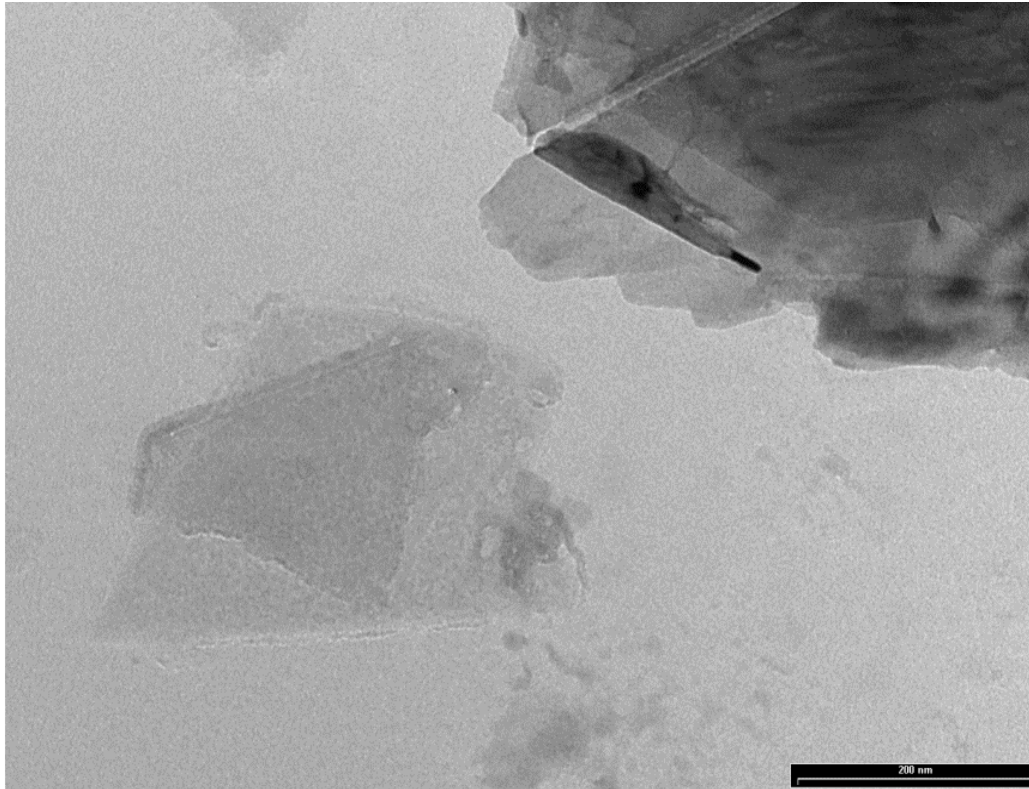
Preparation of 2D-BP in our lab



[§]Prepared as described in: T. Nilges, M. Kersting, T. Pfeifer, *J. Solid State Chem.* **2008**, *181*, 1707-1711.

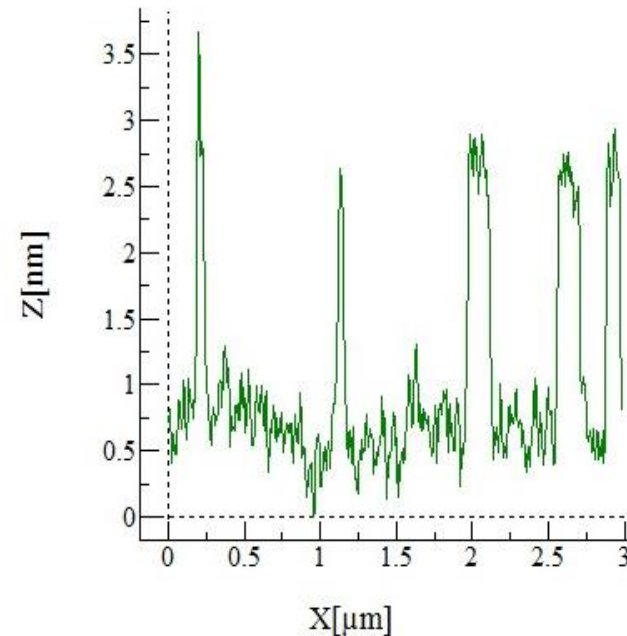
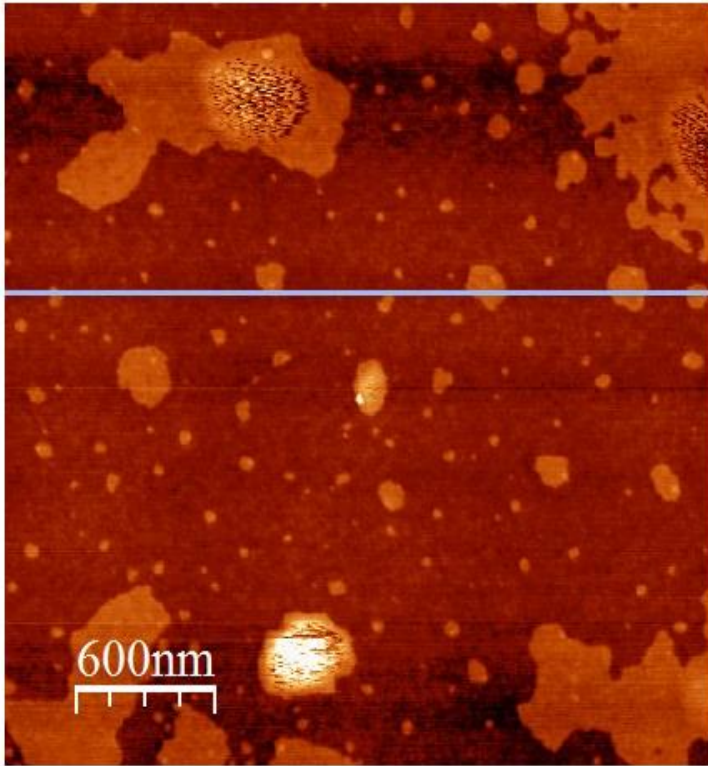
[£]Exfoliation conditions as described in: M. Serrano-Ruiz, M. Caporali, A. Ienco, V. Piazza, S. Heun, M. Peruzzini, *Adv. Mat. Interfaces* **2015**, *3*, 1500441.

Bright-field TEM image of 2D-BP



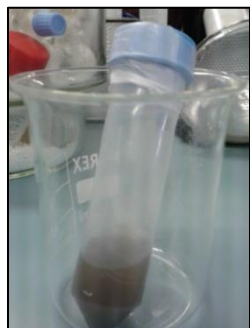
M. Serrano-Ruiz, M. Caporali, A. Ienco, V. Piazza, S. Heun, M. Peruzzini, *Adv. Mat. Interfaces* **2015**, 3, 1500441.

AFM analysis of 2D-black phosphorus

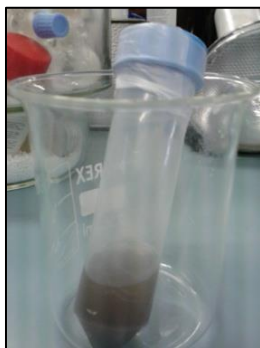


M. Serrano-Ruiz, M. Caporali, A. Ienco, V. Piazza, S. Heun, M. Peruzzini, *Adv. Mat. Interfaces* **2015**, 3, 1500441.

Extraction of 2D-BP from the DMSO of exfoliation

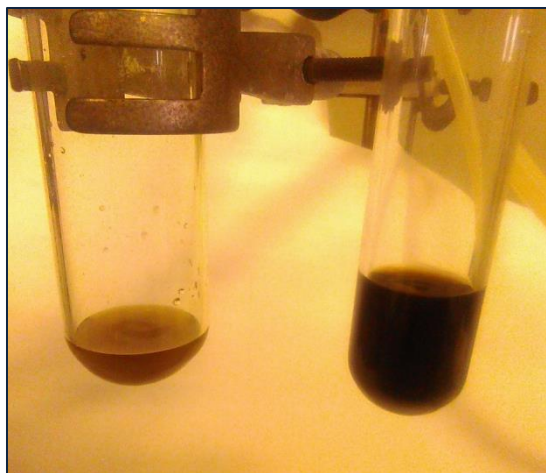
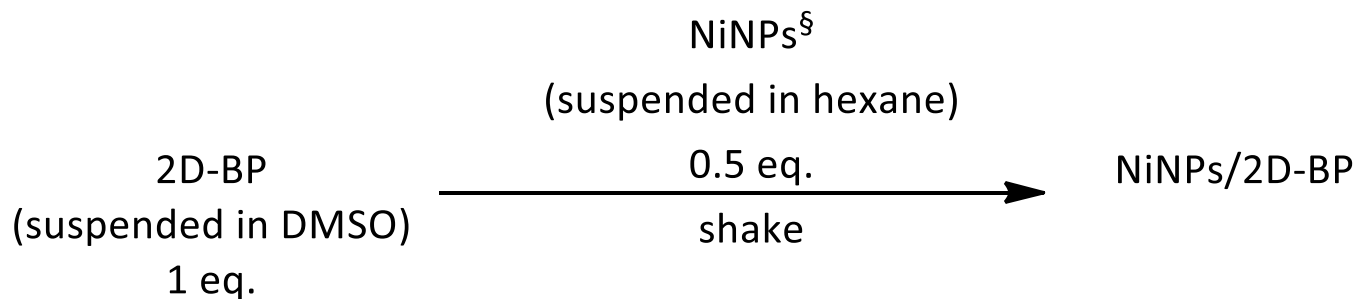


Acetone
Centrifugation (6000 rpm x 2) →



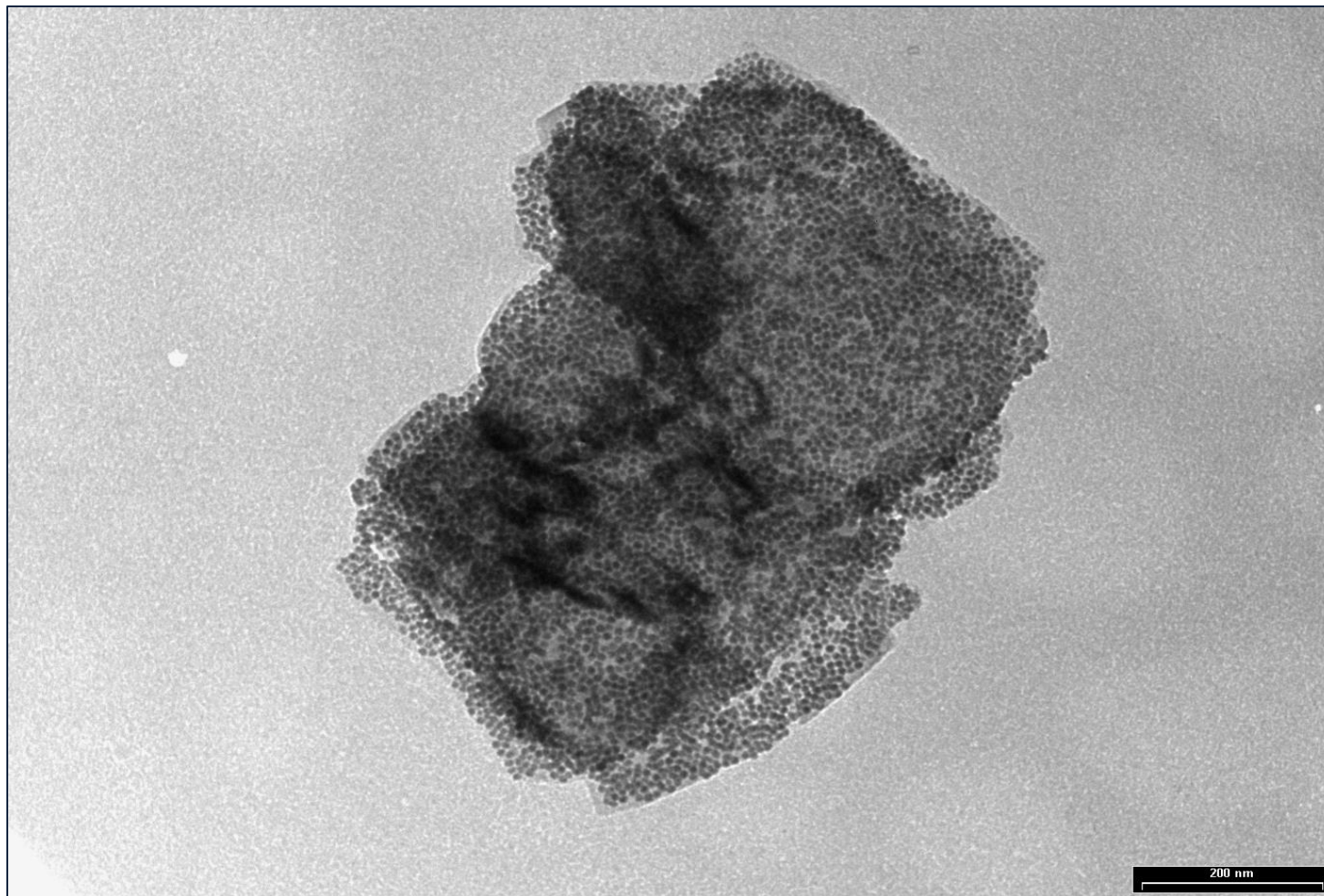
← Solvent, Ultrasound 37 kHz, 30"

Preparation of NiNPs/2D-BP

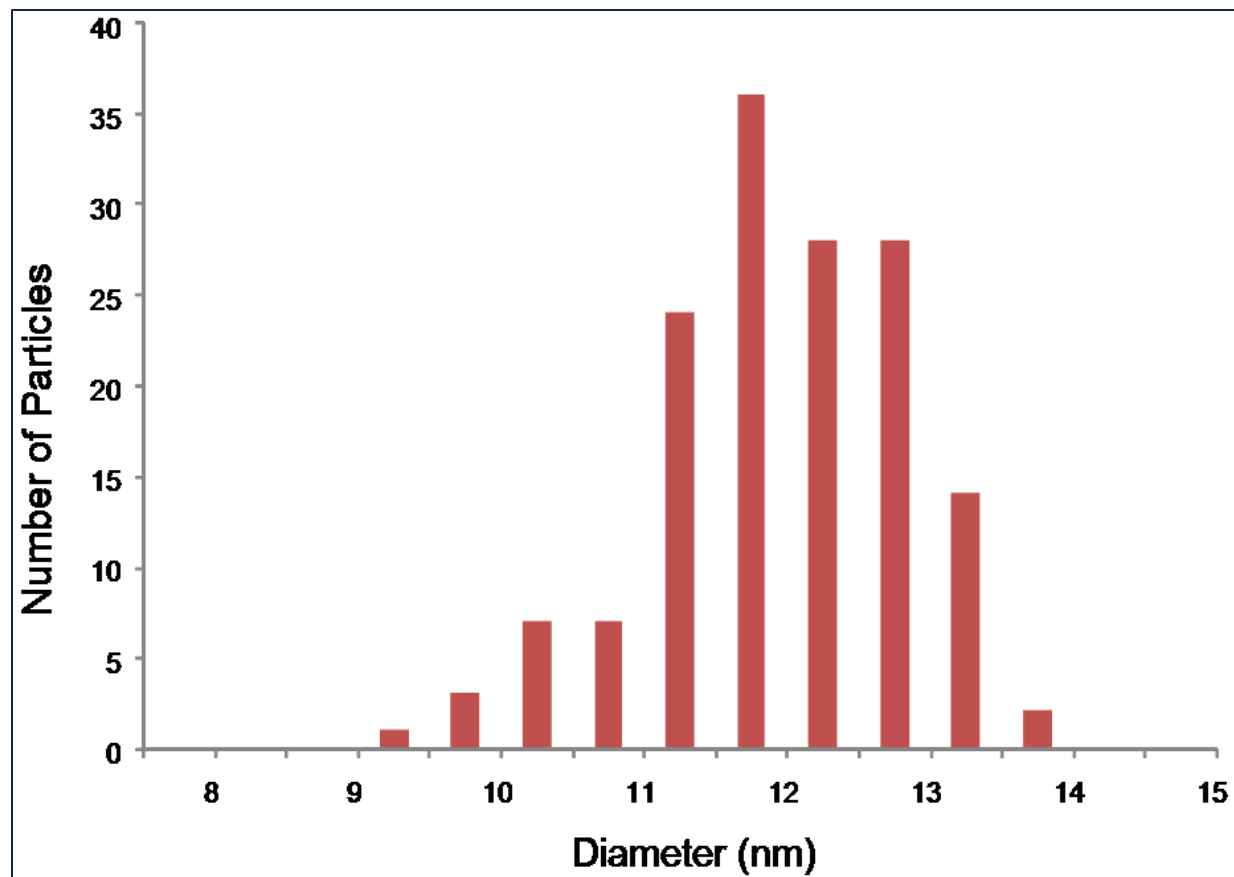
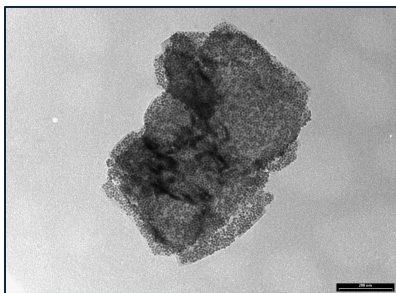


[§] Prepared as described in: N. Mezailles *et al.* *Nano Today* **2012**, 7, 21.

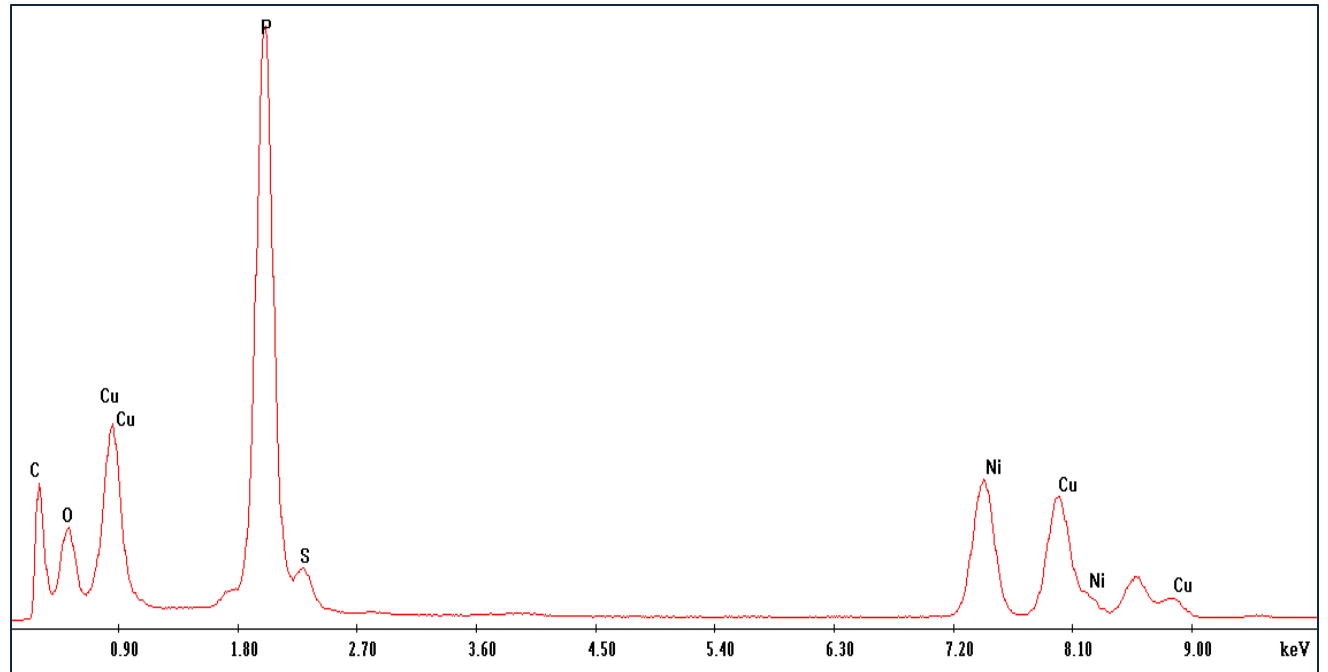
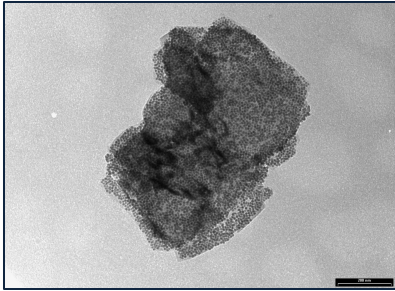
TEM image of NiNPs/2D-BP



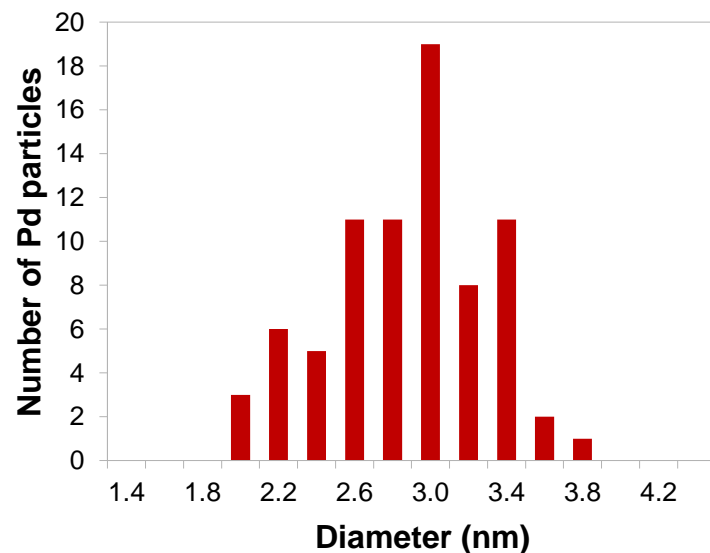
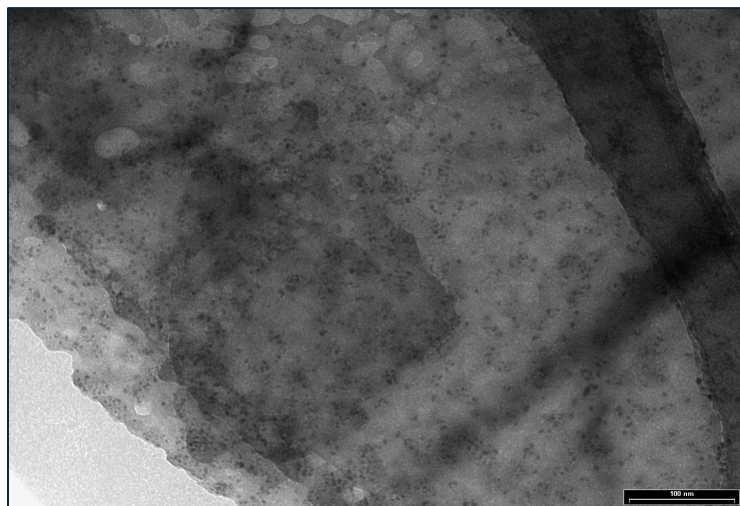
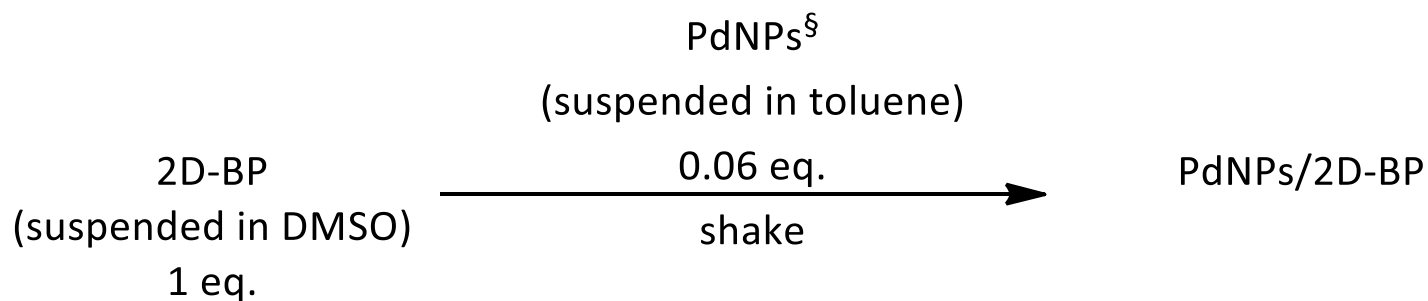
Size distribution of NiNPs anchored on 2D-BP



EDX analysis of NiNPs/2D-BP

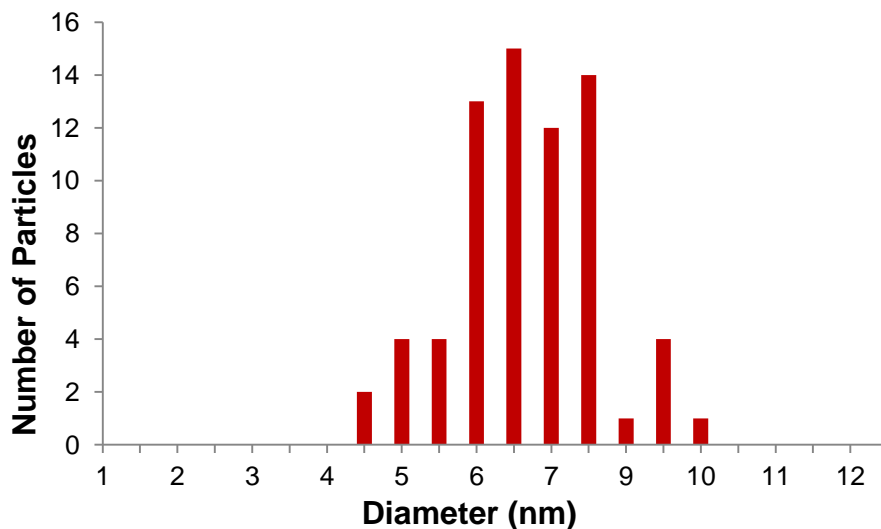
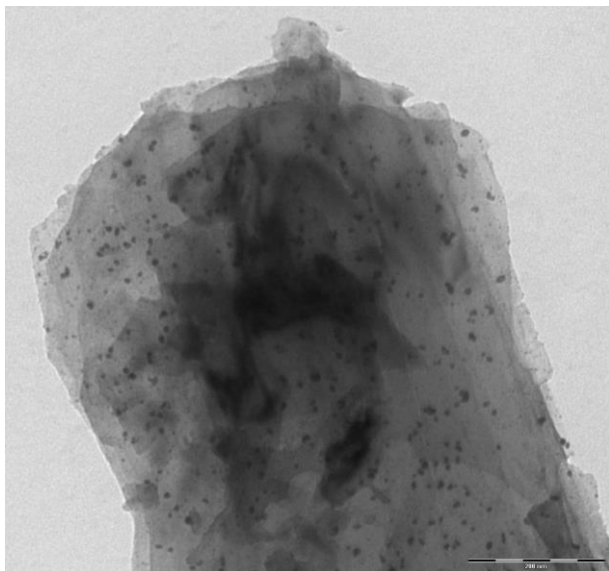
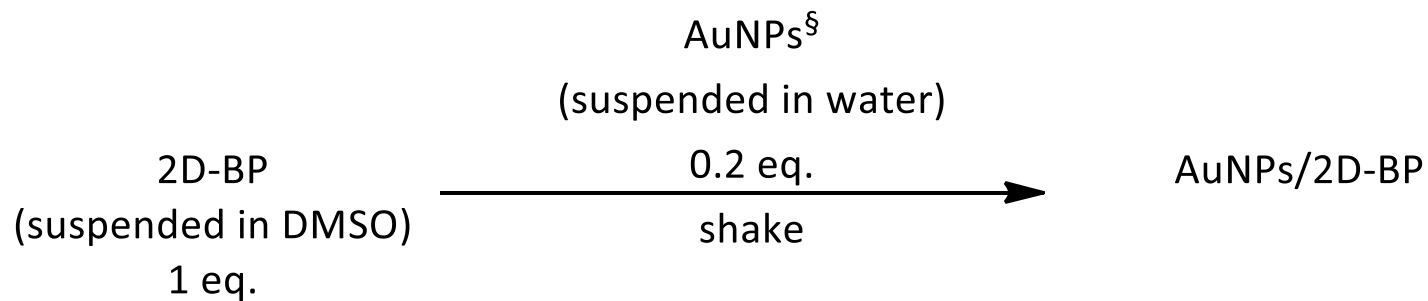


Preparation and TEM analysis of PdNPs/2D-BP



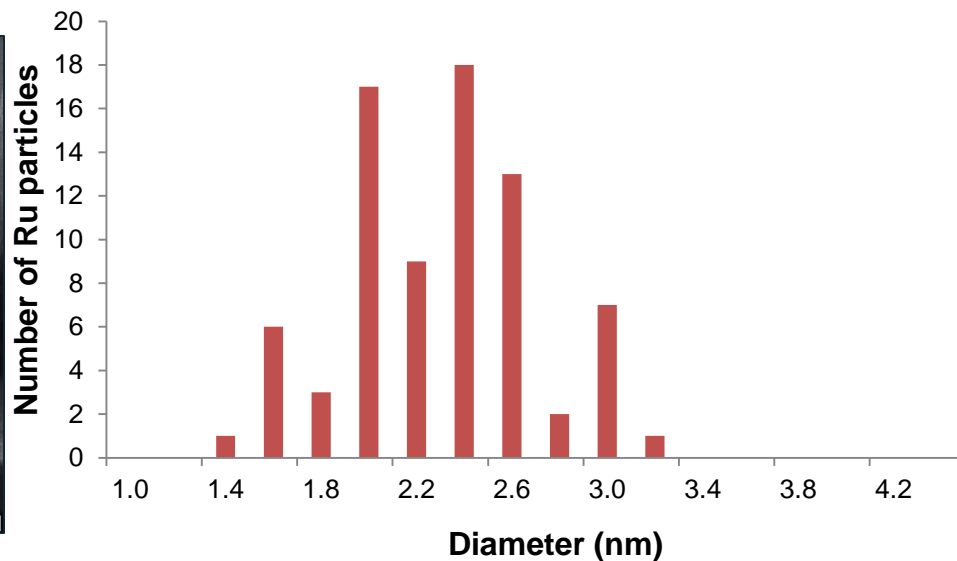
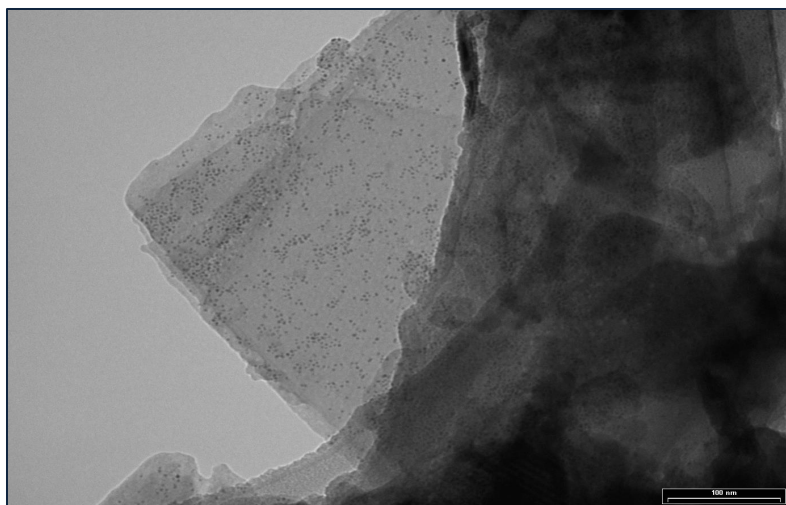
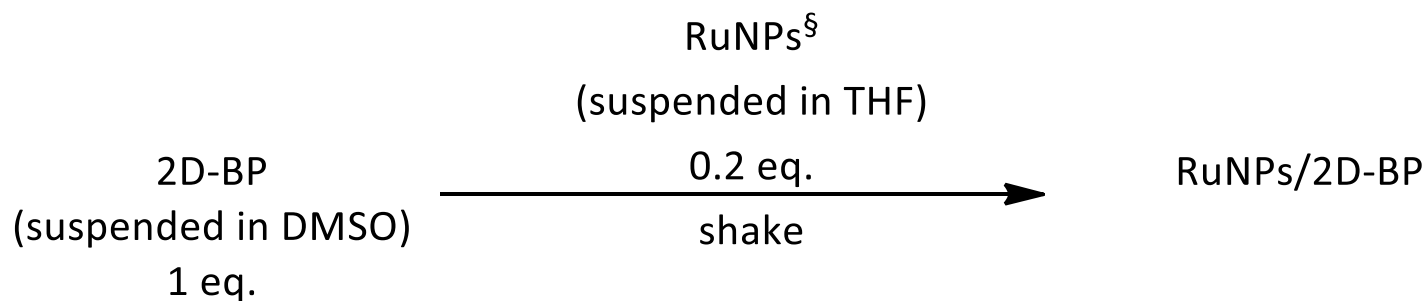
[§]Prepared as described in: Coronado *et al. J. Mater. Chem.* **2008**, *18*, 5682.

Preparation and TEM analysis of AuNPs/2D-BP



[§]Prepared as described in: Yang, J. Dong, Z. Yao, C. Shen, X. Shi, Y. Tian, S. Lin, X. Zhang, *Sci. Rep.* **2014**, 4, 4501.

Preparation and TEM analysis of RuNPs/2D-BP



[§]Prepared as described in: H. Can, Ö. Metin, *Appl. Cat. B-Environ.* **2012**, *125*, 304-310.

Outline

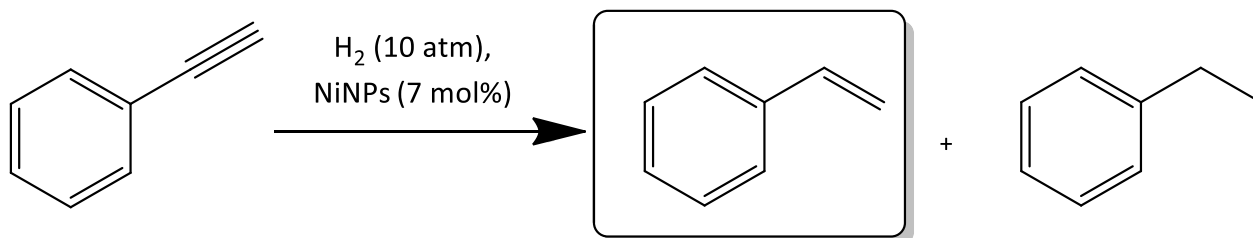
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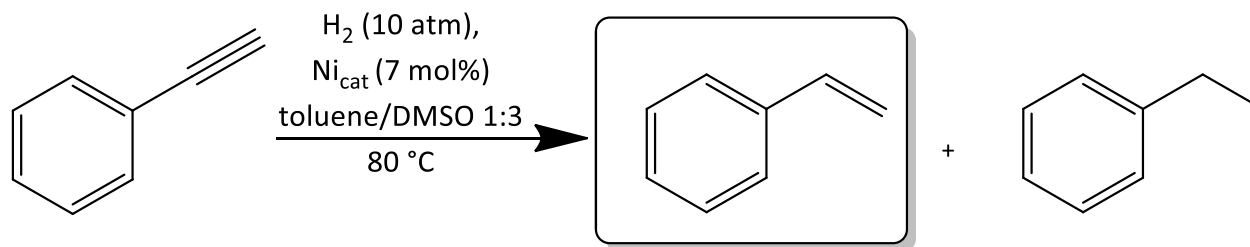
Phenylacetylene hydrogenation catalyzed by NiNPs



Solvent	Temperature (°C)	Time (h)	Conv. (%)*	Selectivity toward styrene(%)*
toluene	80	2	2.2	51.1
toluene/DMSO 1:3	80	1	77.9	94.3
toluene/DMSO 1:3	80	2	99	96.1

* Evaluated by GC

Phenylacetylene hydrogenation catalyzed by NiNPs(TOP)/2D-BP



Nicat = NiNPs

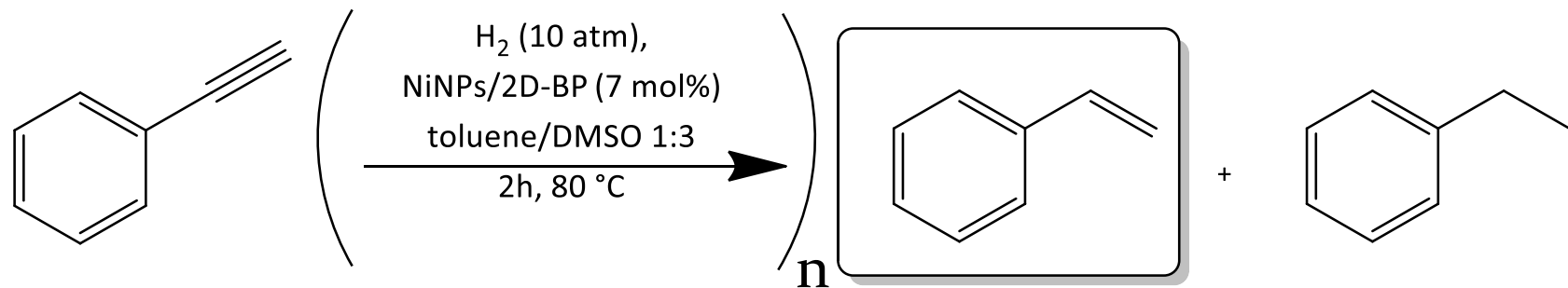
Time (h)	Conv. (%)*	Selectivity toward styrene(%)*
1	77.9	94.3
2	99	96.1

Nicat = NiNPs/2D-BP (P:Ni = 2)

Time (h)	Conv. (%)*	Selectivity toward styrene(%)*
1	45.4	94.3
2	96.2	94.9

* Evaluated by GC

Recycling the catalyst NiNPs/2D-BP



n	Conversion (%)*	Selectivity towards styrene(%)*
1	96.2	94.9
2	91.9	94.7
3	86.9	95.1
4	73.2	95.2
5	66.6	95.6

*Catalyst was isolated by centrifugation (6000 rpm x 2 with acetone) before every recycling test

* Evaluated by GC

Future work and perspectives

- ✓ Evaluate the effective heterogeneity of the studied catalytic system NiNPs/2D-BP (hot filtration tests, leaching tests...) and run a comparison of 2D-BP with other 2D-supports for metal nanoparticles (graphene, carbon nanotubes...).
- ✓ Explore the catalytic activity of the other nanocomposites, RuNPs/2D-BP, PdNPs/2D-BP and AuNPs/2D-BP in several reactions.
- ✓ Clarify the nature and strenght of the interactions between TM nanoparticles and 2D-BP, using appropriate instrumental techniques (Raman, XPS spectroscopy ...).
- ✓ Study the chemical and physical properties of the nanocomposites, in order to expand their uses in other fields (e. g. microelectronics, optoelectronics).

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Acknowledgements



CNR-ICCOM:

- Maria Caporali
- Maurizio Peruzzini
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- Giuseppe Nicotra



CNR-NANO:

- Stefan Heun
- Francesca Telesio
- Shaohua Xiang



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