

# Palladium/black phosphorus nano hybrid: what surface techniques tell us about the Pd-P interaction



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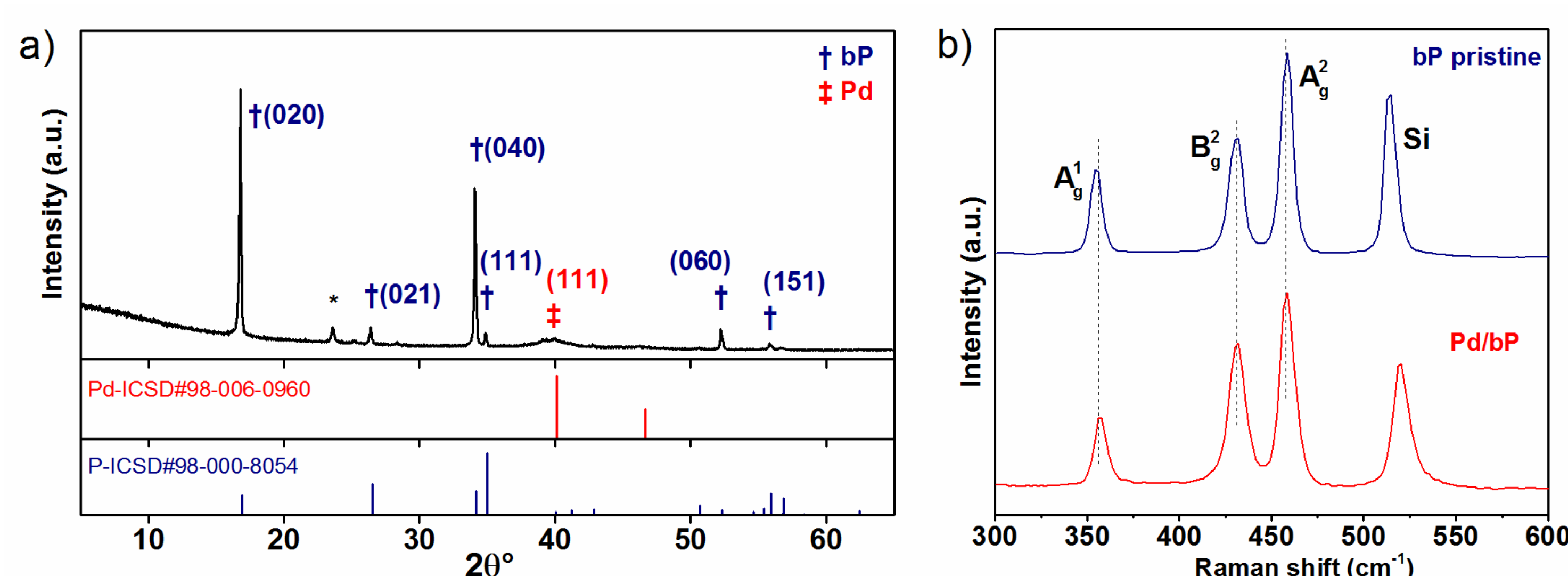
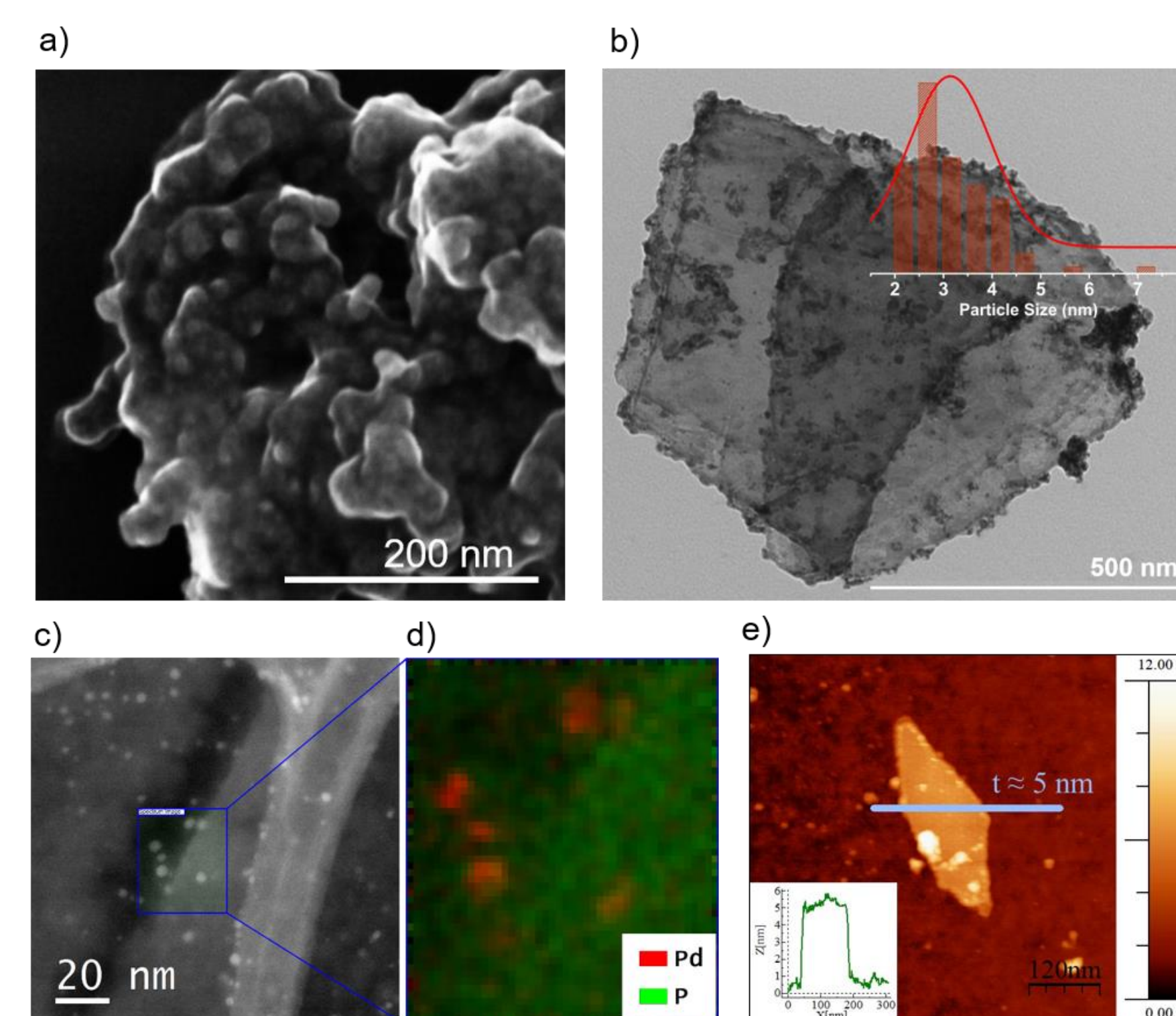
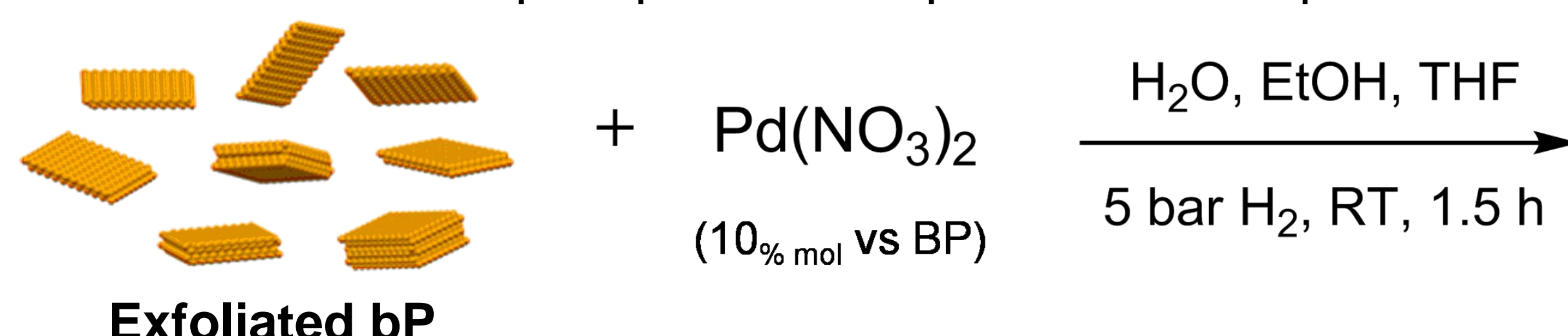
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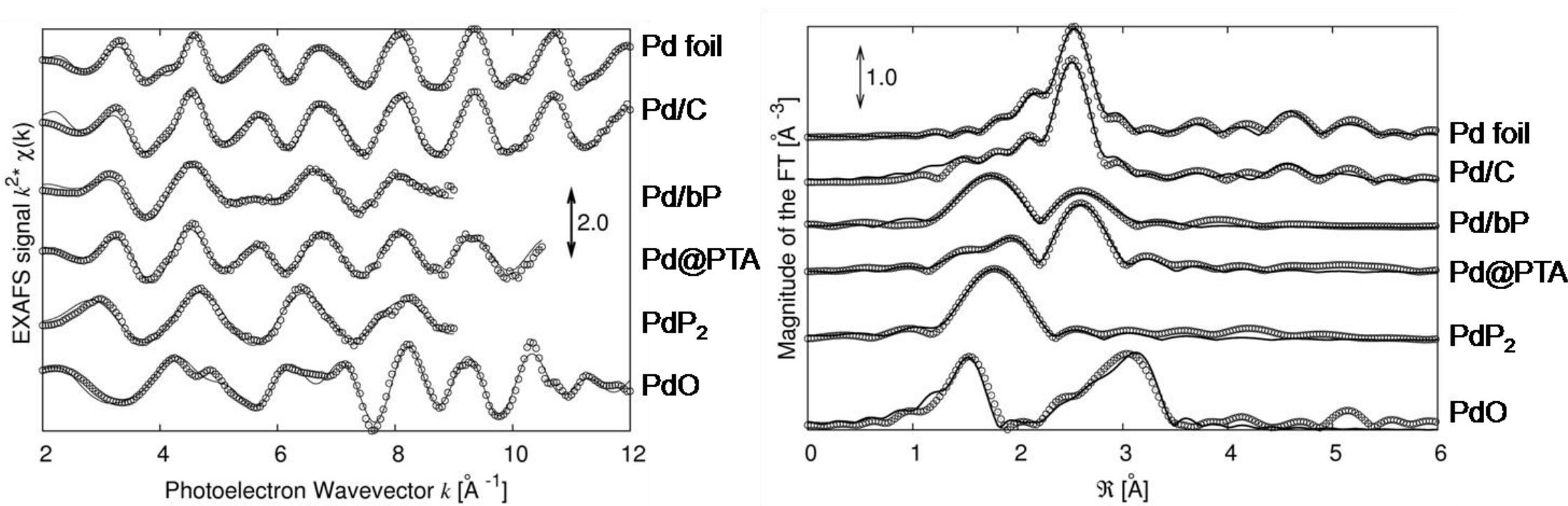
Black phosphorus has drawn major excitement to the 2D-materials community since its first appearance in 2014. Despite its structural resemblance to graphene, each atom in black phosphorus is  $sp^3$  hybridized, bearing a lone pair which may be exploited to achieve the functionalization of its surface. A major problem dealing with surfaces and solid state materials is the elucidation of interactions and structures at a nanometric scale, as common molecular characterization methods are not applicable with these materials. We show here how surface techniques can successfully tackle the problem and tell us a lot about the bonding properties between black phosphorus and palladium nanoparticles.



a) XRD spectrum of Pd/bP b) Raman spectrum of pristine bP (blue) and Pd/bP (red).

a) SEM image of Pd/bP. b) TEM image of Pd/bP c) High resolution HAADF STEM image of Pd/bP on a lacey carbon grid. (d) EDS elemental mapping of the selected area on Pd/bP. e) AFM image of a Pd/bP flake on Si/SiO<sub>2</sub>. The line corresponds to the cross-sectional profile shown as an inset.

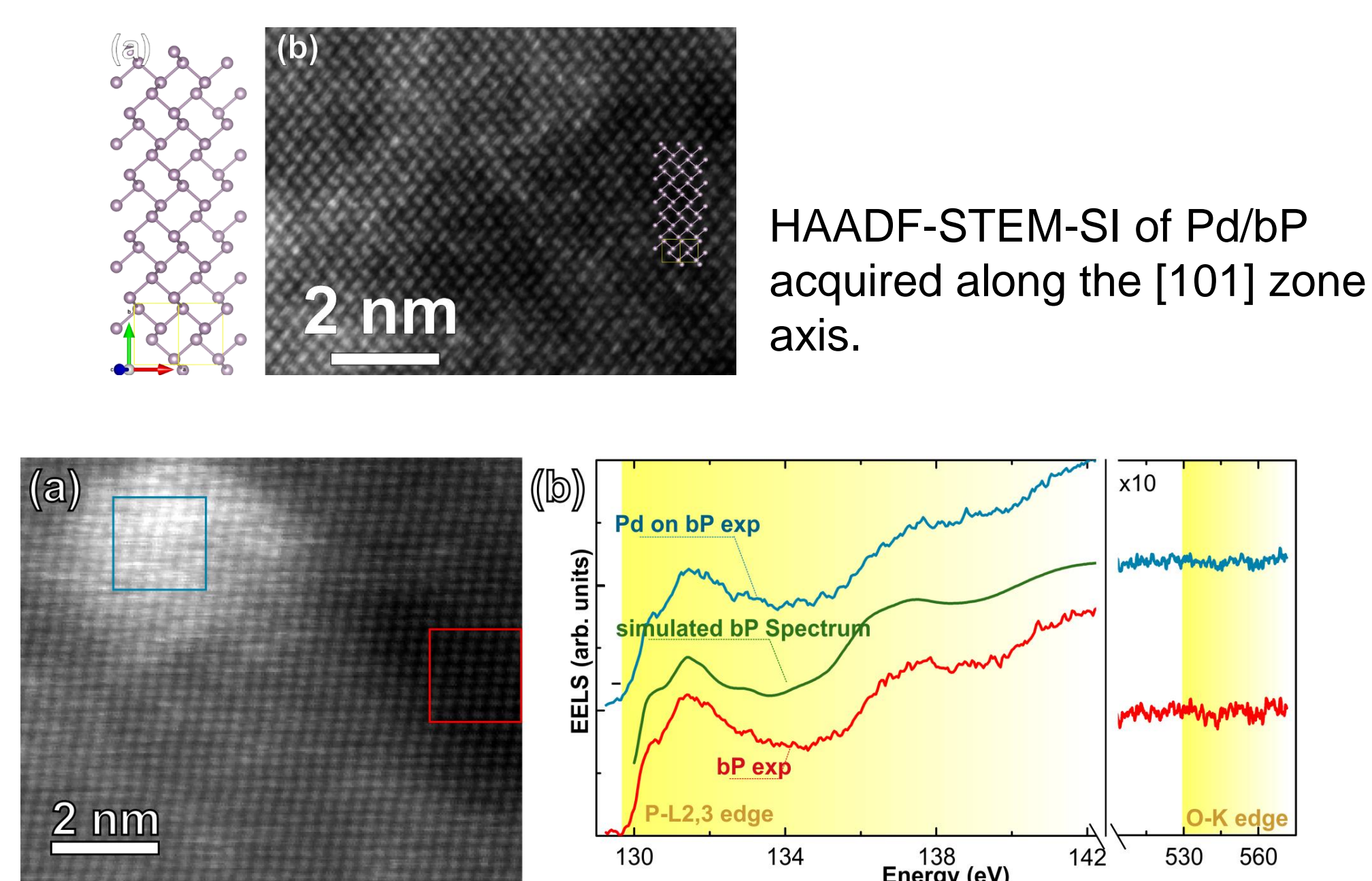
## EXAFS



a) EXAFS data at the Pd-K edge; b) the corresponding Fourier transforms.

EXAFS measurements carried out on Pd/bP and on reference materials evidenced the clear presence of a Pd-P bond at a distance of 2.26(3) Å.

## HRTEM-EELS

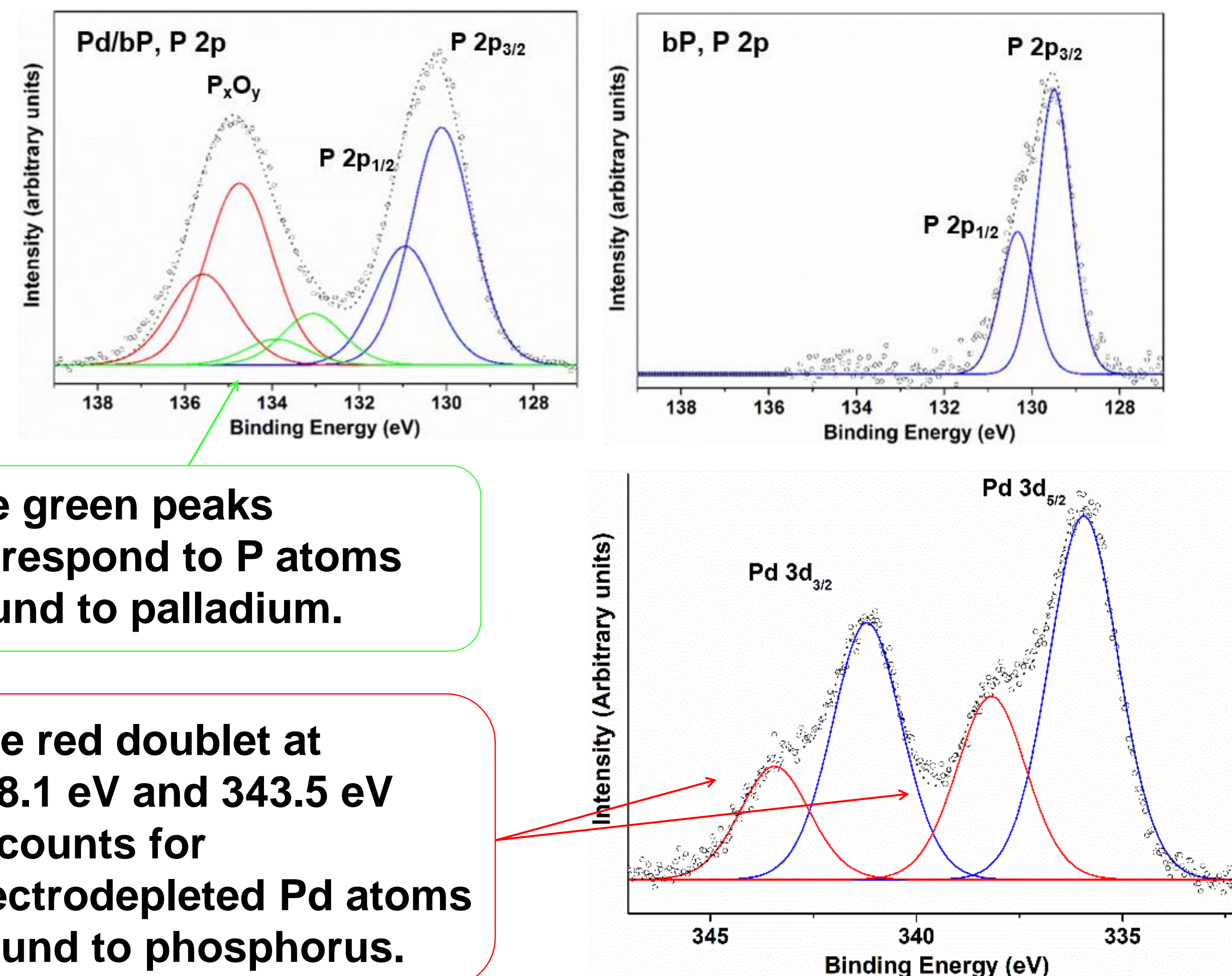


A difference in the  $P_{L_2,3}$ -edge around 138 eV in Pd/bP compared to pristine bP reveals a modification in the electronic structure of bP after functionalization with Pd nanoparticles.

### Acknowledgements

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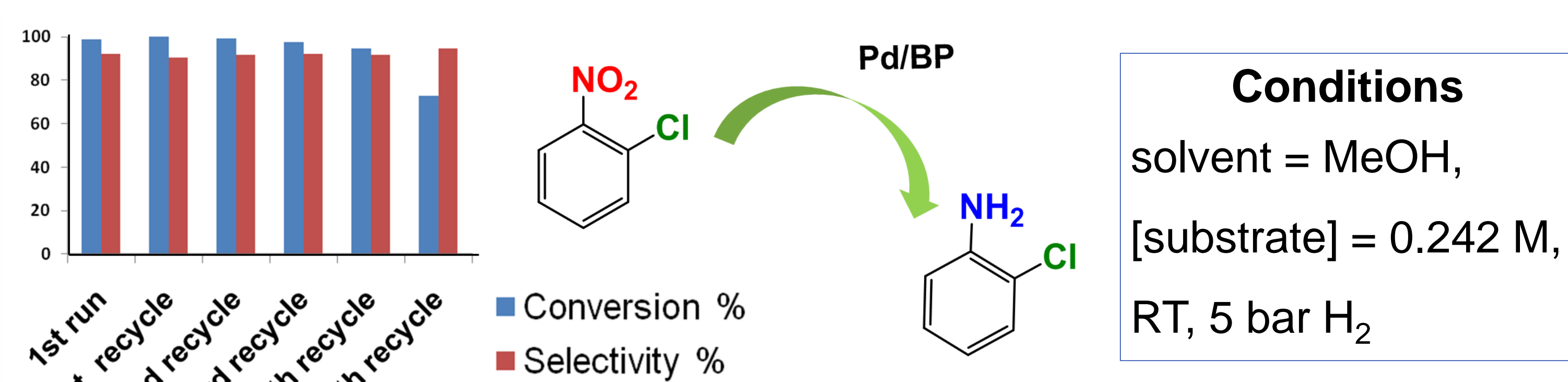
## XPS



The green peaks correspond to P atoms bound to palladium.

The red doublet at 338.1 eV and 343.5 eV accounts for electrodepleted Pd atoms bound to phosphorus.

## o-chloronitrobenzene hydrogenation



**Conditions**  
solvent = MeOH,  
[substrate] = 0.242 M,  
RT, 5 bar H<sub>2</sub>

Catalyst	Substrate	Time (min)	Sub/Cat	Conversion%	Selectivity%	TOF (h <sup>-1</sup> )
Pd/bP	1-chloro-3-nitrobenzene	30	162	99.1	97.7	313
	1-chloro-2-nitrobenzene	40	162	99.5	97.3	235
Pd/C	1-chloro-2-nitrobenzene	30	191	99.9	78.1	298